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Organic geochemistry and element distribution in coals formed in Eocene lagoon facies from the Eastern Black Sea Region, NE Turkey

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Abstract: The amount and type, the inorganic element content, and the maturity of organic materials of Eocene coals, shaly coals, and coaly shales exposed in the Gümüşhane and Bayburt districts of the Eastern Black Sea Region of Turkey were investigated. The depositional environments and hydrocarbon potentials were also interpreted. The total organic carbon concentrations in the studied samples ranged from 0.50% to 63.08%. The samples from Özyurt, Kayadibi, and Tarhanas contained types II and III kerogen, and those from Sökmen and Manas contained type III kerogen. The samples contained Co, Cs, Ga, Hf, Th, U, Y, Mo, Be, Cd, Sb, and La, with average values similar to those of standard brown coals. The samples showed average contents of Co, Ga, Nb, Rb, V, Y, Cu, Pb, Zn, As, Be, and Se, similar to those of other Turkish coals. The sediment source of Eocene samples in the five areas was characterized by rocks with intermediate or mafic geochemical characteristics. The terrigenous/aquatic ratio of coal and shaly coal samples of the areas in question is >1. The sterane distribution was $C_{29} > C_{27} > C_{27}$ and $C_{29} > C_{27} > C_{28}$ for the Özyurt and Tarhanas areas, respectively. The average T_{max} values for samples are between 424 °C and 460 °C. For samples Oz-1 and Ta-2, 22S/(22S + 22R) homohopane (C_{32}) ratios are 0.48 and 0.61, respectively; 20S/(20S + 20R) sterane (C_{29}) ratios are 0.18 and 0.53, respectively; and Ts/(Ts + Tm) ratios are 0.015 and 0.64, respectively. The Pr/Ph ratios of the samples are >3. The studied samples have low sterane/hopane and high ($C_{19}+C_{20}$)/ C_{23}) ratios without anoxic biomarkers (17 α (H)-28,30-bisnorhopane). Based on these data, the coals, shaly coals, and coaly shales were probably deposited under an oxic-suboxic mixture of marine and terrestrial environment conditions; these materials contain terrestrial organic matter and cannot generate hydrocarbon.

Key words: Northeastern Turkey, Eocene coal, geochemistry, total organic carbon, rare earth elements and yttrium, gas chromatographymass spectrometry, organic matter, paleoenvironment

1. Introduction

The total coal reserves of Turkey are estimated to be in the order of 13.4 billion tons of lignite and 0.4 billion tons of bituminous coal. Most of the lignite deposits are located in Tertiary basins, while Eocene lignite deposits are very limited.

Eocene aged clastic rocks of the eastern Pontides (NE Turkey) exhibit two different source characteristics. Volcaniclastic deposits are dominant in the northern section of Gümüşhane and siliciclastic deposits are dominant in the southern section (Kelkit, Köse). In Bayburt, deposition starting with basal conglomerate has volcaniclastic characteristics in the north of Varicna village and has siliciclastic characteristics in other sections (Saydam Eker, 2012, 2015). Eocene aged sedimentary rocks in the Gümüşhane region are composed of siliciclastic deposits. The rocks come over Cretaceous aged sedimentary rocks of this region with discordance. Eocene aged siliciclastic rocks lie with discordance over Late Cretaceous aged limestones in the Bayburt region (Saydam Eker, 2012). The investigated coal occurrences are found within Eocene siliciclastic deposits in the Gümüşhane and Bayburt regions.

Prior to this publication, no record existed on the working details of the organic and inorganic geochemistry of the Eocene aged coals in the Eastern Black Sea Region (Gümüşhane and Bayburt fields). The aim of the present study was to determine organic matter contents and distributions of major, trace, and rare earth elements, and to interpret organic matter types and maturities, depositional environments, and hydrocarbon potentials of selected Eocene coals, shaly coals, and coaly shales in NE Turkey.

2. Geological background

The eastern Pontides belt in the Black Sea Region of Turkey is part of the Alpine metallogenic belt that has been subdivided into northern, southern, and axial zones,

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distinguished from north to south by different lithological units, facies, and tectonic characteristics (Bektaş et al., 1995; Eyuboglu et al., 2006). The northern zone contains Mesozoic-Cenozoic volcanic sequences associated with massive sulfide deposits, calderas, and granitic intrusions (Arslan et al., 1997; Şen et al., 1998; Kaygusuz et al., 2008, 2011; Sipahi, 2011; Temizel et al., 2012, Sipahi et al., 2014). The southern zone includes Mesozoic and Eocene sedimentary rocks, pre-Liassic ultramafic-mafic rocks, and metamorphic-granitic rocks (Figure 1A). Upper mantle peridotites and middle to upper Cretaceous olistostromal mélange occupy much of the axial zone (Eyuboglu et al., 2010). The basement rocks of the eastern Pontides are composed of metamorphic rock and granitoids (Yılmaz, 1972; Çoğulu, 1975; Okay and Şahintürk, 1997; Topuz et al., 2004, 2010; Dokuz, 2011). Liassic volcanics and volcaniclastic and clastic deposits lie unconformably on the basement rocks (Yılmaz, 1997; Şen, 2007). This unit is overlain by pelagic and neritic carbonates of Malm-Lower Cretaceous age. The Upper Cretaceous, largely represented by volcanics in the north, developed into turbiditic facies in the south (Saydam Eker and Korkmaz, 2011). Eocene aged rocks in the Gümüşhane region are composed of volcanics, volcanosediments, and coal interbedded with siliciclastic rocks in various places (Figure 1B). Eocene rocks in the Bayburt region are composed of volcaniclastics, basal conglomerate, and coal interbedded with turbiditic members (Saydam Eker, 2012) (Figure 1B). This sequence is widespread in the region and discordantly overlies the older rocks. Miocene and Pliocene deposits occurred in restricted areas and are characterized by clastic material (Saydam and Korkmaz, 2008; Figure 1A).

3. Samples and methods

In this study, coal, shaly coal, and coaly shale samples were collected from five different areas [Tarhanas, Kayadibi, Özyurt, Sökmen (Gümüşhane), and Manas (Bayburt); total thicknesses of coal bearing claystones are ~10 m, 15 m, 50 m, 8 m, and 15 m, respectively (Figure 1B)] of the Eastern Black Sea Region. Rock-Eval/total organic carbon (TOC) analysis was applied to 22 chosen bulk samples (Özyurt: 6 samples, Kayadibi: 5 samples, Tarhanas: 4 samples, Manas: 4 samples, and Sökmen: 3 samples; Figure 1B). Whole rock major element, trace element, and rare earth element (REE) analyses were separately applied to 20 coal, shaly coal, and coaly shale samples (Özyurt: 6 samples, Kayadibi: 4 samples, Tarhanas: 4 samples, Manas: 4 samples, and Sökmen: 2 samples). Gas chromatography (GC) was used for four samples (one sample each from Özyurt, Tarhanas, Kayadibi, and Manas). Gas chromatography-mass spectrometry (GC-MS) analyses were also performed on one sample each from Özyurt and Tarhanas (labeled as Oz-1 and Ta-2, respectively).

3.1. Organic geochemistry analysis

Rock-Eval pyrolysis/TOC analyses of all the samples were done using a Rock-Eval 6 instrument equipped with a TOC module. The samples were heated from 300 °C (hold time: 3 min) to 650 °C at 25 °C/min. The crushed coal was heated from 400 °C (hold time: 3 min) to 850 °C (hold time: 5 min) at 25 °C/min for oxidation. Extracts were obtained from two coal samples (Oz-1, Ir-2), a shaly coal sample (Ta-2), and a coaly shale sample (Ma-2) by 40 h of Soxhlet extraction of the powdered rock with dichloromethane (CH₂Cl₂). The whole extract was analyzed using an Agilent 6850 gas chromatograph equipped with a flame photometric detector and flame ionization detector. A fused capillary column (100 m, 0.25 mm i.d.) coated with cross-linked dimethylpolysiloxane (J&W, 0.50 µm film thickness) was used for separation and helium was used as the carrier gas. The oven temperature was programmed from 40 °C (hold time: 8 min) to 270 °C (hold time: 60 min) at 4 °C/min. The extract samples were separated into saturated hydrocarbon, aromatic hydrocarbon, and NSOcompound fractions by liquid chromatography. N-hexane, toluene, and methanol were used for eluting the fractions, respectively. GC-MS analyses were run on the two samples (Oz-1, Ta-2) having the highest levels of extract. The GC-MS analyses were conducted on saturated fractions of coal extracts. An Agilent 5975C quadrupole mass spectrometer was coupled to a 7890A gas chromatograph and 7683B automatic liquid sampler. The gas chromatograph was equipped with an HP-1MS fused silica capillary column of 60 m in length, 0.25 mm i.d., and 0.25 µm film thickness. Helium was used as the carrier gas. The oven temperature was programmed from 50 °C (hold time: 10 min) to 200 °C (hold time: 15 min) at 10 °C/min, to 250 °C (hold time: 24 min) at 5 °C/min, and then to 280 °C (hold time: 24 min) at 2 °C/min. Finally, the oven temperature was increased to 290 °C (hold time: 40 min) at 1 °C/min. The mass spectrometer was operated in the EI mode at ionization energy of 70 eV and source temperature of 300 °C. The biomarker contents were determined using single ion recording at m/z 191 (terpane) and m/z 217 (sterane). Compounds were identified by retention time and elution order matching. The analyses were carried out at the Oil and Organic Geochemistry Laboratory of the Turkish Petroleum Corporation (TPAO, Ankara).

3.2. Inorganic geochemistry analysis

Twenty samples were selected for whole rock major element, trace element, and REE analyses. Major and trace elements were determined by inductively coupled plasma (ICP)-emission spectrometry and ICP-mass spectrometry (MS) at ACME Analytical Laboratories Ltd., Vancouver, Canada, using standard techniques. Major and trace elements were analyzed by ICP using 0.2 g of rock powder fused with 1.5 g of LiBO, dissolved in 100 mL of 5% HNO₃.



Figure 1A. Simplified geological map of the Eastern Black Sea Region (after Güven et al., 1993) and location map of the study area. 1- Paleozoic metamorphic basement, 2- Paleozoic granites, 3- Jurassic–Lower Cretaceous sequences, 4- Upper Cretaceous volcanics, 5- Upper Cretaceous sedimentary rocks, 6- Paleocene volcanosedimentary sequences, 7- Paleocene granites, 8- Eocene volcanic and volcanoclastic rocks, 9- Eocene sedimentary rocks, 10- thrust fault, 11- study area (1: Özyurt, 2: Sökmen, 3: Tarhanas, 4: Kayadibi, 5: Manas fields).

Ignition loss was determined on dried samples heated to a temperature of 1050 °C for 15 min. REE analysis was conducted by ICP-MS at ACME.

4. Results and discussion

4.1. Rock-Eval pyrolysis and TOC

Table 1 lists our coal, shaly coal, coaly shale data from the Özyurt (6), Tarhanas (4), Sökmen (3), and Kayadibi (5) areas of the Gümüşhane region, as well as the Manas (4) area of the Bayburt region, including TOC and Rock-Eval pyrolysis analyses. The TOC concentrations in the study area ranged from 0.50% to 63.08%. In this paper, samples characterized by TOC concentrations of >50% are considered as coal, samples with concentrations ranging from 35% to 50% are considered as shaly coal, and samples with values of TOC <35% are considered as coaly shale.

The average highest PY value was calculated in Kayadibi coals and coaly shale (65.8 mg HC/g rock), the

average highest HI value was calculated in Tarhanas shaly coal and coaly shales (112 mg HC/g TOC), and the average highest T_{max} value was calculated in Manas coaly shales (460 °C). The average lowest PY, HI, and T_{max} values were calculated in Sökmen coaly shales (0.12 mg HC/g rock, 24 mg HC/g rock, and 424 °C, respectively) (Table 1).

4.2. Major, trace, and REY elements

Table 2 lists the percentages of major element oxides; concentrations of trace elements and REEs in the samples from Özyurt, Tarhanas, Sökmen, Kayadibi (Gümüşhane), and Manas (Bayburt); average values of major and trace elements of Eocene aged Sorgun coals (Karayigit, et al., 2000a); average values of major and trace elements of Turkish coals (Palmer et al., 2004); global average values of major elements of coal (Valkovic, 1983); and average values of trace elements and REEs of standard brown coal (Ketris and Yudovich, 2009). In the investigated samples from the five studied areas, Mg, K (except coals from



Figure 1B. Stratigraphy of the Eocene coals, coal beds, and positions of Eocene coals, shaly coals, and coaly shales in the Özyurt, Sökmen, Tarhanas, Kayadibi, and Manas fields.

Sökmen), Na, and Ti contents are generally close to one other, whereas Al, Fe, and Ca contents show remarkable differences. Sökmen area coaly shales show the highest average Al content (5.22%), whereas Özyurt coals and shaly coals exhibit the lowest value for Al content (0.75%). Coaly shales from Sökmen exhibit the highest average Fe content (4.56%), whereas the coals and shaly coals of Kayadibi show the lowest Fe value (0.97%). The average highest and lowest Ca contents are 14.9% and 0.08% for samples obtained from Tarhanas and Kayadibi, respectively.

The major element contents of the studied samples were compared with the average major element contents of Eocene aged Sorgun coals (Yozgat, Turkey) (Karayigit et al., 2000a). The Fe, Mg, and Ca contents of Özyurt; the Al, Mg, K, and Ti contents of Kayadibi; the Fe, Mg, Ca, K, and Ti contents of Tarhanas; and the Fe, Ca, K, and Ti contents of Manas are enriched. The remaining elements in the general area are depleted (except the Al contents of Manas samples). When the major element components of the investigated samples were compared with the average major element components of Turkish coals (Palmer et al., 2004), the Fe and Mg contents of Özyurt; the Al, Mg, K, and Ti contents of Tarhanas and Kayadibi; and the Al, Fe, Mg, K, and Ti contents of Manas show similarities.

The studied samples are depleted in Na (except Sökmen samples). All Sökmen samples contain low amounts of organic matter and only comprise coaly shale; therefore, the major element contents of the samples are enriched in Sorgun and Turkish coals. The major element contents of samples from the investigated areas were compared with the average major element contents of world coal (Valkovic, 1983). The Al, Na, and Ti contents of Özyurt; the Na content of Tarhanas; the Ca content of Sökmen; the Fe and Ti contents of Kayadibi; and the Na and Ti contents of Manas samples exhibit similarities to the average content of the major elements of the world's coals. The rest of the elements present are enriched (except the Na content of Kayadibi coals and shaly coal) (Table 2; Figure 2).

Generally, the average contents of W, Be, Cd, Hg, Sb, Se, and Tb in the examined samples show similarities. However, Be in Sökmen, Sb in Manas, and Se in Özyurt samples differed from those of the other samples. The lowest average contents of Co, Cs, Ga, Rb, V, and Zn values were calculated for Özyurt coals and shaly coals. The highest average Co, Cs, Ga, Rb, V, and Zn values were calculated in Sökmen coaly shales. Samples from Özyurt show the lowest average concentrations of Hf, Nb, Th, Zr, Cu, and Pb. Samples from Manas exhibit the highest

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Table 1. TOC and Rock-Eval results.

Sample	TOC	S ₁ mg	S ₂ mg	S ₃ mg	PY (S_1+S_2) mg	PI (S ₁ /	T °C	HI (S ₂ /TOC)	OI (S ₃ /TOC)	I ith alo are
ID	%Wt	HC/g rock	HC/g rock	O_2/g rock	CO_2/g rock	$S_1 + S_2$)	I _{max} C	mgHČ/g TOC	mgCO ₂ /g TOC	Lithology
Gümüşha	ne regio	n					-			
Özyurt										
Oz-1	53.87	1.32	91.6	3.09	92.9	0.01	434	156	5	Coal
Oz-2	38.5	0.13	19.35	30.14	19.5	0.01	444	50	78	Shaly coal
Oz-3	55.03	0.51	45.57	20.74	46.1	0.01	440	83	38	Coal
Oz-4	35.46	0.25	16.57	26.22	16.82	0.01	445	47	74	Shaly coal
Oz-5	36.74	0.81	23.04	3.15	23.85	0.03	407	63	9	Shaly coal
Oz-7	43.74	0.16	21.55	34.08	21.71	0.01	442	49	78	Shaly coal
Average	43.89	0.53	36.28	19.57	36.8	0.013	435	74.7	47	
Tarhanas										
Ta-1	10.04	0.03	4.97	6.72	5	0.01	455	50	67	Coaly shale
Ta-2	35.2	1.32	38.27	14.31	39.6	0.03	451	111	41	Coaly shale
Ta-3	3.43	0.21	8.54	1.29	8.75	0.02	444	249	38	Coaly shale
Ta-4	4.52	0.02	1.66	0.18	1.68	0.01	472	37	4	Coaly shale
Average	13	0.4	13.4	5.6	13.8	0.02	456	112	38	
Sökmen										
S-1	0.50	0	0.14	0.26	0.14	0.02	439	35	65	Coaly shale
S-2	0.53	0	0.07	0.18	0.07	0.05	348	16	42	Coaly shale
S-3	0.60	0.01	0.13	0.08	0.14	0.09	485	22	13	Coaly shale
Average	0.54	0.004	0.118	0.178	0.12	0.05	424	24	40	
Kayadibi										
Ir-1	58.39	0.36	51.88	11.22	52.3	0.01	435	89	19	Coal
Ir-2	63.08	0.68	65.01	10.54	65.7	0.01	432	103	17	Coal
Ir-3	29.24	29.24	16.6	11.5	45.8	0.01	426	57	39	Shaly coal
Ir-4	62.24	62.24	52.26	17.15	114.5	0.01	429	84	28	Coal
Ir-5	53.83	0.35	50.37	0.71	50.7	0.01	468	94	1	Coal
Average	53.4	18.6	47.2	10.2	65.8	0.01	438	85	20	
Bayburt r	region									
Manas						~				
Ma-1	3.38	0.01	2.11	2.46	2.12	0	440	62	73	Coaly shale
Ma-2	7.85	0.02	4.4	5.7	4.42	0	440	56	73	Coaly shale
Ma-3	4.97	0.03	1.74	0.58	1.7	0.01	487	35	12	Coaly shale
Ma-4	3.13	0.01	0.09	0.61	0.1	0.01	473	32	19	Coaly shale
Average	4.80	0.02	2.10	2.30	2.12	0.005	460	46	44	

average concentrations of Hf, Nb, Th, Zr, Cu, and Pb. The lowest average values of Ba, Sr, and Ni are found in the Kayadibi area. Sökmen, Tarhanas, and Manas samples show the highest average values. The lowest average Y, Mo, and As values are observed for the Özyurt and Sökmen samples. Kayadibi, Manas, and Özyurt samples show the highest average values of Y, Mo, and As. The average trace element contents of the studied coal samples (Oz-1, Oz-3, Ir-1, Ir-2, Ir-4) and shaly coal samples (Oz-2, Oz-4, Oz-5, Oz-7, Ta-2, Ir-3) from three locations and Eocene aged Sorgun coals (Karayigit et al., 2000a) were compared with those of standard brown coals (Ketris and Yudovich, 2009). The results showed high similarity to the coal samples (Table 2; Figure 3A). Ba and

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Sample	Al ₂ O3	Fe ₂ O ₃	MgO	CaO	$\mathrm{Na_2O}$	K ₂ O	TiO_2	Ba	Co	Cs	Ga	Ηf	Ŋ	Rb	Sr	ЧL	n	>	Μ	Zr	Y	Mo	Cu
Özyurt																							
Oz-1	0.44	1.48	0.23	10.8	0.007	0.07	0.04	37	6.3	0.7	3.3	0.5	2	7	140	1.7	5.7	67	0.9	16.4	11.4	3.2	14.2
Oz-2	0.64	3.4	0.62	5.96	0.011	0.11	0.06	56	16.8	6.0	4.1	1	2.7	11.3	255	2	4.8	91	6.0	28.6	18.1	3.4	24.2
Oz-3	0.27	1.54	0.14	1.31	0.004	0.02	0.02	10	6.1	0.5	0.7	0.3	0.6		80	0.9	1.6	24	0.7	8.6	6.3	2.1	6.6
Oz-4	0.85	2.96	0.77	5.29	0.03	0.12	0.07	54	12.4	1.2	3.8	6.0	2.9	12	369	1.5	7.8	53	0.5	227.8	8.7	2.5	12.8
Oz-5	0.26	2.59	0.11	5.64	0.004	0.04	0.02	16	6.8	0.5	1.2	0.3	1.2	2.9	89.5	1.1	2.2	43	0.5	10.5	7.2	4.6	6.9
Oz-7	2.03	2.36	0.85	3.79	0.045	0.41	0.15	141	3.8	2	9.3	1.8	6.3	49.2	250	4.4	7	74	0.5	60.4	11.8	1.6	16.8
Kayadibi																							
Ir-1	0.56	0.8	0.12	0.51	0.004	0.08	0.03	16	3.1	6.0	2.1	0.5	1.1	8.3	21.4	1.7	1.4	20	0.5	16.5	13.9	3.5	7.2
Ir-2	3.75	0.88	0.65	0.25	0.007	0.74	0.19	46	15.4	6.3	15.3	3.5	10.9	66	110	5.3	10.9	122	0.5	139.5	40.3	4.1	32.3
Ir-3	4.3	1.23	0.73	0.27	0.007	0.83	0.21	53	15.9	7	16.1	3.5	10.8	67.3	120	7.5	8.9	110	0.7	113.7	44	1.8	29.7
Ir-4	0.36	0.98	0.05	0.33	0.004	0.04	0.02	6	2.1	0.5	1.9	0.3	0.8	3.5	15.7	0.8	8	52	0.6	12.7	6.3	7.6	7.5
Tarhanas																							
Ta-1	3.08	4.72	0.4	4.5	0.096	0.58	0.18	110	26.7	3.3	11.7	3	7.1	44.9	264	4.2	5.1	210	0.6	126	14.6	6.6	59.3
Ta-2	1.2	1.45	0.35	18.5	0.015	0.25	0.07	239	7.8	1.6	4.1	6.0	3.4	18.8	656	8.9	2.6	94	0.5	35.6	6.5	2.9	14.3
Ta-3	0.23	0.67	0.49	36.6	0.007	0.05	0.01	56	1.9	0.5	0.5	0.2	0.3	2.8	2149	0.6	1.1	38	0.5	6.8	2.4	0.9	2.3
Ta-4	7.65	0.86	0.25	0.3	0.026	1.11	0.34	311	8	10	35.3	7	18.8	101	246	18.7	5.6	144	4.7	239.5	53.4	0.3	44.3
Sökmen																							
S-1	5.08	4.66	1.83	1.21	0.412	1.1	0.26	451	20.1	9.3	22	3.4	8.6	92	408	6	2.8	196	1.3	133	25	0.6	57
S-3	5.35	4.5	1.77	1.24	0.415	1.1	0.28	453	19.3	9.6	21.9	3.6	8.4	91.2	409	6.2	2.7	197	1.5	132.1	24.9	0.5	57.6
Manas																							
Ma-1	3.52	1.38	0.86	2.59	0.108	0.92	0.16	764	8.2	5.4	15.5	3.9	10.6	118	128	11.9	6.5	88	2.2	134.9	31.6	0.7	68.1
Ma-2	3.14	1.16	0.71	1.03	0.1	0.84	0.15	202	12	5.1	13.1	3.6	9.4	114	173	10.6	8.5	91	2.2	135.8	35.2	1.7	77.2
Ma-3	6.51	1.26	0.32	0.09	0.041	1.3	0.31	287	14.1	5.9	32.3	7.1	19.7	89.1	193	17.9	8.7	128	9.8	250.4	25.9	2.9	28.1
Ma-4	1.4	13.7	0.02	0.01	0.004	0.08	0.1	28	36.2	1	4	2.7	5.1	5.8	28.3	4.1	1.8	96	2.4	101.2	9.9	15.1	56.6
Sorgun Basin ^a	3.23	1.32	0.08	0.57	1	0.18	0.06	107	,	4.8	9.6	1.5	3.4	25	77	21	2.8	22	26	17	3.7	3.7	4.3
Turkey average ^b	2.1	2.3	0.53	1.8	0.11	0.32	0.097	130	10	4.9	5.8	,	4.0	25	210	,	13	65	1	,	8.3	9.8	20
World⁰		1	0.02		0.02	0.01	0.05																
World ^d								150	4.2	0.98	5.5	1.2	3.3	10	120	3.3	2.9	22	1.2	35	8.6	2.2	15

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^aKarayigit et al. (2000a). ^bPalmer at al. (2004). ^cValkovic (1983). ^dKetris and Yudovich (2009).

Table 2. (Continued).

Lu		0.15	0.22	0.09	0.12	0.1	0.22		0.16	0.55	0.63	0.09		0.25	0.13	0.05	0.75		0.4	0.42		0.48	0.52	0.48	0.19	1	1	0.19
Yb		0.81	1.42	0.47	0.74	0.63	1.41		0.99	3.81	3.95	0.53		1.67	0.77	0.24	4.92		2.63	2.62		2.92	3.23	3.05	1.26	I	I	1
Tm		0.15	0.26	0.1	0.13	0.12	0.2		0.17	0.59	0.62	0.08		0.22	0.13	0.05	0.78		0.43	0.41		0.45	0.47	0.46	0.2	1		0.31
Er		1.1	1.75	0.58	0.8	0.61	1.27		1.14	4.12	4.23	0.49		1.41	0.64	0.2	5.27		2.81	2.79		2.87	3.26	2.94	1.28	1	1	0.85
Но		0.39	0.56	0.22	0.33	0.25	0.39		0.39	1.37	1.58	0.2		0.48	0.26	0.07	1.82		0.87	0.9		1.03	1.09	0.97	0.44	ı	ı	0.5
Dy		1.8	3.06	1	1.31	1.09	1.68		1.9	7.32	7.53	0.92		2.09	1.29	0.33	9.57		4.51	4.65		4.87	5.4	4.82	1.97	ı	1	2
Tb		0.35	0.55	0.18	0.24	0.2	0.27		0.33	1.21	1.28	0.16		0.34	0.25	0.07	1.58		0.82	0.77		0.83	0.94	0.8	0.26	ı	1	0.32
Gd		2.11	3.49	1.09	1.29	1.06	1.59		2.09	7.16	7.27	0.98		2.09	2.05	0.47	9.56		4.81	4.74		5.5	6.24	5.11	1.2			2.6
Eu		0.56	0.88	0.26	0.37	0.27	0.39		0.53	1.64	1.67	0.25		0.51	0.55	0.1	2.26		1.11	1.16		1.02	1.26	1.17	0.25			0.5
Sm		2.1	2.68	1.06	1.23	1.09	1.45		2.01	6.53	6.24	0.79		1.9	2.29	0.43	9.81		4.22	4.37		4.96	6.02	5.85	0.87	5.5		1.9
PN		8.6	13.7	3.5	5.1	4.4	7.8		9.3	26.5	28.2	4.1		6	16.9	2	50.3		18.9	19.1		22.7	26.9	33.5	4.4	1	1	11
Pr		2.07	3.18	0.89	1.33	0.98	2.06		2.42	6.83	7.1	1.03		2.38	4.59	0.48	14.32		5.2	4.9		5.53	6.87	9.29	1.24			3.5
		6.61	21.9	5.5	10.9	8.8	20.4		22.6	54	54.9	8.9		19.8	12.2	3.6	140.3		42.9	13.6		38.9	51.5	87.4	11.4	23		22
e e		9.1	11.6	3.3	5.4	1.5	12.4		12.4	29.7	30.7	1.6		1.4	20.2	2.5	59.5		20.6	20.1		9.6	38.5	17.9	7.4	5		0
- -		.1	.2	.4		2	8.		4.	<i>с</i> і		2		8.	.1	Ŀ.	5		5	ت. ر		.6	<u>ь</u>	5.	.5	8.	.2	
s		6 1	2 3	6 8	4 3	2	6		1	4 2	2 1	1 2		3 1	2 1	1 0	3		2 0	1 0		2 0	5 1	6 0	5.7 5	5 2	7 2	84 1
g St		05 0.	07 1.	57 0.	39 1.	23 1.	39 1		06 0.	1 0.	06 0.	1 0.		34 1.	05 0.	02 0.	07 0.		1 0.	1 0.		02 0.	08 0.	07 0.	82 56	0.	11 2.	1 0.
H		0.0	0.0	0.	0	0	0		0.	0.	0.0			0	0	0	0.0		0.	0.		0.0	0.	0.	2.	1	0.	.4 0.
C		0.1	0.3	0.1	0.1	0.1	0.1		0.2	1.2	0.7	0.2		0.1	0.1	0.1	0.4		0.2	0.1		0.7	-	0.2	0.1		1	0.2
Be		-	1	3		5	5		ŝ	1	-			-	2	ŝ	4		9	~		5	5	5	1	1	1.3	1.2
As		49.2	70.4	188.6	196	334.9	165.8		19	7.9	~	28.8		70.5	12.4	6.1	24		12.8	13.4		9.1	15.5	62.4	293.1	33	65	7.6
Ņ		40.4	153.1	49	80.6	27.3	22.5		8.8	35.1	31.1	9		114.3	29.9	4.8	10.8		30.2	29.9		32.4	33.6	33.6	178.1	2.2	150	6
Zn		29	44	23	33	21	22		55	118	139	22		80	35	10	109		112	112		118	128	93	19	18	40	18
Pb		9.8	5.4	1.6	3.3	2.4	6.2		4.1	13.1	8.1	5.3		38.2	5.9	1.1	28		15.8	16.4		27.8	45.2	20.8	282	20	9.3	6.6
Sample	Özyurt	Oz-1	Oz-2	Oz-3	Oz-4	Oz-5	Oz-7	Kayadibi	Ir-1	Ir-2	Ir-3	Ir-4	Tarhanas	Ta-1	Ta-2	Ta-3	Ta-4	Sökmen	S-1	S-3	Manas	Ma-1	Ma-2	Ma-3	Ma-4	Sorgun Basin ^a	Turkey average ^b	World ^d

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Figure 2. Diagram of average major element contents for Özyurt, Tarhanas, Sökmen, Kayadibi, and Manas samples; Sorgun (Karayigit et al., 2000a); Turkey (Palmer et al., 2004); and the world (Valkovic, 1983) coal averages.

Sb are depleted in the coal samples, with a concentration coefficient of <0.5 (CC = ratio of element concentrations in studied samples vs. standard brown coals) (Dai et al., 2015a, 2015c). V, Zn, Ni, and Se are slightly enriched (2 < CC < 5), and As is enriched (CC > 5). The remaining elements (0.5 < CC < 2) are close to the average values for standard brown coals. In shaly coal samples, W is depleted, whereas Co, Cs, Rb, Sr, V, Zr, Zn, Ni, Hg, and Se are slightly enriched. As is enriched, and other elements are close to the average values for standard brown coals (Figure 3B). In the Sorgun coal samples, Zr, Y, Cu, and Ni are depleted; Cs, Rb, Th, Pb, As, and Se are slightly enriched; W is enriched; and the remaining elements are relatively close to the average values for standard brown coals (Figure 3C). The depletion of Ba, Cs, Sr, U, Mo, Ni, Sb, and other elements in the coal samples is similar to the average values for Turkish coals (Figure 3D). The shaly coal samples show depleted contents of Cs, U, Mo, Ni, and Sb, whereas As is slightly enriched with amounts similar to that of Turkish coals in other elements (Figure 3E). Sorgun coals are fairly similar to Turkish coals in terms of Ba, Cs, Nb, Rb, As, and Se contents; the Sorgun coals are slightly enriched with Pb and depleted in other elements (Figure 3F). Accordingly, the studied coal and shaly coal samples are similar to Sorgun coals in Ga, Hf, Nb, Sr, U, Mo, and La, as well as Ba, Ga, Hf, Nb, Sr, U, Mo, Sb, and La samples, respectively.

REE and yttrium (Y) are not significantly affected by sedimentary processes (Taylor and McLennan, 1985; Bhatia and Crook, 1986; Wronkijewicz and Condie, 1987, 1989, 1990; Saydam Eker, 2012). REE and Y are used as geochemical indicators of the sedimentary environment and postsedimentary history of coal deposits (Hower et al., 1999; Seredin and Dai, 2012; Dai et al., 2015b).

The studied coals, namely shaly coals and coaly shales, exhibit similar REE and Y (REY) distribution patterns. The light REY (LREY = La, Ce, Pr, Nd, and Sm) elements are slightly depleted in Kayadibi, Tarhanas, Sökmen, and Manas samples. Medium REY (MREY = Eu, Gd, Tb, Dy, and Y) elements and heavy REY (HREY = Ho, Er, Tm, Yb, and Lu) elements are slightly enriched in Kayadibi, Sökmen, and Manas samples but slightly depleted in the Tarhanas sample. LREY elements are highly depleted in Özyurt samples, whereas MREY and HREY are moderately depleted compared with the upper continental crust (Taylor and McLennan, 1985; Figure 4). Eu in all samples shows slight or no anomalies. Mafic rocks exhibit low LREE/HREE ratios and contain nonanomalous Eu. However, silicic rocks show high LREE/HREE ratios and contain negative Eu anomalies (Cullers and Graf, 1983; Bauluz et al., 2000). In this study, the mean La_N/Lu_N ratios of the samples are <1. The La_N/Lu_N ratios range from 0.38 to 0.58 in Özyurt samples, 0.56 to 0.81 in Kayadibi samples, 0.47 to 0.96 (except Ta2 = 1.6) in Tarhanas samples, 0.50to 0.54 in Sökmen samples, and 0.40 to 0.77 (except Ma-3 = 1.04) in Manas samples. Based on these results, the studied samples were characterized by MREY and HREY enrichment types. Moreover, the Al₂O₃/TiO₂ ratio is a generally useful provenance indicator for sedimentary rocks (Hayashi et al., 1997; He et al., 2010) and sediments associated with coal deposits (Dai et al., 2015a, 2015b). The characteristic Al₂O₃/TiO₂ ratios are 3-8 for sediments



Figure 3. Concentration coefficient (CC) of trace elements in the samples and Sorgun coal samples.

derived from mafic, 8–21 for sediments derived from intermediate, and 21–70 for sediments derived from felsic igneous rocks (Hayashi et al., 1997; Dai et al., 2015a). The Al_2O_3/TiO_2 ratios range from 12.3 to 19.9 in the studied samples (the average Al_2O_3/TiO_2 ratios are 12.3, 19.2, 19.9, 19.3, and 19.5 for Özyurt, Kayadibi, Tarhanas, Sökmen, and Manas, respectively). This result indicates that the sediment source regions for Eocene coals, shaly coals, and coaly shales in the Gümüşhane and Bayburt regions

are characterized by rocks with intermediate or mafic geochemical properties.

In the analyzed samples from the Özyurt area, a weak negative correlation (r = -0.46, P < 0.05) and a strong negative correlation (r = -0.62, P < 0.05) were detected between As and Cd as well as between As and Zn, respectively. However, a strong positive correlation (r = 0.85, P < 0.05) exists between Cd and Zn. Shaly coal and coaly shales from Tarhanas demonstrate a weak positive



Figure 4. Distribution patterns of rare earth elements and yttrium in the Özyurt, Kayadibi, Tarhanas, Sökmen, and Manas samples. REY elements are normalized by Upper Continental Crust (UCC) (after Taylor and McLennan, 1985).

correlation (r = 0.55, P < 0.05) between As and Zn and a strong positive correlation (r = 0.76) between Cd and Zn; however, no relationship was observed between As and Cd. In Kayadibi samples, a strong negative correlation (r = -0.82, P < 0.05) and a very strong negative correlation (r = -0.98, P < 0.05) were observed between As and Cd and between As and Zn, respectively. In addition, a strong positive correlation (r = 0.80, P < 0.05) was observed between Cd and Zn. In the Manas area, a strong negative correlation (r = -0.74, P < 0.05) exists between As and Cd, a very strong negative correlation (r = -0.99, P < 0.05) was found between As and Zn, and a strong positive correlation (r = 0.83) was found between Cd and Zn. In the studied samples from Özyurt, Kayadibi, and Manas, the negative correlation between As and Cd as well as between As and Zn indicates a possible organic origin for As and a mineral origin for Cd and Zn. Based on this assumption, all three elements (As, Cd, and Zn) in the Tarhanas samples exhibit inorganic origins. Copper in the investigated samples is generally assumed to be associated with chalcopyrite and pyrite (Swanie, 1990; Finkelman, 1995).

Nickel is associated with both organic (Swaine, 1990; Orem and Finkelman, 2003) and inorganic (Finkelman, 1995) materials as well as with sulfites (Querol et al., 1998; Spears and Zheng, 1999; Goodarzi, 2002; Ribeiro et al., 2010). In the samples of Özyurt, a weak positive correlation (r = 0.54, P < 0.05) was found between Cu and Ni, and a strong positive correlation (r = 0.70, P <0.05) exists between Pb and Ni. In the Tarhanas area, Cu and Ni show a strong positive correlation (r = 0.72, P <0.05), whereas Cu and Pb exhibit a very strong positive correlation (r = 1.00, P < 0.05). Very strong correlations were also observed between Cu and Ni (r = 1.00, P < 0.05) and between Cu and Pb (r = 0.89, P < 0.05) in Kayadibi samples. However, no correlation was observed between these element pairs in the Manas samples. These values indicate that Cu, Ni, and Pb elements in all coal, shaly coal, and coaly shale samples (except the Manas area) may have inorganic origins. Furthermore, the existence of a very strong correlation between Ni and As in samples from Tarhanas confirms that the latter has inorganic origin. In most cases, As in the studied samples is associated with epigenetic cleat and fracture-filling pyrite. In some cases, As is associated with fine-grained, syngenetic pyrite and occurs in arsenopyrite (Finkelman, 1994; Karayigit et al., 2000b). U (Finkelman, 1995) and V in the studied samples may have been derived from both organic materials and clays (Finkelman, 1995; Querol et al., 1996; Goodarzi, 2002; Ribeiro et al., 2010).

In the Özyurt samples, a strong positive correlation was observed between V and Ba (r = 0.60, P < 0.05), and a weak positive correlation exists between the element pairs of V–U, V–K, and V–Al (r = 0.59, r = 0.47, and r = 0.49, respectively). Strong positive correlations are also observed between U–Ba, U–K, and U–Al, with r = 0.70, r = 0.67, and r = 0.62, respectively. In the Tarhanas samples, weak positive correlations (r = 0.54, P < 0.05) were observed between V and Al, a strong positive correlation (r = 0.64, P < 0.05) was found between V and K, and a very strong positive correlation (r = 0.89, P < 0.05) exists between V and U; however, no correlation was detected between V and Ba. In these samples, weak positive correlations were observed between U and Ba (r = 0.55, P < 0.05) and between U and Al (r = 0.54, P < 0.05). In the Kayadibi samples, very strong positive correlations were found between element pairs of V-Ba, V-U, V-K, and V-Al, with r = 0.89, r = 0.90, r = 0.93,

and r = 0.93, respectively. In these samples, strong positive correlations were observed between U and Ba (r = 0.60, P < 0.05), U and K (r = 0.68, P < 0.05), and U and Al (r = 0.68, P < 0.05), and U and Al (r = 0.68, P < 0.05). In the Manas samples, a very weak positive correlation was detected between V and U (r = 0.35, P < 0.05), a weak positive correlation was found between V and K (r = 0.53, P < 0.05), and a strong positive correlation was determined between V and Al (r = 0.81, P < 0.05); conversely, no correlation exists between V and Ba. In the Manas area, V and Ba show a weak positive correlation (r = 0.35, P < 0.05), V and Al exhibit a strong positive correlation (r = 0.78, P < 0.05), and V and K demonstrate a very strong positive correlation (r = 0.93, P < 0.05). These results indicate that U and V in the investigated samples have silicate origins.

4.3. Molecular geochemistry of coal extracts

4.3.1. Isoprenoids and n-alkanes

In the Gümüşhane region, n-alkanes are recorded within the C_{14} - C_{35} range in the gas chromatogram of the Oz-1

coal sample from the Özyurt area. Among n-alkanes, C_{29} n-alkane has the maximum peak value. However, the highest peak pristane (Figure 5A) Pr/Ph value of the Oz-1 coal sample was calculated as 13.05 using gas chromatograms (Table 3). The carbon preference index (CPI) was calculated using n-alkanes in the range of C_{25} - C_{30} (Tissot and Welte, 1984; Barker, 1986; Marzi et al., 1993; Peters and Moldowan, 1993) and C_{23} - C_{29} (Bray and Evans, 1961) of the gas chromatograms. Accordingly, CPI values 1 and 2 of coal sample Oz-1 were determined as 1.64 and 1.38, respectively (Table 3).

N-alkanes in the range of $C_{14}-C_{35}$ are recorded in gas chromatograms of the Ta-2 shaly coal sample from the Tarhanas area. Maximum peak values belong to C_{21} , C_{22} , and C_{23} n-alkanes (Figure 5B). The Pr/Ph ratio of coal sample Ta-2 is calculated as 3.6; the CPI values (1.05 and 1.02; Table 3) indicate that n-alkanes with odd carbon numbers and n-alkanes with even carbon numbers have almost the same values. In gas chromatograms of the Ir-2



Figure 5. Gas chromatograms of extracts from the selected samples.

Sample ID	CPI1	CPI2	Pr/Ph	Pr/n-C ₁₇	Ph/n-C ₁₈	TAR
Gümüşhane		-			-	
Ir-2	3.155	2.687	8.557	13.18	1.21	15.86
Oz-1	1.644	1.379	13.053	19.44	2.13	5.59
Ta-2	1.054	1.02	3.615	0.99	0.21	1.5
Bayburt						
Ma-2	1.332	1.208	12.902	0	0.42	4.32

Table 3. The parameters calculated from gas chromatograms for selected samples.

$$\begin{split} & \text{CPI1} = \frac{1}{2} * \{ [(C_{25} + C_{27} + C_{29}) / (C_{24} + C_{26} + C_{28})] + [(C_{25} + C_{27} + C_{29}) / (C_{26} + C_{28} + C_{30})] \} \text{ (Tissot and Welte, 1984).} \\ & \text{CPI2} = \{ [(C_{23} + C_{25} + C_{27}) + (C_{25} + C_{27} + C_{29})] / [2 * (C_{24} + C_{26} + C_{28})] \} \text{ (Bray and Evans, 1961).} \\ & \text{TAR} = (C_{27} + C_{29} + C_{31}) / (C_{15} + C_{17} + C_{19}) \text{ (Bourbonniere and Meyers, 1996).} \end{split}$$

coal sample from the Kayadibi area, n-alkanes in the range of $C_{16}-C_{35}$ are recorded. NC₂₇ forms the maximum peak value (Figure 5C). The Pr/Ph ratio of the Ir-2 sample was calculated as 8.56. In this coal sample, CPI1 and CPI2 values were determined as 3.16 and 2.69, respectively (Table 3).

In the Bayburt region, n-alkanes were recorded in the range of C_{18} - C_{35} in gas chromatograms of the Ma-2 coaly shale sample from the Manas area (Figure 5D). The maximum peak value belonged to alkane C_{29} . The Pr/Ph ratio of the Ma-2 coaly shale sample was calculated as 12.9. Using gas chromatograms, CPI values 1 and 2 were calculated as 1.3 and 1.2, respectively (Table 3), indicating that n-alkanes with odd carbon numbers are slightly more dominant compared with n-alkanes with even carbon numbers.

4.3.2. Biomarkers

4.3.2.1. Terpanes

The relative abundance and distribution of terpanes obtained from m/z mass chromatograms of investigated coal and shaly coal samples are given in Figures 6A and 6B, and their calculated parameters are presented in Table 4. Whereas tricyclic terpanes are insufficiently recorded in sample Oz-1, tricyclic terpanes with lower numbers were recorded as more dominant in sample Ta-2. Accordingly, the C_{25}/C_{26} tricyclic terpane ratio cannot be calculated in sample Oz-1, but this value was calculated as 0.72 in sample Ta-2. In sample Oz-1, Tm (C₂₇ 17a(H)-22,29,30trisnorhopane) is more abundant compared with Ts (C_{27} 18a(H)-22,29,30-trisnorneohopane); in sample Ta-2, Ts is more abundant than Tm. The C₂₉/C₃₀ hopane ratios in samples Oz-1 and Ta-2 are calculated as 0.76 and 0.62, respectively. The $C_{31}R$ homohopane/ C_{30} hopane ratio was calculated as 0.36 in sample Oz-1 and as 0.16 in sample Ta-2. In samples Oz-1 and Ta-2, C₃₁ homohopane is more dominant compared with C32-C35 homohopanes. A steady decrease is observed in peak height from the C_{31} member toward C_{35} (Figures 6A, 6B, and 7). The C_{35}/C_{31} -C₃₅ homohopane ratios of samples Oz-1 and Ta-2 were determined as 0.011 and 0.039, respectively. The C₃₅/C₃₄ homohopane ratio was calculated as 0.39 in sample Oz-1 and as 0.55 in sample Ta-2. The moretane/hopane ratio was determined as 0.50 in sample Oz-1 and as 0.10 in sample Ta-2. In the Ta-2 sample, the (oleanane/ C_{30} hopane) × 100 rate is calculated as 11.9%, but the ratio is unmeasurable in sample Oz-1.

4.3.2.2. Steranes

The distributions and relative abundances of steranes obtained from m/z 217 ion chromatograms are given in Figure 8; the parameters calculated using these chromatograms are presented in Table 5. Based on their relative abundances, C_{27} - C_{28} - C_{29} steranes of sample Oz-1

Table 4. Calculated parameters from m/z 191 mass chromatograms for selected samples.

Sample ID	1	2	3	4	5	6	7	8	9	10
Oz-1	0.76	0.36	4.55	0.39	0.48	0.015	0.50	ND	ND	ND
Ta-2	0.62	0.16	0.81	0.55	0.61	0.64	0.10	0.50	11.9 %	0.72

1) C_{29}/C_{30} hopane, 2) C_{31} R homohopane/ C_{30} hopane, 3) $(C_{19+}C_{20})/C_{23}$ tricyclic terpane, 4) C_{35}/C_{34} homohopane, 5) 22S/(22S+22R) homohopane (for C_{32}), 6) Ts/(Ts+Tm), 7) moretane/hopane (for C_{30}), 8) (gammacerane/ C_{30} hopane) × 10, 9) (oleanane/ C_{30} hopane) × 100, 10) C_{25}/C_{26} tricyclic terpane.



Figure 6. The m/z 191 mass chromatograms of selected extracted samples.

are ranked as $C_{29} > C_{28} > C_{27}$, and those of sample Ta-2 as $C_{29} > C_{27} > C_{28}$ (Figures 8A, 8B, 9A, and 9B). The C_{28}/C_{29} sterane ratios in samples Oz-1 and Ta-2 are calculated respectively as 0.44 and 0.36. Diasterane/sterane ratios of samples Oz-1 and Ta-2 were calculated as 0.3 and 1.3, respectively.

4.4. Type of organic matter

 $\rm HI-T_{max}$ (Mukhopadyay et al., 1995) and HI-TOC (Subroto et al., 2010) diagrams were used to determine organic material types of coal, shaly coal, and coaly shale samples from Özyurt, Tarhanas, Sökmen, Kayadibi, and Manas (Figures 10 and 11). In the HI-T_{max} diagrams, two samples



Figure 7. C_{31} - C_{35} (22R-22S) homohopane distributions of two of the studied coal and shaly coal samples.

from Tarhanas and one sample each from Özyurt and Kayadibi fall within the type II area, and the other samples fall within the type III area. In the HI-TOC diagram, one sample from the Tarhanas area is classified as type II and the remaining sample plots fall in the type III area. N-alkane distribution of samples is bimodal, and n-alkanes with high carbon numbers are dominant compared with n-alkane with low carbon numbers. Accordingly, the depositional environment may be dominated by terrestrial organic matter input (Tissot and Welte, 1984; Waples, 1985). The calculated CPI > 1 values in Ir-2, Oz-1, and Ma-2 samples as well as CPI \geq 1 in the Ta-2 sample (Table 3) confirm the presence of terrestrial organic matter (Tissot et al., 1987). A high Pr/nC_{17} ratio (>0.6) is observed in the samples (Table 3), thereby confirming the presence of terrigenous organic matter (Peters and Moldowan, 1993). In addition, the TAR results show the ratio of terrestrial organic matter to aquatic organic substance (Bourbonniere and Meyers, 1996; Peters et al., 2005; Varandas da Silva, 2008). TAR values were determined as TAR > 1 in four samples (Ir-2, Oz-1, Ta-2, and Ma-2) (Table 3). The oleanane amount of Oz-1 from Özyurt is immeasurable. However, the oleanane index of Ta-2 from Tarhanas was calculated as 11.9% (Table 4), confirming the terrestrial organic matter input (Hunt, 1995). The relative proportions of C_{27} - C_{29} regular steranes in living organisms are related to particular environments and show that steranes in sediments may supply valuable paleoenvironment information (Huang and Meinschein, 1979). A powerful terrestrial contribution is exhibited by a predominance of C₂₉ steranes because the marine effect is indicated by the domination of C₂₇ sterols. C₂₈ occurs less frequently than C_{27} and C_{29} and generally characterizes lacustrine environments (Huang and Meinschein, 1979). Nishimura (1982) emphasized that some C_{28} and C_{27} steranes can be derived from plankton or algae in lacustrine environments. The C_{29} sterane is abundant in Oz-1 and Ta-2 samples compared with C_{27} sterane. Accordingly, the investigated samples are mainly composed of higher terrestrial plant forms (Figures 9A and 9B). Humic and waxy coals often indicate a strong predominance of C_{29} (Farhaduzzaman et al., 2012).

4.5. Maturity of organic matter

Depending on T_{max} values, coal and shaly coal samples from Özyurt thermally change from immature to mature. The coaly shale samples from Sökmen and the coal and shaly coal samples from Kayadibi change from immature to extremely mature. The shaly coals and coaly shales from Tarhanas and Manas change from mature to extremely mature (Figure 10).

Thermal maturity was determined by calculating the isoprenoid/n-alkane ratios obtained from the gas chromatograms of coal, shaly coal, and coaly shale samples of Ir-2, Oz-1, Ta-2, and Ma-2 (Table 3). The Pr/n- C_{17} ratio could not be calculated because the $n-C_{17}$ of the Ma-2 sample cannot be measured, but the Ph/n- C_{18} ratio is 0.42. The Pr/n- C_{17} and Ph/n- C_{18} ratios of sample Ta-2 are 0.99 and 0.21, respectively. These findings confirm that both samples are mature. The calculated Pr/n- C_{17} ratios of samples Ir-2 and Oz-1 are 13.18 and 19.44, respectively, and their respective Ph/n- C_{18} ratios are 1.21 and 2.13. The T_{max} values are within the limit of maturity (i.e. 432 °C and 434 °C), depending on the oxidative nature of the deposition environment; thus, high Pr/n- C_{17} ratio can be related to increase in Pr value.

The 22S/(22S + 22R) homohopane (C₃₂) was used as a maturity parameter because the 22S isomer is more resistant to temperature increase than the 22R isomer (0.57, 0.62 = equilibrium; Seifert and Moldowan, 1986).The calculated 22S/(22S + 22R) homohopane (C₃₂) ratios of samples Oz-1 and Ta-2 are 0.48 and 0.61, respectively. These values show that the homohopane conversion does not fully reach equilibrium in sample Oz-1, in contrast to sample Ta-2 (Seifert and Moldowan, 1986). The 20S isomer increases compared with the 20R isomer with increasing maturity of the 20S/(20S + 20R) sterane (C₂₀) ratio; the 20S/(20S + 20R) reaches equilibrium (0.55) as the maturity continues to increase. This equilibrium value corresponds to the petroleum formation peak (Gürgey, 1999). The bb/(bb+aa) sterane (C_{29}) ratio of sample Oz-1 could not be calculated, but its 20S/(20S + 20R) sterane (C_{20}) ratio is 0.18. When the 20S/(20S + 20R) sterane (C_{20}) ratio is considered, the sample in question is not thermally mature. The calculated 20S/(20S + 20R) sterane (C₂₀) and bb/(bb+aa) sterane (C20) ratios of sample Ta-2 are 0.53 and 0.56, respectively; these values are approximate to the equilibrium value.

The moretane/hopane ratio of sample Oz-1 is higher than 0.15, but that of sample Ta-2 is lower than 0.15 (Table 4).



Figure 8. The m/z 217 mass chromatograms of selected extracted samples.

Accordingly, sample Ta-2 is thermally mature (Waples and Machihara, 1991). In addition, the high diasterane/sterane ratio of sample Ta-2 (Table 5) confirms its thermal maturity (Peters et al., 2005b), which is interpreted using the Ts/(Ts + Tm) rate. The Ts/(Ts + Tm) ratios of samples Oz-1 and Ta-2 are 0.015 and 0.64, respectively (Table 4). The high Ts/ (Ts + Tm) ratio of sample Ta-2 confirms that the sample is thermally mature.

Hanson et al. (2001) identified upper Oligocene lacustrine oils from the northern Qaidam Basin of Northwest China, with C32, C33, and C34 22S/(22S + 22R) of < 0.50; low C29 sterane 20S/(20S + 20R); and low Ts/Tm support with considerably low maturity (Peters at al., 2005b). Thus, the thermal maturity of all parameters should be interpreted. In this context, Pr/n-C₁₇ and Ph/ n-C₁₈ moretane/hopane and diasterane/sterane ratios

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Fabl	e 5.	Cal	culat	ed p	varameters	from	m/z 2	17	mass c	hroma	tograms	for se	lected	sampl	les.
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Sample ID	1	2	3	4	5	6
Oz-1	15.5, 25.9, 58.6	0.18	ND	0.44	0.34	0.17
Ta-2	26.7, 19.5, 53.8	0.53	0.56	0.36	1.30	0.24

1) C_{27} , C_{28} , C_{29} steranes (%); 2) 20S/(20S+20R) sterane (for C_{29}); 3) bb/(bb+aa) sterane (for C_{29}); 4) C_{28}/C_{29} sterane; 5) diasterane/sterane (for C_{27}); 6) sterane/hopane.



Figure 9. A) Relationship between sterane compositions, source input, and depositional environment, by which two samples (Oz-1, Ta-2) are shown to be dominated by terrigenous organic matter input (after Farhaduzzaman et al., 2012). **B)** Pr/nC_{17} vs. Ph/ nC_{18} for three samples infer oxicity and organic matter type of the source rock depositional environment (after Peters et al., 2005a, 2005b; Koeverden et al., 2011).

as well as the T_{max} value confirm the thermal maturity of sample Ta-2.

In the studied region, volcanic activity and sedimentation occur at various locations during the Eocene (Saydam Eker, 2012). This volcanic activity could be due to differences in the thermal maturity of the samples.

4.6. Depositional environment

Eocene aged coals, shaly coals, and coaly shales exposed in the Gümüşhane and Bayburt regions have high Pr/Ph ratios, with values between 3.615 and 13.053. The high value of Pr/Ph indicates the presence of an environment with overall oxygen atmosphere during the deposition of organic material (Mello and Maxwell, 1990; Philp, 1994; Huang et al., 2003). Peters and Moldowan (1993) reported that samples with high Pr/Ph ratios (>3) within the oilgenerative window exhibit terrestrial organic matter input under oxic conditions. Coals with a Pr/Ph ratio of >4 are characteristically precipitated under a peat-swamp depositional setting and oxic depositional conditions (Wan Hasiah and Abolins, 1998). These findings are consistent with the depositional environments of the studied coal, shaly coal, and coaly shale samples, three of which have Pr/Ph > 4 and one of which has Pr/Ph > 3.

In addition to hopane/sterane and Pr/Ph diagrams (Subroto et al., 2010), sample Oz-1 falls into highly oxidizing terrestrial areas, whereas sample Ta-2 falls into anoxic-suboxic areas with active terrestrial influence (Figure 12). The absence of the anoxic biomarker $(17\alpha(H)-$ 28,30-bisnorhopane) indicates that the studied coals are deposited in an oxic-suboxic environment (Katz and Elrod, 1983; Peters et al., 2005b; Hoş-Çebi and Korkmaz, 2015). The high sterane/hopane ratio for coals would likely characterize a marine environment (Mann et al., 1998). The sterane/hopane ratios of the analyzed coal and shaly coal are very low at 0.17 and 0.24 for the Oz-1 and Ta-2 samples, respectively. The ratios are indicative of dominantly terrestrial and microbial organic matter input in depositional environments. Peters and Moldowan (1993) suggested that a high $(C_{19}+C_{20})/C_{23}$ tricyclic terpane ratio indicates terrestrial organic matter input.



Figure 10. Distribution of HI vs. T_{max} for the studied samples (Mukhopadyay et al., 1995).

The $(C_{19}+C_{20})/C_{23}$ tricyclic terpane ratios of the Oz-1 and Ta-2 samples (4.55 and 0.81, respectively) reflect terrestrial organic matter input. Additionally, the abundances of C_{27} , C_{28} , and C_{29} are in the following order: $C_{29} > C_{28} > C_{27}$ in Oz-1 and $C_{29} > C_{27} > C_{28}$ in Ta-2. The dominance of C_{29} sterane indicates a terrestrial organic matter input (Huang and Meinschein, 1979; Robinson, 1987).

 C_{30} sterane was also recorded in both samples. This finding is a sign that both fields are influenced by marine conditions (Moldowan et al., 1985; Peters, 1986; Peters and Moldowan, 1993; Hunt, 1995; Mann et al., 1998). The gammacerane index is 0.50 in the Ta-2 sample. A low gammacerane index indicates the presence of at least some salts in the depositional environment of sample Ta-2 (Comet and Eglinton, 1987; Routh et al., 1999).

The high Pr/Ph ratios, high $(C_{19}+C_{20})/C_{23}$ tricyclic terpane ratio (Oz-1), high abundance of C_{29} , high sterane/diasterane ratios, absence of an anoxic biomarker (17 α (H)-28,30-bisnorhopane), and low sterane/hopane ratios indicate that coaly organic matter is derived from higher land plants in terrestrial environments under oxic and suboxic depositional conditions. The presence of biomarkers such as C_{30} sterane (Oz-1 and Ta-2 samples) and gammacerane (Ta-2 sample) indicates the presence of a marine environmental influence on coal and shaly coal samples.



Figure 11. Distribution of the analyzed samples into HI vs. TOC plots (Subroto et al., 2010).



Figure 12. Distribution of the selected samples into total hopanes/steranes vs. Pr/Ph plots (after Subroto et al., 2010).

Ba and Sr concentrations are higher in seawater than in freshwater (Reimann and Caritat, 1998; Li et al., 2016). Furthermore, the Sr/Ba ratio (>1) suggests the involvement of a marine influence on coal formation (Shao et al., 1998). Moreover, Ba and Sr contents of the investigated samples range from 9 ppm to 764 ppm and from 15.7 ppm to 2149 ppm, respectively (Table 2). The Sr/Ba ratios of these samples fluctuate between 0.17 and 38.4. These results indicate that the depositional environment of marine and terrestrial conditions is effective.

Eocene aged clastic rocks of the eastern Pontides (NE Turkey) show two different source characteristics. Volcaniclastic deposits are dominant in the northern section of Gümüşhane, and siliciclastic deposits are dominant in the southern section (Kelkit, Köse) and the Bayburt region. The studied coaly sequences were measured from siliciclastic deposits, which likely are lagoon environments (Figure 13). However, Eocene aged Yeniçeltek-Amasya (Central Black Sea Region) and Salıpazarı-Bolu (Western Black Sea Region) coals are suboxic in lacustrine or lacustrine swamps, whereas Aspiras-Kastamonu (Western Black Sea Region) coals are in paralic and suboxic brack water swamps, and coals contain terrestrial organic matter (Hoş-Çebi and Korkmaz, 2013) (Figure 9A).

The investigated coals, shaly coals, and coaly shales were probably deposited in an environment dominated by marine and terrestrial settings with oxic or suboxic conditions. These samples comprise terrestrial organic matter.

4.7. Hydrocarbon potential of Eocene aged coal

This study generally included coals, shaly coals, and coaly shales that belong to type III kerogen (Figures 10 and 11). The average PY values of Özyurt and Kayadibi coals and shaly coals were high (36.8 mg HC/g rock and 65.8 mg HC/g rock, respectively), whereas the PY values of coals belonging to other areas were low. The average HI values of samples belonging to all areas were determined as <200 mg HC/g TOC. Based on these data, Eocene aged coals, shaly coals, and coaly shales exposed in the Gümüşhane and Bayburt regions do not have the potential to form liquid hydrocarbons, but they might be able to produce gas. Eocene aged coals in Yeniçeltek (Amasya) and Salıpazarı (Bolu) with type II kerogen can generate oil; in Aspiras (Kastamonu) with type III kerogen they can generate gas (Hoş-Çebi and Korkmaz, 2013).

5. Conclusions

The average TOC values of Eocene aged samples from the Özyurt, Tarhanas, Sökmen, Kayadibi, and Manas areas were calculated to be 43.89%, 13%, 0.54%, 53.83%, and 4.80%, respectively. Coals and shaly coals from the Kayadibi area had the highest average PY value of 65.80 mg HC/g rock, whereas coaly shales from the Sökmen area had the lowest average PY value of 0.12 mg HC/g rock. The highest average HI value was calculated as 112 mg HC/g TOC for Tarhanas shaly coals and coaly shales, whereas the lowest average HI value was calculated as 24 mg HC/g TOC for Sökmen coaly shales.

The major element contents of the studied samples were compared with those of samples worldwide. Several similarities were demonstrated. Özyurt samples showed similarities in terms of Al, Na, and Ti; coaly shales from Sökmen in terms of Ca; Kayadibi samples in terms of Fe and Ti; and Manas shaly coals in terms of Na and Ti. Coaly samples from the abovementioned areas were enriched in terms of other major elements. Fe and Mg contents



Figure 13. Schematic diagram represents the paleodepositional environment of Eocene coals, shaly coals, and coaly shales in the Gümüşhane and Bayburt regions.

of Özyurt; Al, Mg, K, and Ti contents of Tarhanas and Kayadibi; and Al, Fe, Mg, K, and Ti contents of Manas show similarities to Turkish coals. In the studied samples, Co, Cs, Ga, Hf, Th, U, Y, Mo, Be, Cd, Sb, and La generally are similar to the average values for brown coals. The samples are close to the average values for Turkish coals in Co, Ga, Nb, Rb, V, Y, Cu, Pb, Zn, As, Be, and Se. As mean La_N/Lu_N ratios of studied samples were <1, these samples were characterized by MREY and HREY enrichment types. The Al_2O_3/TiO_2 ratios of the coal samples indicated that the sediment source region for Eocene coals, shaly coals, and coaly shales in the Gümüşhane and Bayburt regions were likely characterized by rocks with intermediate or mafic geochemical characteristics.

As in the Özyurt, Kayadibi, and Manas samples is possibly of organic nature, whereas As in the Tarhanas samples possibly has inorganic origin. Cd, Zn, Cu, Ni, Pb (sulfide), V, and U (silicates, in particular clay minerals) were assumed to be derived from mineral substances.

N-alkanes with high carbon numbers were more dominant compared with n-alkanes with low carbon numbers. The calculated CPI > 1 and TAR > 1 values of the analyzed samples indicated the presence of terrestrial organic material in the depositional environment. Moreover, the great majority of samples belonging to these five areas contain type III kerogen. The biomarker data indicated that the studied coal and shaly coal samples dominantly contained terrestrial and microbial organic matter.

According to T_{max} values, samples from the Özyurt, Sökmen, and Kayadibi areas changed from immature to mature, whereas the samples from Tarhanas and Manas changed from mature to extremely mature. The determined CPI > 1 and TAR > 1 values of samples Ta-2 and Ma-2 confirm that these shaly coal and coaly shales were thermally mature. Meanwhile, the calculated Ph/ n-C₁₈ > 1 value of samples Ir-2 and Oz-1 showed that these coals were thermally immature. The calculated 22S/(22S + 22R) homohopane (C₃₂), 20S/(20S + 20R) sterane (C₂₉), and Ts/(Ts + Tm) ratios of sample Oz-1 were 0.48, 0.18, and 0.015, respectively. Given that these values did not reach equilibrium (0.55), the said coals were immature. The 22S/ (22S + 22R) homohopane (C₃₂), 20S/(20S + 20R) sterane (C₂₉), and Ts/(Ts + Tm) of sample Ta-2 were calculated as 0.61, 0.56, and 0.64, respectively, emphasizing that Tarhanas shaly coals were thermally mature.

The calculated Pr/Ph ratio (3.615) of sample Ta-2 showed that the environment was suboxic during the deposition of organic material. The determined Pr/Ph ratios for samples Ir-2, Oz-1, and Ma-2 were 8.557, 13.053, and 12.902, respectively; these values indicated that the environment was oxic. The absence of anoxic biomarkers indicated that the studied samples were deposited in a terrestrial environment under oxic-suboxic conditions. The low sterane/hopane ratios of the abovementioned samples indicated that the environment was terrigenous. However, the presence of C₃₀ sterane and gammacerane in samples pointed out that the depositional environment was under the influence of the sea. In addition, the Sr/Ba ratios of the samples ranged from 0.17 to 38.4, indicating that the depositional environment of marine and terrestrial conditions were effective.

References

- Arslan M, Tüysüz N, Korkmaz S, Kurt H (1997). Geochemistry and petrogenesis of the Eastern Pontide volcanic rocks, Northeast Turkey. Chemie der Erde 57: 157-187.
- Barker C (1986). Organic geochemistry in petroleum exploration. AAPG Bull 10: 39-41.
- Bauluz B, Mayayo MJ, Fernandez-Nieto C, Lopez JMG (2000). Geochemistry of Precambrian and Paleozoic siliciclastic rocks from the Iberian Range (NE Spain): implications for sourcearea weathering, sorting, provenance, and tectonic setting. Chem Geol 168: 135-150.
- Bektaş O, Yılmaz C, Taslı K, Akdağ K, Özgür S (1995). Cretaceous rifting of the eastern Pontide carbonate platform (NE Turkey): the formation of carbonates breccias and turbidites as evidences of a drowned platform. Geologia 57: 233-244.
- Bhatia MR, Crook KAW (1986). Trace element characteristics of graywackes and tectonic setting discrimination of sedimentary basins. Contrib Mineral Petrol 45: 97-113.
- Bourbonniere RA, Meyers PA (1996). Sedimentary geolipid records of historical changes in the watersheds and productivities of lakes Ontario and Erie. Limn Ocean 41: 352-359.
- Bray EE, Evans ED (1961). Distribution of n-paraffins as a clue to recognition of source beds. Geochim Cosmochim Ac 22: 2-15.
- Çoğulu E (1975). Gümüşhane ve Rize granitik plutonlarının mukayeseli petrojeolojik ve jeokronolojik etüdü. PhD, İstanbul Technical University, İstanbul, Turkey (in Turkish).
- Comet PA, Eglinton G (1987). The use of lipids as facies indicators. In: Fleet AJ, editor. Marine Petroleum Source Rocks. London, UK: Geological Society Special Publications, pp. 99-117.
- Cullers RL, Graf J (1983). Rare earth elements in igneous rocks of the continental crust: intermediate and silicic rocks, ore petrogenesis. In: Henderson P, editor. Rare-Earth Geochemistry. Amsterdam, the Netherlands: Elsevier, pp. 275-312.

According to geochemical data, Eocene aged coals, shaly coals, and coaly shales exposed in the Gümüşhane and Bayburt regions exhibited similar properties; the coal samples were likely to be humic coals that were thermally immature–mature. The studied samples were probably deposited under an oxic-suboxic mixture of marine and terrestrial environment conditions, comprising terrestrial organic matter without the potential to form hydrocarbon.

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- Dai S, Hower JC, Ward CR, Guo W, Song H, O'Keefe JMK, Xie P, Hood MM, Yan X (2015a). Elements and phosphorus in the middle Jurassic inertinite-rich coals of the Muli Coalfield on the Tibetan Plateau. Int J Coal Geol 144-145: 23-47.
- Dai S, Li T, Jiang Y, Ward CR, Hower JC, Sun J, Liu J, Song H, Wei J, Li Q et al. (2015b). Mineralogical and geochemical compositions of the Pennsylvanian coal in the Hailiushu Mine, Daqingshan Coalfield, Inner Mongolia, China: implications of sediment source region and acid hydrothermal solutions. Int J Coal Geol 137: 92-110.
- Dai S, Seredin VV, Ward CR, Hower JC, Xing Y, Zhang W, Song W, Wang P (2015c). Enrichment of U-Se-Mo-Re-V in coals preserved within marine carbonate successions: geochemical and mineralogical data from the Late Permian Guiding Coalfield, Guizhou, China. Miner Deposita 50: 159-186.
- Dokuz A (2011). A slab detachment and delamination model for the generation of carboniferous high-potassium I-type magmatism in the eastern Pontides, NE Turkey: the Köse Composite Pluton. Gondwana Res 19: 926-944.
- Eyuboglu Y, Bektas O, Seren A, Maden N, Jacoby WR, Özer R (2006). Three axial extensional deformation and formation of Liassic rift basins in the Eastern Pontides (NE Turkey). Geol Carpath 57: 337-346.
- Eyuboglu Y, Dilek Y, Bozkurt E, Bektas O, Rojay B, Şen C (2010). Structure and geochemistry of an Alaskan-type ultramafic-mafic complex in the eastern Pontides, NE Turkey. Gondwana Res 18: 230-252.
- Farhaduzzaman M, Abdullah WH, Islam AA (2012). Depositional environment and hydrocarbon source potential of the Permian Gondwana coals from the Barapukuria Basin, Northwest Bangladesh. Int J Coal Geol 90-91: 162-179.
- Finkelman RB (1994). Modes of occurrence of potentially hazardous elements in coal: level of confidence. Fuel Proc Technol 39: 21-34.

- Finkelman RB (1995). Modes of occurrences of environmentallysensitive trace elements in coal. In: Swaine DJ, Goodarzi F, editors. Environmental Aspects of Trace Elements in Coal. Dordrecht, the Netherlands: Kluwer Academic Publishers, pp. 24-44.
- Goodarzi F (2002). Mineralogy, elemental composition and modes of occurrence of elements in Canadian feed-coals. Fuel 81: 1199-1213.
- Gürgey K (1999). Geochemical characteristics and thermal maturity of oils from the Thrace basin (Western Turkey). J Petrol Geol 22: 167-189.
- Güven IH (1993). 1/100.000 Scaled Geological Map Series of Turkey. No. 57-60. Ankara, Turkey: MTA Publications.
- Hanson AD, Ritts BD, Zinniker D, Moldowan JM, Biffi U (2001). Upper Oligocene lacustrine source rocks and petroleum system of the northern Qaidam Basin, northwest China. AAPG Bull 85: 601-619.
- Hayashi KI, Fujisawa H, Holland HD, Ohmoto H (1997). Geochemistry of ~1.9 Ga sedimentary rocks from northeastern Labrador, Canada. Geochim Cosmochim Ac 61: 4115-4137.
- He B, Xu YG, Zhong YT, Guan JP (2010). The Guadalupian– Lopingian boundary mudstones at Chaotian (SW China) are clastic rocks rather than acidic tuffs: implication for a temporal coincidence between the end-Guadalupian mass extinction and the Emeishan volcanism. Lithos 119: 10-19.
- Hoş-Çebi F, Korkmaz S (2013). Organic geochemistry and depositional environments of Eocene coals in northern Anatolia, Turkey. Fuel 113: 481-496.
- Hoș-Çebi F, Korkmaz S (2015). Organic geochemistry of Ağaçbaşı Yayla Peat Deposits, Köprübaşı/Trabzon, NE Turkey. Int J Coal Geol 146: 155-165.
- Hower JC, Ruppert LF, Eble CF (1999). Lanthanide, yttrium, and zirconium anomalies in the Fire Clay coal bed, Eastern Kentucky. Int J Coal Geol 39: 141-154.
- Huang H, Larter SR, Love GD (2003). Analysis of wax hydrocarbons in petroleum source rocks from the Damintun depression, eastern China, using high temperature gas chromatography. Org Geochem 34: 1673-168.
- Huang WY, Meinschein WG (1979). Sterols as ecological indicators. Geochim Cosmochim Ac 43: 739-745.
- Hunt JM (1995). Petroleum Geochemistry and Geology. 2nd ed. New York, NY, USA: Freeman and Company.
- Karayigit AI, Spears DA, Booth CA (2000a). Distribution of environmental sensitive trace elements in the Eocene Sorgun coals, Turkey. Int J Coal Geol 42: 297-314.
- Karayigit AI, Spears DA, Booth CA (2000b). Antimony and arsenic anomalies in the coal seams from the Gokler coalfield, Gediz, Turkey. Int J Coal Geol 44: 1-17.
- Katz BJ, Elrod LW (1983). Organic geochemistry of DSDP Site 467, Offshore California, Middle Miocene to Lower Pliocene. Geochim Cosmochim Ac 47: 389-396.

- Kaygusuz A, Arslan M, Siebel W, Şen C (2011). Geochemical and Sr-Nd isotopic characteristics of post-collisional calc-alkaline volcanics in the eastern Pontides (NE Turkey). Turkish J Earth Sci 20: 137-159.
- Kaygusuz A, Siebel WS, Şen C, Satır M (2008). Petrochemistry and petrology of I-type granitoids in an arc setting: the composite Torul pluton, Eastern Pontides, NE Turkey. Int J Earth Sci 97: 739-764.
- Ketris MP, Yudovich YaE (2009). Estimations of Clarkes for Carbonaceous biotithes: world averages for trace element contents in black shales and coals. Int J Coal Geol 78: 135-148.
- Koeverden JHV, Karlsen DA, Backer-Owe K (2011). Carboniferous non-marine source rocks from Spitsbergen and Bjørnøya: comparison with the western Arctic. J Petroleum Geol 34: 53-66.
- Li B, Zhuang X, Li J, Querol X, Font O, Moreno N (2016). Geological controls on mineralogy and geochemistry of the Late Permian coals in the Liulong Mine of the Liuzhi Coalfield, Guizhou Province, Southwest China. Int J Coal Geol 154-155: 1-15.
- Mann U, Korkmaz S, Boreham CJ, Hertle M, Radke M, Wilkes H (1998). Regional geology, depositional environment and maturity of organic matter of Early to Middle Jurassic coals, coly shales, shales and claystones from the eastern Pontides, NE Turkey. Int J Coal Geol 37: 257-286.
- Marzi R, Torkelson BE, Olson RK (1993). Note, A revised carbon preference index. Org Geochem 20: 1303-1306.
- Mello MR, Maxwell JR (1990). Organic geochemical and biological marker characterization of source rocks and oils from lacustrine environments in the Brazilian Continental Margin. In: Katz BJ, editor. Lacustrine Basin Exploration. Tulsa, OK, USA: AAPG Memoirs, pp. 77-97.
- Moldowan JM, Seifert WK, Gallegos E.J (1985). Relationship between petroleum composition and depositional environment of petroleum source rocks. AAPG Bull 69: 1255-1268.
- Mukhopadhyay PK, Wade JA, Kruge MA (1995). Organic facies and maturation of Jurassic, Cretaceous rocks and possible oil-source rock correlation based on pyrolysis of asphaltenes, Scotian Basin, Canada. Org Geochem 22: 85-104.
- Nishimura M (1982). 5fl-Isomers of stanols and stanones as potential markers of sedimentary organic quality and depositional paleoenvironments. Geochim Cosmochim Ac 46: 423-432.
- Okay AI, Şahintürk Ö (1997). Geology of the eastern Pontides regional and petroleum geology of the Black Sea and surrounding region. In: Robinson AG, editor. Tulsa, OK, USA: AAPG Memoirs, pp. 291-311.
- Orem WH, Finkelman RB (2003). Coal and geochemistry. In: Holland HD, Turekian KK editors. Treatise on Geochemistry. Amsterdam, the Netherlands: Elsevier, pp. 191-222.
- Palmer CA, Tuncalı E, Dennen KO, Coburn TC, Finkelman RB (2004). Characterization of Turkish coals: a nationwide perspective. Int J Coal Geol 60: 85-115.
- Peters KE (1986). Guidelines for evaluating petroleum source rock using programmed pyrolysis. AAPG Bull 70: 318-329.

- Peters KE, Moldowan JM (1993). The Biomarker Guide. Interpreting Molecular Fossils in Petroleum and Ancient Sediments. Englewood Cliffs, NJ, USA: Prentice Hall.
- Peters KE, Walters CC, Moldowan JM (2005a). The Biomarker Guide. Biomarkers and Isotopes in the Environment and Human History, Volume 1. 2nd ed. Cambridge, UK: Cambridge University Press.
- Peters KE, Walters CC, Moldowan JM (2005b). The Biomarker Guide. Biomarkers and Isotopes in Petroleum Exploration and Earth History, Volume 2. 2nd ed. Cambridge, UK: Cambridge University Press.
- Philp PR (1994). Geochemical characteristics of oils derived predominantly from terrigenous source materials. In: Scot AC, Fleet AJ, editors. Coal and Coal-Bearing Strata as Oil-Prone Source Rocks? Special Publications No. 77. London, UK: Geological Society, pp. 71-91.
- Querol X, Cabrera L, Pickel W, Lopez-Soler A, Hagemann HW, Fernandez-Turiel JL (1996). Geological controls on the coal quality of the Mequinenza subbituminous coal deposit, northeast Spain. Int J Coal Geol 29: 57-91.
- Querol X, Finkelman RB, Alastuey A, Huerta A, Palmer CA, Mroczkowski S, Kolker A, Chenery SNR, Robinson JJ, Juan R et al. (1998). Quantitative determination of modes of occurrence of major, minor and trace elements in coal: a comparison of results from different methods. In: AIE 8th Australian Coal Science Conference, pp. 51-56.
- Reimann C, Caritat P (1998). Chemical Elements in the Environment. New York, NY, USA: Springer.
- Ribeiro J, Ferreira da Silva EF, Li Z, Ward C, Flores D (2010). Petrographic, mineralogical and geochemical characterization of the Serrinha coal waste pile (Douro Coalfield, Portugal) and the potential environmental impacts on soil, sediments and surface waters. Int J Coal Geol 83: 456-466.
- Robinson KM (1987). An overview of source rocks and oils in Indonesia. In: Proceedings of the Indonesian Petroleum Association Sixteenth Annual Convention. Jakarta, Indonesia: Indonesian Petroleum Association, pp. 97-122.
- Routh J, McDonald TJ, Grossman EL (1999). Sedimentary organic matter sources and depositional environment in the Yegua Formation (Brazos County, Texas). Org Geochem 30: 1437-1453.
- Saydam Eker C (2012). Petrography and geochemistry of Eocene sandstones from eastern Pontides (NE TURKEY): implications for source area weathering, provenance and tectonic setting. Geochem Int 50: 683-701.
- Saydam Eker C (2015). Organic maturity and hydrocarbon potential of Liassic coals from the Eastern Pontides, NE Turkey. Energ Source Part A 37: 1260-1267.
- Saydam C, Korkmaz S (2008). Source rock characteristics and hydrocarbon potential of the Late Cretaceous deposits in the Eastern Pontides, NE Turkey. Energ Source Part A 30: 1141-1151.

- Saydam Eker C, Korkmaz S (2011). Mineralogy and whole rock geochemistry of late Cretaceous sandstones from the eastern Pontides (NE Turkey). Neues Jahrb Mineral 188: 235-256.
- Seifert WK, Moldowan JM (1986). Use of biological markers in petroleum exploration. Methods Geochem Geoph 24: 261-290.
- Şen C (2007). Jurassic volcanism in the Eastern Pontides: is it rift related or subduction related? Turkish J Earth Sci 16: 523-539.
- Şen C, Arslan M, Van A (1998). Geochemical and petrological characteristics of the Pontide Eocene (?) alkaline province, NE Turkey. Turkish J Earth Sci 7: 231-239.
- Seredin VV, Dai S (2012). Coal deposits as a potential alternative source for lanthanides and yttrium. Int J Coal Geol 94: 67-93.
- Shao L, Zhang P, Ren D, Lei J (1998). Late Permian coal-bearing carbonate succession in southern China; coal accumulation on carbonate platform. Int J Coal Geol 37: 235-256.
- Sipahi F (2011). Formation of skarns at Gümüşhane (Northeastern Turkey). Neues Jahrb Mineral 188: 169-190.
- Sipahi F, Sadıklar MB, Şen C (2014). The geochemical and Sr-Nd isotopic characteristics of Murgul (Artvin) volcanics in the Eastern Black Sea Region (NE Turkey). Chemie der Erde 74: 331-342.
- Spears DA, Zheng Y (1999). Geochemistry and origin of elements in some UK coals. Int J Coal Geol 38: 161-179.
- Subroto EA, Hermanto E, Kamtono P, Kamtono K (2010). Source rock geochemical study in the southwestern Java, a potential hydrocarbon basin in Indonesia. In: AAPG International Conference and Exhibition, pp. 15-18.
- Swaine DJ (1990). Trace Elements in Coal. London, UK: Butterworth.
- Taylor SR (1964). Abundance of chemical elements in the continental crust: a new table. Geochim Cosmochim Ac 28: 1273-1285.
- Taylor SR, McLennan SM (1985). The Continental Crust: Its Composition and Evolution. Oxford, UK: Blackwell Scientific Publishing.
- Temizel I, Arslan M, Ruffet G, Peucat JJ (2012). Petrochemistry, geochronology and Sr–Nd isotopic systematics of the Tertiary collisional and post-collisional volcanic rocks from Ulubey (Ordu) area, eastern Pontide, NE Turkey: implications for extension-related origin and mantle source characteristics. Lithos 128: 126-147.
- Tissot BP, Pelet R, Ungerer P (1987). Thermal history of sedimentary basins, maturation indices and kinetics of oil and gas generation. AAPG Bull 71: 1445-1466.
- Tissot BP, Welte D H (1984). Petroleum Formation and Occurrence. Berlin, Germany: Springer Verlag.
- Topuz G, Altherr R, Kalt A, Satır M, Werner O, Schwarz W (2004). Aluminous granulites from the Pulur complex, NE Turkey: a case of partial melting, efficient melt extraction and crystallization. Lithos 72: 183-207.
- Topuz G, Altherr R, Siebel W, Schwarz W, Zack T, Hasözbek A, Barth M, Satır M, Şen C (2010). Carboniferous high-potassium I-type granitoid magmatism in the Eastern Pontides: the Gümüşhane Pluton (NE Turkey). Lithos 116: 92-110.

- Valkovic V (1983). Trace Elements in Coal. Boca Raton, FL, USA: CRC Press.
- Varandas da Silva LS, Piovano EL, de Almedia Azevedo D, de Aquino Neto FR (2008). Quantitative evaluation of sedimentary organic matter from Laguna Mar Chiquita, Argentina. Org Geochem 3: 450-464.
- Wan Hasiah A, Abolins P (1998). Organic petrological and organic geochemical characterization of the Tertiary coal-bearing sequence of Batu Arang, Selangor, Malaysia. J Asian Earth Sci 16: 351-367.
- Waples DW (1985). Geochemistry in Petroleum Exploration. Berlin, Germany: Springer Verlag.
- Waples DW, Machihara T (1991). Biomarkers for Geologists-A Practical Guide to the Application of Steranes and Triterpanes in Petroleum Geology. AAPG Bulletin, Methods in Exploration Series, No: 9. Tulsa, OK, USA: AAPG.
- Wronkiewicz DJ, Condie KC (1987). Geochemistry of Archean shales from the Witwatersrand Supergroup, South Africa: sourcearea weathering and provenance. Geochim Cosmochim Ac 53: 1537-1549.

- Wronkiewicz DJ, Condie KC (1989). Geochemistry and provenance of sediments from the Pongola Supergroup, South Africa: evidence for a 3.0 Ga-old continental craton. Geochim Cosmochim Ac 53: 1537-1549.
- Wronkiewicz DJ, Condie KC (1990). Geochemistry and mineralogy of sediments from the Ventersdrop and Transvaal supergroups, South Africa: cratonic evolution during the Early Proterozoic. Geochim Cosmochim Ac 54: 343-354.
- Yılmaz Y (1972). Petrology and structure of the Gümüşhane granite and surroundings rocks, North-Eastern Anatolia. PhD, London University, London, UK.
- Yılmaz C (1997). The sedimentological records Cretaceous platformbasin transition Gümüşhane region (NE Turkey). Géologie Méditerranéenne 24: 125-135.