

# Structural Fusion of Condensed Benzenoid Systems

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On the basis of ring topologies, some theorems for the structural fusion of condensed benzenoids are presented. It has been proved that the oddness or evenness of the fused system is dictated by the degree of fusion.

## Introduction

Benzenoid hydrocarbons form an important and well investigated class of conjugated compounds. Benzenoid graphs are the networks obtained by arranging congruent regular hexagons in the plane so that two hexagons are either disjoint or possess a common edge<sup>1</sup>.

The concept of all-benzenoid hydrocarbons was introduced by Clar and Zander<sup>2</sup>, and elaboration on it eventually led to the formulation of Clar's well known aromatic sextet theory<sup>3</sup>.

Actually there exist many publications on the theory of benzenoid hydrocarbons<sup>4-11</sup>, and a good account of topological properties of benzenoid systems is presented by Gutman<sup>12</sup>.

Although in the last decade there have been innumerable articles on the structural theory of benzenoid hydrocarbons, some of which are cited above, there is still more to be investigated. In the present study, some light has been shed on the fusion of condensed benzenoid systems.

## Theory

### Structural Fusion of Benzenoid Systems

Theoretically, a large benzenoid system can be regarded as resulting from the fusion of two or more smaller benzenoid systems. The process of fusion in that sense requires the overlap of a certain number of edges and vertices of the corresponding molecular graphs of the systems being considered.

Let  $V_1$  and  $V_2$  be vertex sets of benzenoid graphs  $G_1$  and  $G_2$ , respectively, such that

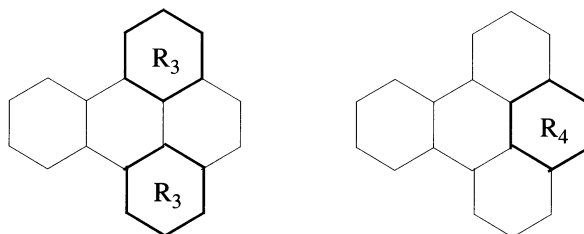
$$V_1 = \{v_i : v_i \in V_1\} \quad (1)$$

$$V_2 = \{v_j : v_j \in V_2\} \quad (2)$$

**Definition 1:** The degree of fusion (DF) of either of benzenoid graphs  $G_1$  and  $G_2$  (generating graphs) in the process of fusion is given by  $DF = V_1 \cap V_2$ , where  $V_1$  and  $V_2$  are the corresponding vertex sets.

Note that in general the fusion of two-ring systems associated with  $DF = 1$  yields a spiro compound, the possibility of which is excluded in the case of benzenoid compounds.

In the case of condensed benzenoid systems, an individually fused ring may have a DF of 2-6. In the present study, such rings are denoted as  $R_2$ ,  $R_3$ ,  $R_4$ , etc. Figure 1 depicts some  $R_3$ - and  $R_4$ - type rings present in benzopyrene.

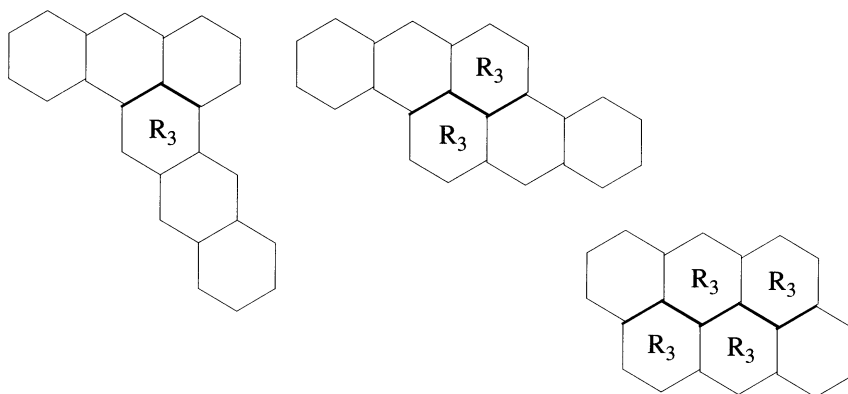


**Figure 1.**  $R_3$ - and  $R_4$ - type rings of benzopyrene

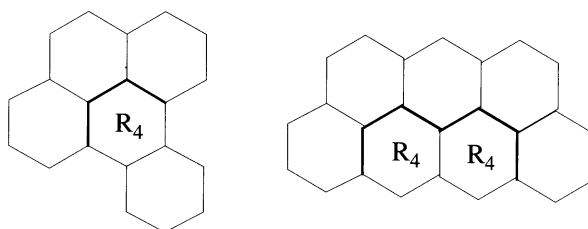
Let  $G_1$  and  $G_2$  be two benzenoid graphs and  $F_i$  be an  $i$ -type fusion operator that operates on  $G_1$  and  $G_2$  to produce a larger graph,  $G$ , which possesses  $m$  of  $R_i$ -type rings. The whole fusion process can then be represented as

$$F_i(G_1, G_2) = G(mR_i) \quad (3)$$

The topologies of the simple and the complex type  $R_3$  rings obtained from anthracene system are shown in Figure 2, whereas some  $R_4$ -type rings are depicted in Figure 3. Note that various fissures and bays<sup>8</sup> are involved in the formation of  $R_3$ - and  $R_4$ -type rings, respectively.



**Figure 2.** Topologies of  $R_3$ -type rings obtained from anthracene



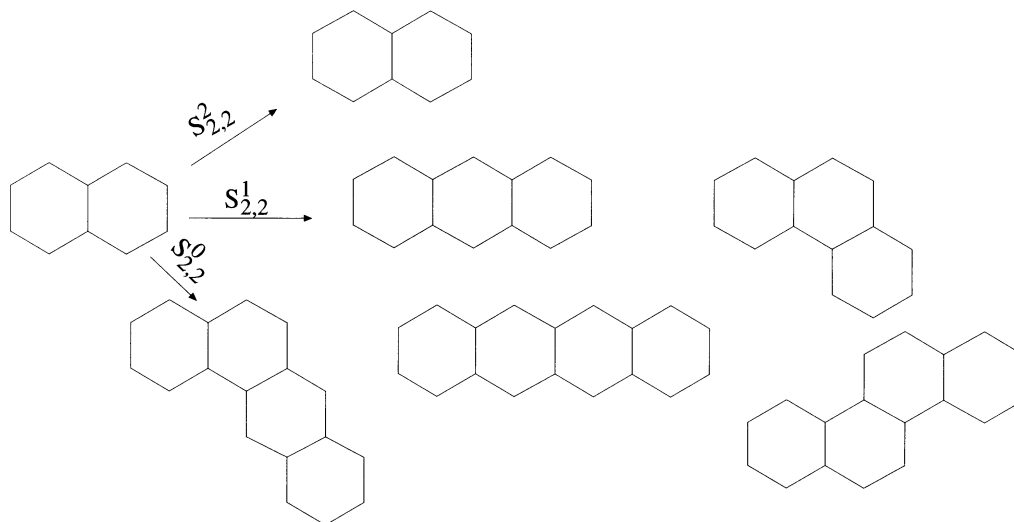
**Figure 3.** Topologies of  $R_4$ -type rings

## Superimposition of Benzenoid Systems

**Definition 2:** The process in which one or more rings of two benzenoid graphs overlap is called superimposition.

**Definition 3:** The number of benzenoid rings overlapped in the process of superimposition is called the degree of superimposition (DS).

The superimposition operator,  $S_{ij}^{DS}$  (where  $i$  and  $j$  stand for the number of benzenoid rings present in graphs  $G_i$  and  $G_j$ , respectively, and DS is the degree of superimposition) engenders larger benzenoid graphs starting with smaller members. Figure 4 depicts some superimposition patterns of naphthalene obtained by the overlap of another naphthalene system, and shows the ring topologies<sup>13</sup> of the resultant cata-condensed systems as well. Note that isomeric systems



**Figure 4.** Some superimposition patterns of naphthalene

are produced by the same superimposition operator. In general, the number of benzenoid rings ( $R$ ) in the final system is given as

$$R = i + j - DS \quad (4)$$

Hence, cata-condensed benzenoids produced by the same superimposition operator are evidently isomers of one another.

**Theorem 1.** Let  $G_1$  and  $G_2$  be any two add alternant benzenoid graphs. If the fusion of  $G_1$  and  $G_2$  to produce graph  $G$  is accompanied by the emergence of an even (odd) number of simple  $R_3$ -type rings, then  $G$  is an even (odd) alternant system.

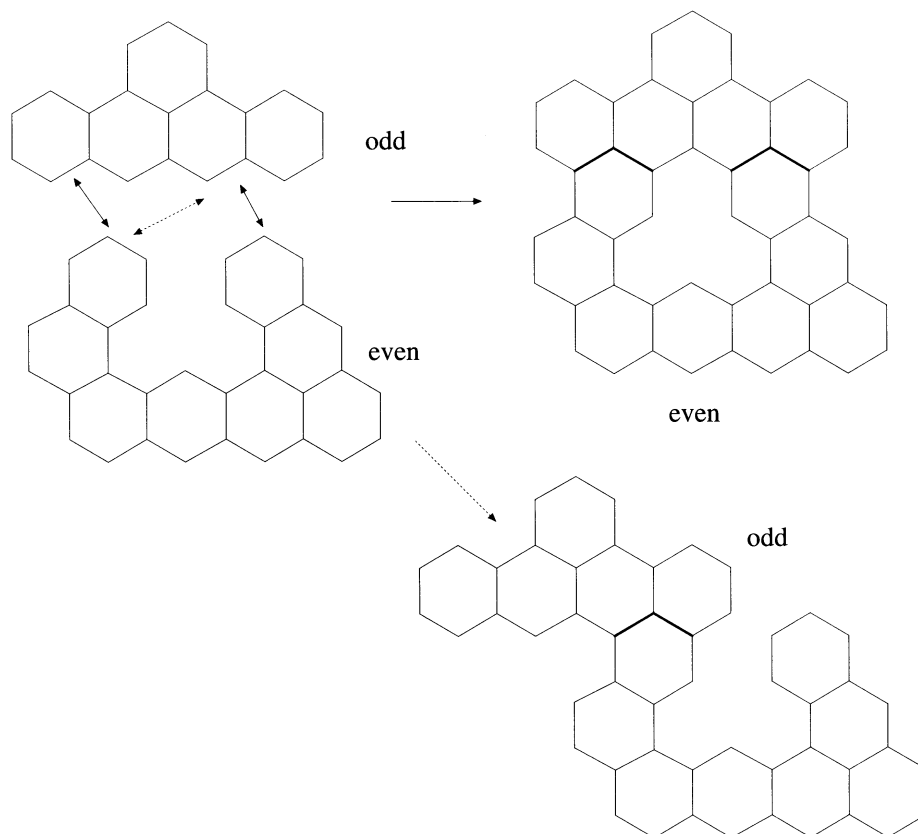
**Proof.** Let  $v_1, v_2$  and  $v$  be the number of vertices in graphs  $G_1, G_2$  and  $G$ , respectively. Since  $G_1$  and  $G_2$  correspond to odd alternant hydrocarbons, suppose  $v_1 = 2k + 1$  and  $v_2 = 2t + 1$  where  $k$  and  $t$  are any integers. Since the presence of every simple  $R_3$ -type ring means three vertices of  $G_1$  and  $G_2$  are mutually shared to produce graph  $G$ , which is  $F_3(G_1, G_2) = G(mR_3)$ , then

$$v = v_1 + v_2 - 3m \quad (5)$$

where  $m$  is the number of  $R_3$ -type rings. Inserting the above parametric forms of  $v_1$  and  $v_2$  into equation (5), one obtains

$$v = 2k + 2t + 2 - 3m \quad (6)$$

Note that the degree of fusion in this case is equal to  $3m$ . Since  $2(k + t + 1)$  is an even number, the oddness or evenness of  $v$  is dictated directly by the parity of  $m$ ; that is, if  $m$  is even (odd) then  $v$  is even (odd), which proves the theorem. Below, figure 5 illustrates the application of Theorem 1.



**Figure 5.** The fusion of two odd benzenoid systems

**Corollary 2.** Let  $G_1$  and  $G_2$  be any odd even benzenoid graphs, respectively. If an even (odd) number of simple  $R_3$ -type rings appears in the fusion process of  $G_1$  and  $G_2$  to produce graph  $G$ , then  $G$  is an odd (even) alternant system.

**Corollary 3.** Let  $G_1$  and  $G_2$  be any two even benzenoid graphs. If their fusion to produced graph  $G$  is accompanied by the emergence of an even (odd) number of simple  $R_3$ -type rings, then  $G$  is an even (odd) alternant system.

**Corollary 4.** If the fusion of any two add (even) benzenoid graphs  $G_1$  and  $G_2$  is accompanied by the emergence of any number of simple  $R_4$ -rings, then the resultant graph,  $G$ , is an odd (even) alternant system.

**Corollary 5.** Let  $G_1$  and  $G_2$  be any odd and even benzenoid graphs, respectively. If the fusion of  $G_1$  and  $G_2$  to produce graph  $G$  results in any number of simple  $R_4$ -type rings, then  $G$  is an odd alternant system.

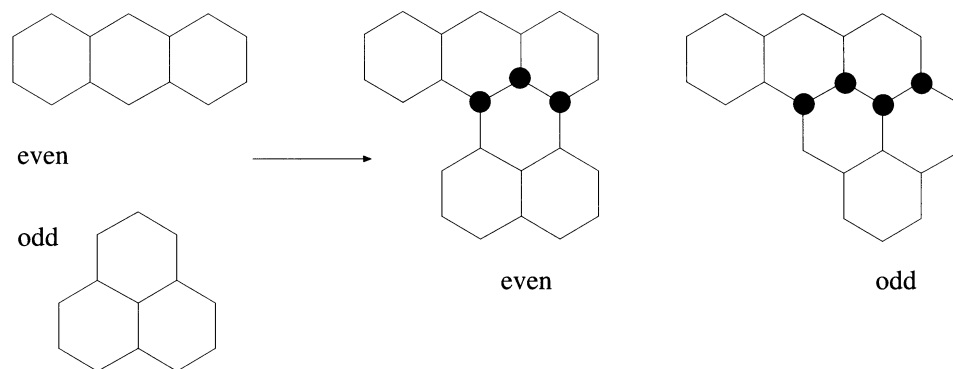
**Definition 4.** In the process of the fusion of graphs  $G_1$  and  $G_2$ , the mutually shared vertices are called the fusion points. Note that the total number of fusion points is equal to  $DF$ .

**Corollary 6.** Let  $G_1$  and  $G_2$  be any odd or even benzenoid graphs. If the number of fusion points between them is odd (even), then the combined graph  $G$  obtained by the fusion of  $G_1$  and  $G_2$  is an odd (even) alternant system.

**Corollary 7.** Let  $G_1$  and  $G_2$  be any odd and even benzenoid graphs, respectively. If the number of fusion points between them is odd (even), then the combined graph  $G$  obtained by the fusion of  $G_1$  and  $G_2$  is an even (odd) alternant system (Figure 6).

## Conclusion

The results of the preceding section offer a straightforward method for the structural recognition of benzenoid hydrocarbons, and it is perfectly suited to the generations, searches and classification of benzenoid systems. The present theorems shed some light on how variations in the structural fusion of components dictate the oddness or evenness of the benzenoid hydrocarbons; thus, certain properties of alternant hydrocarbons, such as the existence of nonbonding molecular orbitals, can be considered componentwise. Obviously, investigations into how fused systems affect certain other theoretical and physico-chemical properties of the combined systems would make very interesting future studies.



**Figure 6.** The fusion of an odd and an even benzenoid systems

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