

Mineralogy, petrology and geochemistry of mafic dikes in the southeast Jebal-E-Barez Granitoids (Bam, Kerman province, Iran): Studies from mafic dikes formed in a volcanic arc-setting

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Abstract: In the southeastern part of the Jebal-E-Barez Oligocene granitoids (Kerman province, Iran), there are numerous mafic dikes and this work attempts to investigate the petrological, geochemical and mineral chemistry of these dikes. The dikes clearly follow faults which trend mainly N50-70W and N20-40W. Their thicknesses vary from 1 cm to 3 m and their contacts with the host granitic rocks are sharp without any extensive contact metamorphism which may be due to the lack of magmatic fluids. The Jebal-E-Barez mafic dikes contain plagioclase, clinopyroxene and amphibole as major minerals and overall show porphyritic texture, but intergranular, ophitic and subophitic textures are also visible. The amphiboles are classified mainly as paragasite and tschermakite. These are magmatic, as indicated by Si content of the studied mafic dikes. Based on Al-in-hornblende geothermobarometry, average crystallization pressures are estimated from 5.25 to 7.07 kbar and temperatures from 867 to 872 °C. Thermobarometric estimations show temperatures of 1000 ± 200 °C and depth of 30–40 km for crystallization of the clinopyroxenes. Chemical composition of the amphiboles suggests the parent magmas of the mafic dikes derived from a mantle source. Whole rock geochemistry data reveal that the studied dikes belong to the calc-alkaline magma series. They are enriched in light rare earth elements (LREE) and large ion lithophile elements (LILE) such as Cs, Rb, Ba, and Sr, with negative anomalies for high-field-strength elements (HFSE) such as Nb and Ta. The weak negative Eu anomaly is evidence of a slightly earlier separation of a mineral phase such as plagioclase. According to trace element abundances, Ba/Nb, Ce/Yb, and Zr/Y ratios, and tectono-magmatic discrimination diagrams, the primitive magma derived from an enriched garnet-free mantle source, contaminated by the crust and emplaced in continental arc environment.

Key words: Mafic dikes, mineral chemistry, petrology, calc-alkaline, geothermobarometer, continental arc, Central Iran

1. Introduction

Petrogenesis and geodynamic characteristics of mantle-sourced mafic dikes provide important information about tectono-magmatic evolution of orogenic belts (Gorring and Kay, 2001; Yang et al., 2007; Wang et al., 2009). Mafic dikes can be sourced from asthenospheric mantle related to mantle plumes, subduction and rift activities (Hoek and Seitz, 1995; Buchan et al., 2006; Xu et al., 2017). These shallow masses intrude into the continental lithosphere through extensional fractures in various extensional settings such as midocean spreading ridges (Robinson et al., 2008), back-arc, and post-collisional tectonic regimes (Taylor and Martinez, 2003; Xu et al., 2012).

Numerous studies have been performed on mafic dikes in Iran (Arvin and Robinson, 1994; Allahyari et al., 2010; Saccani et al., 2013). According to these studies, diabase dikes (Berberian and King, 1981; Bagheri and

Stampfli, 2008; Moghadam et al., 2013) and lamprophyre dikes (Torabi, 2010; Bayat and Torabi, 2011; Torabi, 2011; Pandey et al., 2017) are the most abundant types of dikes in Central Iran.

The diabase mafic dikes in this study are located in Central Iran in the southeast of the Urumieh-Dokhtar Magmatic Assemblage (UDMA), which extends to Kerman province where it is referred to as the Dehaj-Sarduieh volcano-plutonic belt. These dikes were injected into the Jebal-E-Barez granitoid (JBG) after extensive Eocene magmatism.

Due to a lack of studies about these dikes, the present work aims to determine petrographical and petrological characteristics along with mineral chemistry of these diabase dikes in JBG. For this purpose, the study was organized into four phases: 1) investigating parental magma composition, 2) specifying petrogenesis processes,

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3) defining tectonic settings and evolution of lithospheric mantle, and 4) estimating the depth of the magma chamber. Furthermore, we aim to provide new insights into Eocene magmatic processes in the region.

2. Geological setting

Convergence of the Iranian and Arabic plates in the late Mesozoic and continuation in the Early Cenozoic led to subduction of oceanic crust below Central Iran (Stöcklin, 1974; Agard et al., 2011; Castro et al., 2013, Sarjoughian and Kananian, 2017). The convergence caused extensive magmatism along the Sanandaj-Sirjan and Urumieh-Dokhtar belts in the Mesozoic and Tertiary, respectively (Agard et al., 2011).

The studied dikes and the host granitic rocks are located along the UDMA in Central Iran (Stocklin and Nabavi, 1973). Magmatic activity in this zone began with volcanic eruptions in Late Jurassic which peaked in the Eocene (Berberian and King, 1981; Berberian et al., 1982). The studied dikes crop out in the JBG and are located southwest of Bam city (E 58°00' to 58° 45' and N 28°15' to 29°00') (Figure 1) (Sohrabi et al., 2015; Dimitrijevic, 1973). The JBG, with an area of 14 km (NW–SE) × 45 km (N–S), belongs to the Oligocene period (Stocklin and Nabavi, 1973), so these dikes are younger than Oligocene (Figure 2a). The JBG generally intrudes into Eocene volcanic rocks and the main lithologies are granodiorite and granite. The formation of the JBG complex was followed by the extension of volcanic complexes during the Eocene (Razak complex) and Oligocene (Hezar complex), which contain andesite, trachyandesite, trachybasalt, andesite–basalt, and acidic tuff (Shafiei et al., 2009).

These mafic dikes follow N50–70W and N20–40W trends (Figure 1c) and are 50 cm to 3 m in width and 10 m or longer in length. Based on field observations, the dikes exhibit porphyry to fine-grained textures and dark gray to dark green colors. The dikes are only slightly altered and in some cases, pyrites are visible to the naked eye. There are very dark green duplicate dikes in this complex which injected into other dikes (Figure 2b), and have finer grains than those of the host dikes.

The boundary between the dikes and their host rocks has no reaction rims, indicating a lack of magmatic fluids (Figures 2c and 3a). Numerous joints and faults across the studied area imply high tectonic activity in the region. The faults and fractures developed along the same trends as the dikes.

3. Analysis methods

Within the scope of this study, 80 thin sections were prepared for microscopic studies. Ten fresh samples selected for whole-rock geochemical analyses were

crushed and powdered in an agate mill. Then, 0.2 g of the sample was added to 0.9 g lithium metaborate flux and mixed well before being fused in a furnace at 1000 °C. The resulting melt was then cooled and dissolved in 100 mL of 4% HNO₃ and 2% HCl solution and major and trace elements were analyzed by inductively coupled plasma-mass spectrometry (ICP-MS) and inductively coupled plasma atomic emission spectroscopy (ICP-AES) at the ALS Chemex laboratory in Canada. The analysis results are presented in Table 1.

Electron microprobe analysis of minerals was performed using an automated JEOL JXA-8900 microprobe (Yamagata University) at an accelerating voltage of 15 kV, a beam current of 20 nA, a beam diameter of about 5 μm, detection limit of 0.05 wt.%, and a maximum counting interval of 40 s. The diameter of the focused electron beam was about 5 μm. Data were processed by an online computer equipped with XM-86 PAC Software (JEOL) with the oxide ZAF method implemented. Calibration standards for the mentioned minerals were apatite, wollastonite, albite, adularia, synthetic SiO₂, TiO₂, Al₂O₃, Fe₂O₃, MnO, MgO, CaF₂, and NaCl. For each sample, several grains and several points on each mineral were analyzed based on textural relationships, and average values of the analytical results were taken as representative of typical composition of that mineral in each sample. The results of analyses for each mineral are shown in individual tables. Formula calculations for feldspar, amphibole and pyroxene are based on 8, 23, and 6 atoms of oxygen, respectively.

4. Results

4.1. Petrography

Host granitic rock units are composed of granodiorite and granite. Plagioclase (35%), quartz (25%), K-feldspar (20%), amphibole (10%), and biotite (7%) are the main minerals in granodiorite. Apatite, zircon and opaque minerals are accessory minerals. Granitic rocks, on the other hand, are mainly composed of K-feldspar (30%), plagioclase (25%), quartz (25%), biotite (10%), and amphibole (8%), with apatite, zircon, titanite, and opaque minerals as accessory mineral constituents. The granitoids exhibit a subhedral granular texture.

According to the results of petrographic studies, diabase dikes exhibited no obvious evidence of deformation and metamorphism. The samples show an aphanitic matrix and diabase textures and they exhibit ophitic and subophitic texture (Figure 3b) (Kretz, 1983). In terms of composition, the diabase dikes are made up of plagioclase (50%), clinopyroxene (25%), K-feldspar (10%), and anhedral quartz (less than 5%). Some samples contained amphibole (10%) (Figure 3c). The plagioclase is found both as phenocrysts and microlites. The phenocrysts of

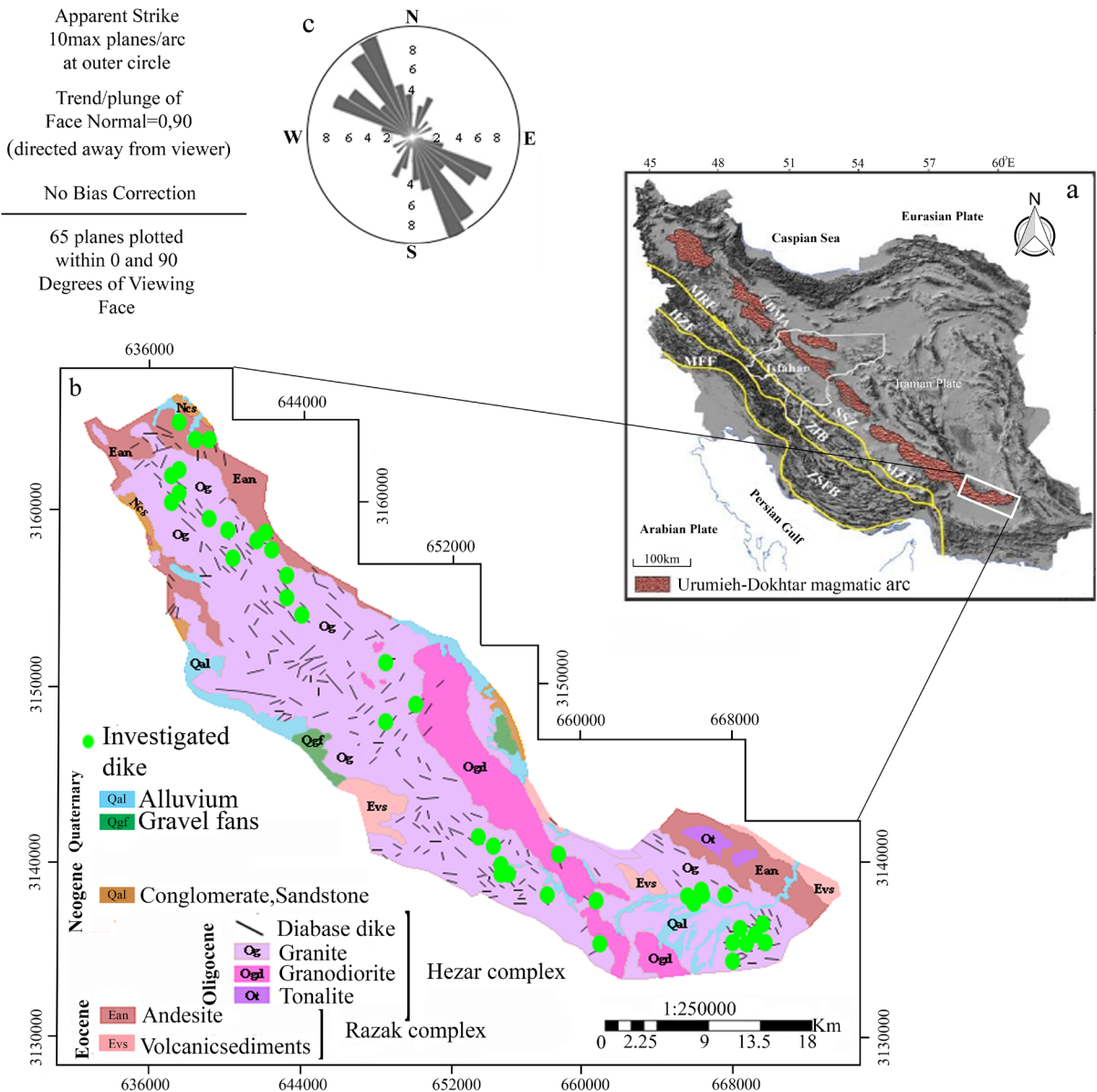


Figure 1. Map of the studied area, showing (a) Main tectonic units of Iran. Tectonic zones of the Zagros Orogen: ZSFB - Zagros Simply Folded Belt, ZIB - Zagros Imbricate Belt (High Zagros Belt), SSZ Sanandaj-Sirjan Zone, UDMA - Urumieh-Dokhtar Magmatic Arc, MFF Mountain Frontal Fault, MRF Main Recent Fault, HZF High Zagros Fault. Major faults: MZT - Main Zagros Thrust (Sohrabi et al., 2015 and references therein), (b) Simplified geological map of the studied area (Dimitrijevic, 1973), and (c) Rose diagrams of the dikes.

plagioclase are 1–4 mm in size (Figure 3d). The subhedral clinopyroxene phenocrysts range between 2 and 4 mm in size. These phenocrysts are augitic in composition and are located in a fine-grained matrix made of plagioclase laths, small clinopyroxene grains, and opaque minerals. Accessory minerals are apatite, zircon and opaque minerals. The suite of secondary minerals includes epidote, sericite, calcite, zeolite and chlorite. Microscopic studies show that

the dominant texture of rocks in this area is porphyry, and intergranular in some cases.

4.2. Mineral chemistry

The major element composition of clinopyroxenes, plagioclase, and amphibole are presented in Tables 2, 3, and 4, respectively.

Clinopyroxenes exhibit augitic composition ($Wo_{27.71-28.85} En_{42.70-48.13} Fs_{23.28-29}$) (Figure 4a). They plot in the Ca-Na

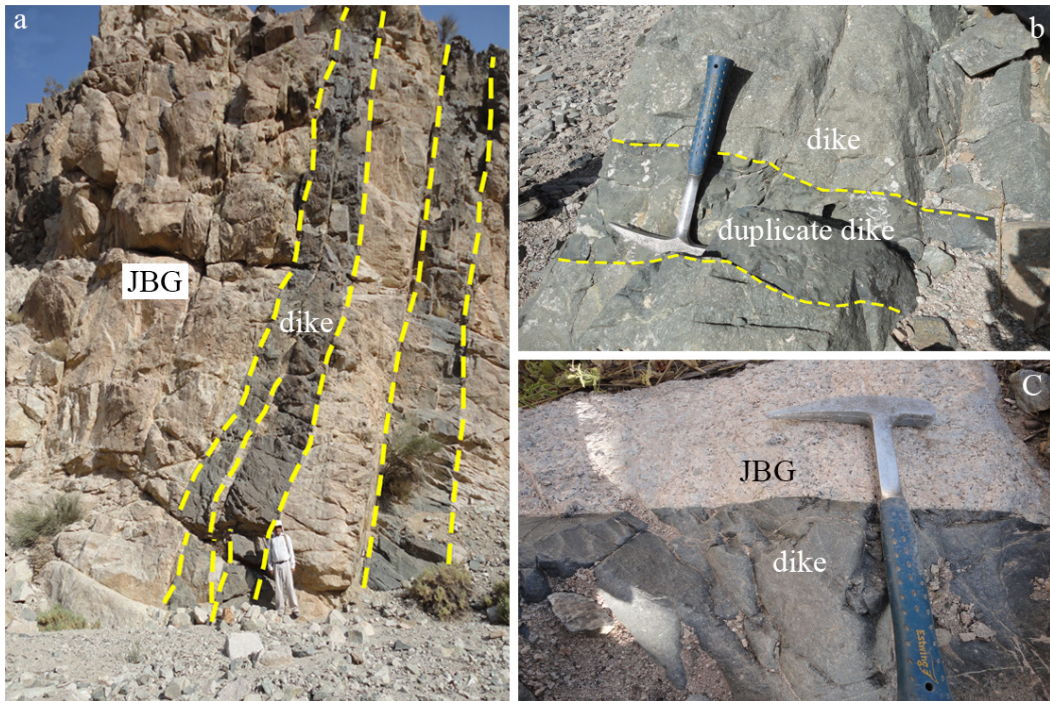


Figure 2. a) The studied dikes outcropping in the Jebal-E-Barez granitoid (JBG). b) Duplicate dikes injecting into other dikes, are finer grained than those of host dikes. c) The boundary between the dikes and their host rocks has no reaction margin.

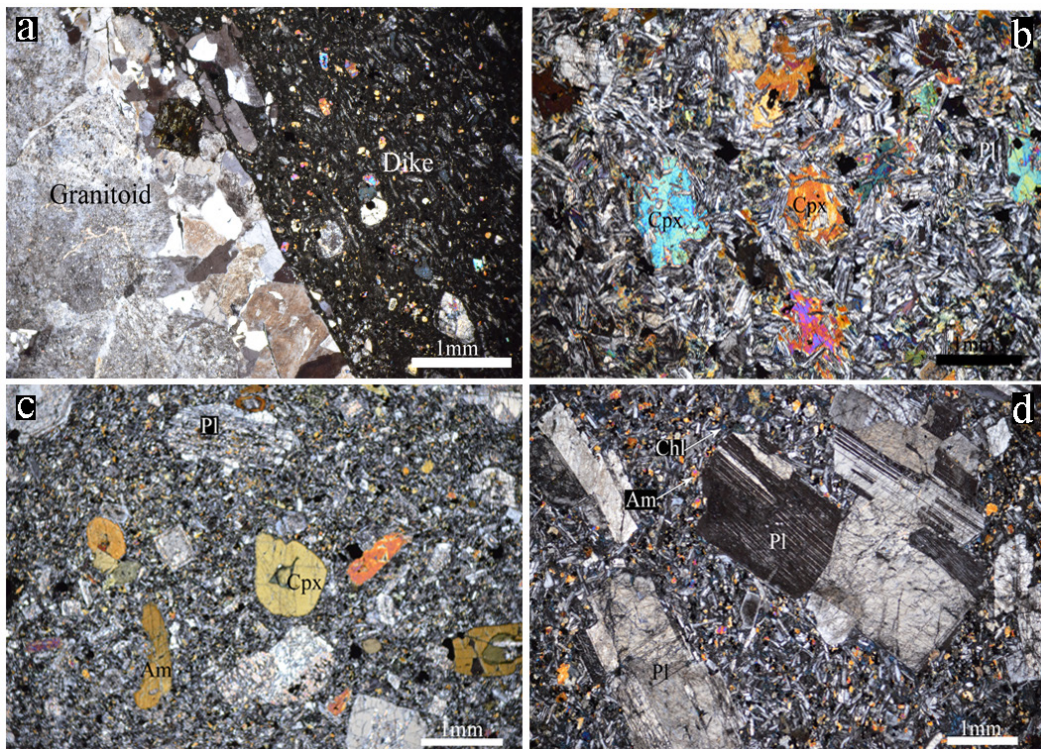


Figure 3. Representative photomicrographs of petrographic features of the JBG mafic dikes, cross-polarized light: a) the contact between the dikes and host rock, b) ophitic and sub-ophitic texture in dikes, c) primary amphibole in dikes, d) phenocrysts and microlites of plagioclase in dike. Clinopyroxene (Cpx), amphibole (Am), plagioclase (Pl), chlorite (Chl) (Kretz, 1983).

Table 1. Chemical composition of the JBG mafic dykes.

Samples	1jb8	2jb20	2jb33	2jb35	3jb14	3jb9	5jb16	5jb29	6jb15	9jb3
SiO ₂ (wt.%)	46.80	48.00	48.50	46.30	48.90	50.00	49.00	47.80	48.00	50.20
TiO ₂	0.90	0.99	0.96	1.01	0.89	0.76	0.78	0.82	0.81	0.70
Al ₂ O ₃	17.40	18.45	18.45	16.65	17.60	18.95	18.65	17.35	19.25	17.25
Fe ₂ O ₃ ^T	10.65	11.65	11.75	10.50	11.10	9.61	11.25	9.67	10.05	9.92
MnO	0.23	0.18	0.23	0.20	0.20	0.19	0.19	0.17	0.18	0.23
MgO	6.19	4.82	4.68	8.34	5.46	3.78	4.25	6.79	5.40	6.36
CaO	9.72	8.27	8.43	10.65	9.30	9.55	8.02	8.85	10.10	9.42
Na ₂ O	2.48	2.99	2.38	1.76	2.39	2.40	2.83	2.65	2.30	2.26
K ₂ O	0.95	0.97	0.86	0.95	0.60	0.80	0.80	1.04	0.72	0.65
P ₂ O ₅	0.28	0.16	0.13	0.34	0.17	0.13	0.15	0.17	0.14	0.12
Cr ₂ O ₃	0.02	<0.01	<0.01	0.05	<0.01	<0.01	<0.01	0.02	0.01	0.01
Total	98.27	98.87	98.82	99.92	99.84	100.55	98.46	98.40	99.19	99.31
Mg#	36.76	29.27	28.48	44.27	32.97	28.23	27.42	41.25	34.95	39.07
Ba (ppm)	199	233	251	170.5	149	261	171.5	227	161.5	191.5
Ce	30.7	16.1	16.5	42.4	25.1	22.1	19.5	26.8	19.9	17.1
Cr	120	20	10	360	20	10	<10	180	60	50
Cs	0.84	0.85	1.04	1.2	2.29	2.15	2.11	1.05	0.7	0.92
Cu	74	118	96	55	86	93	118	77	78	41
Dy	3.49	3.31	3.33	3.78	3.98	3.38	3.53	3.29	3.03	2.82
Er	2.1	2.04	2.04	1.99	2.46	2.05	2.19	1.98	1.83	1.75
Eu	1.11	0.96	0.88	1.36	1.04	0.89	1.01	0.97	0.87	0.74
Ga	19.2	19.7	14.8	16	17.4	16.6	16.6	16	16.9	15.3
Gd	3.68	3.47	3.16	4.42	4.12	3.47	3.6	3.32	3.28	2.73
Hf	2.5	1.8	1.8	3	2.9	2.6	2.1	2.1	1.9	2
Ho	0.75	0.7	0.7	0.77	0.85	0.71	0.77	0.72	0.63	0.62
La	13.9	6.8	7.2	20.2	11	10.3	8.5	12.5	9	7.7
Lu	0.28	0.28	0.3	0.27	0.35	0.33	0.39	0.29	0.27	0.26
Nb	6.2	2.5	2.9	10.8	4.5	3.7	3.2	4.1	2.9	3.2
Nd	16.4	11	10.1	20.8	14.1	12.2	11.5	13.6	11.4	8.9
Ni	45	<5	<5	123	16	5	<5	73	34	28
Pr	3.98	2.3	2.21	5.2	3.29	2.78	2.65	3.25	2.59	2.3
Rb	34.7	29.5	33	29.8	25.3	31.8	32.2	28	16.4	20.6
Sm	3.86	3.03	2.7	4.45	3.67	3.21	3.27	3.27	2.91	2.48
Sr	555	543	477	710	441	437	488	665	474	416
Ta	0.3	0.1	0.2	0.6	0.2	0.2	0.2	0.2	0.1	0.2
Tb	0.58	0.6	0.54	0.67	0.68	0.57	0.6	0.54	0.47	0.45
Th	1.19	1.07	0.97	2.06	1.95	2.19	1.36	2.42	1.59	1.93
Tm	0.31	0.28	0.31	0.3	0.38	0.34	0.36	0.28	0.3	0.26
U	0.37	0.32	0.43	0.74	0.56	0.69	0.47	0.6	0.43	0.52
V	220	298	278	203	271	239	257	272	292	253
Y	19.2	18.6	18.5	19.7	22.4	19.2	20.4	18.1	16.7	17.1
Yb	2.25	2.1	1.99	1.93	2.38	1.88	2.1	1.78	1.68	1.77
Zr	108	62	61	113	104	90	74	77	63	80
Eu/Eu*	0.39	0.37	0.42	0.49	0.39	0.37	0.41	0.40	0.37	0.36
(La/Sm) _n	2.32	1.45	1.72	2.93	1.93	2.07	1.68	2.47	2.00	2.00
(La/Yb) _n	4.43	2.32	2.60	7.51	3.32	3.93	2.90	5.04	3.84	3.12

Table 2. Representative chemical composition and calculated mineral formulae of clinopyroxenes from the JBG mafic dikes. Formulae calculated on the basis of 6 oxygens.

Sample	b1	b2	b3	b4	b5	b6	d1	d2	d3	d4	d5	d6
Oxides (wt%)												
SiO ₂	53.41	53.57	53.65	52.92	53.20	54.48	53.96	53.16	53.80	52.49	53.80	52.57
TiO ₂	0.18	0.68	0.79	0.75	0.43	0.27	0.23	0.53	0.41	0.28	0.28	0.00
Al ₂ O ₃	3.77	3.62	3.97	4.42	4.38	4.05	3.58	5.28	4.41	4.89	4.81	4.84
FeO	15.71	12.30	13.50	14.41	14.29	13.40	15.37	14.54	15.16	14.73	14.23	15.55
MnO	0.60	0.62	0.42	0.63	0.50	0.50	0.59	0.40	0.57	0.59	0.49	0.15
MgO	13.47	14.99	14.37	13.61	13.83	14.69	13.76	13.72	13.48	13.64	14.04	13.59
CaO	12.42	12.39	12.39	11.96	12.53	12.50	12.24	11.99	12.23	12.50	11.90	12.64
Na ₂ O	0.26	0.16	0.29	0.27	0.26	0.18	0.19	0.27	0.28	0.28	0.22	0.25
K ₂ O	0.15	0.09	0.09	0.12	0.17	0.15	0.08	0.13	0.13	0.14	0.08	0.17
Cr ₂ O ₃	0.03	0.09	0.05	0.32	0.02	0.16	0.39	0.16	0.41	0.11	0.23	0.11
Total	99.99	98.51	99.52	99.40	99.62	100.37	100.39	100.16	100.88	99.64	100.07	99.86
Cations (apfu)												
Si	1.99	1.99	1.98	1.97	1.98	1.99	2.00	1.96	1.98	1.96	1.98	1.96
Al _(IV)	0.01	0.01	0.02	0.03	0.02	0.01	0.00	0.04	0.02	0.04	0.02	0.04
Al _(VI)	0.15	0.15	0.16	0.16	0.17	0.17	0.15	0.19	0.17	0.17	0.19	0.17
Al ^T	0.17	0.16	0.17	0.19	0.19	0.17	0.16	0.23	0.19	0.21	0.21	0.21
Fe	0.49	0.38	0.42	0.45	0.44	0.41	0.48	0.45	0.47	0.46	0.44	0.48
Ti	0.01	0.02	0.02	0.02	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.00
Cr	0.00	0.00	0.00	0.01	0.00	0.00	0.01	0.00	0.01	0.00	0.01	0.00
Mg	0.75	0.83	0.79	0.76	0.77	0.80	0.76	0.75	0.74	0.76	0.77	0.75
Fe ²	0.40	0.38	0.39	0.40	0.39	0.39	0.40	0.41	0.40	0.40	0.41	0.41
Mn	0.02	0.02	0.01	0.02	0.02	0.02	0.02	0.01	0.02	0.02	0.02	0.00
Ca	0.50	0.49	0.49	0.48	0.50	0.49	0.48	0.47	0.48	0.50	0.47	0.50
Na	0.02	0.01	0.02	0.02	0.02	0.01	0.01	0.02	0.02	0.02	0.02	0.02
K	0.01	0.00	0.00	0.01	0.01	0.01	0.00	0.01	0.01	0.01	0.00	0.01
En	42.71	48.13	46.21	44.42	44.42	46.67	43.67	44.66	43.34	43.69	45.51	43.17
Fs	29.01	23.28	25.13	27.53	26.66	24.79	28.42	27.29	28.40	27.54	26.77	27.98
Wo	28.28	28.58	28.65	28.05	28.92	28.55	27.91	28.05	28.26	28.78	27.72	28.86

pyroxene area on the Q-J diagram (Figure 4b) (Morimoto, 1988). Chemical composition and the environment of dikes affected the clinopyroxene composition. According to Al₂O₃ vs. SiO₂, the magmatic series is subalkaline (Figure 4c) (Le Bas, 1962).

The phenocryst composition of the plagioclase changes from andesine to bytownite (An₄₂₋₈₇) (Figure 5a). Normal and oscillatory zoning is relatively common in

the phenocrysts (Figure 5b). H₂O pressure (NBr and Son, 1992) and thermodynamic changes (Bottinga et al., 1966) in the magmatic system affects the composition and zoning of plagioclase during magmatic crystallization.

In the study area, amphiboles are found in both primary and secondary forms. Primary amphiboles are calcic and plot in the paragonite, tschermakite, and tschermakite field (Hawthorne et al., 2012); on the other hand, secondary

Table 3. Representative chemical composition and calculated mineral formulae of plagioclases from the JBG mafic dikes. Formulae calculated on the basis of 8 oxygens.

Samples	2jb5-1	2jb5-2	2jb5-3	2jb4.1	2jb4.2	5jb-1	5jb-2	5jb3	5jb4	jb1	jb2	jb3	jb4	2jb6	2jb3	2jb35
Oxides (wt%)																
SiO ₂	47.73	48.29	48.57	49.21	51.40	49.72	49.42	49.00	50.01	48.19	48.23	48.51	48.26	64.08	55.34	60.69
TiO ₂	0.00	0.00	0.04	0.07	0.01	0.00	0.00	0.01	0.00	0.00	0.00	0.00	0.08	0.01	0.04	0.01
Al ₂ O ₃	32.39	33.69	32.95	32.34	30.13	33.79	33.20	34.42	32.36	33.35	33.47	33.71	33.14	22.97	27.97	25.96
FeO	0.55	0.51	0.52	0.59	0.49	0.47	0.43	0.36	0.80	0.50	0.45	0.51	0.55	0.04	0.62	0.22
MnO	0.07	0.09	0.00	0.01	0.05	0.01	0.09	0.00	0.09	0.00	0.00	0.06	0.00	0.03	0.00	0.02
MgO	0.05	0.09	0.08	0.06	0.10	0.32	0.14	0.04	0.54	0.05	0.05	0.08	0.06	0.00	0.04	0.00
CaO	17.59	17.08	16.29	15.62	15.43	14.24	14.05	15.49	13.33	17.47	17.18	16.97	16.30	3.73	10.88	7.12
Na ₂ O	1.55	1.52	1.76	2.14	2.85	1.50	1.57	1.61	2.47	1.34	1.34	1.39	1.30	9.91	5.46	5.33
K ₂ O	0.11	0.11	0.13	0.15	0.22	0.77	0.48	0.05	1.07	0.11	0.09	0.09	0.09	0.20	0.52	0.18
Total	100.04	101.38	100.34	100.18	100.68	100.82	99.38	100.97	100.66	101.02	100.81	101.31	99.78	100.96	100.87	99.52
Cations (apfu)																
Si	2.19	2.18	2.21	2.25	2.33	2.24	2.26	2.21	2.27	2.19	2.19	2.19	2.21	1.91	1.91	2.68
Al	1.75	1.79	1.77	1.74	1.61	1.80	1.79	1.83	1.73	1.79	1.79	1.80	1.79	1.76	1.77	1.35
Ti	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Fe	0.02	0.02	0.02	0.02	0.02	0.02	0.02	0.01	0.03	0.02	0.02	0.02	0.02	0.04	0.04	0.01
Mn	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00
Mg	0.00	0.01	0.01	0.00	0.01	0.02	0.01	0.00	0.04	0.00	0.00	0.01	0.00	0.00	0.00	0.00
Ca	0.86	0.83	0.79	0.76	0.75	0.69	0.69	0.75	0.65	0.85	0.84	0.82	0.80	1.38	1.36	0.34
Na	0.14	0.13	0.15	0.19	0.25	0.13	0.14	0.14	0.22	0.12	0.12	0.12	0.12	0.21	0.21	0.46
K	0.01	0.01	0.01	0.01	0.01	0.04	0.03	0.00	0.06	0.01	0.01	0.00	0.01	0.02	0.01	0.01
Or	0.64	0.63	0.80	0.89	1.26	5.15	3.26	0.35	6.68	0.64	0.54	0.52	0.55	1.06	0.89	1.24
Ab	13.62	13.81	16.19	19.69	24.75	15.20	16.23	15.75	23.42	12.12	12.30	12.85	12.50	13.17	13.37	56.81
An	85.73	85.56	83.01	79.42	73.99	79.65	80.51	83.90	69.90	87.24	87.16	86.63	86.95	85.77	85.74	41.94

amphiboles are tremolite (Figure 5c). The Si content of magmatic amphibole is less than 7.4 apfu (Agemar et al., 1999). In this study, the amphiboles are magmatic, as indicated by the Si content of the studied samples. Ti, K, Na, and Al contents of calc-alkaline amphiboles are lower than those of alkaline amphiboles. Accordingly, the studied amphiboles are calc-alkaline. In this study, primary amphiboles are used to calculate temperature and pressure of amphibole crystallization.

4.3. Whole-rock geochemistry

Geochemical data of the 10 samples from the studied diabase dikes with minimum weathering are presented in Table 1.

The samples are positioned in the field of subalkaline basalt on the Zr/TiO₂ vs. Nb/Y diagram (Figure 6a) (Winchester and Floyd, 1977). On the K₂O vs. SiO₂

diagram (Le Maitre et al., 1989), samples plot within the calc-alkaline field (Figure 6b). The samples have high Al₂O₃ (16.65- 19.25) and moderately low Mg# (27.41-44.26), along with largely variable amounts of K₂O (0.6-1.04).

The pattern of primitive mantle-normalized incompatible trace elements is shown in Figure 6c (McDonough and Sun, 1995). The pattern shows enrichment in large ion lithophile elements (LILE; e.g., Sr, K, Rb Ba) and light rare earth elements (LREE; e.g., Ce) and depletion in high field strength elements (HFSE; e.g., Ta, Nb, Ti, Zr, Hf, and Y) and heavy rare earth elements (HREE; e.g., Yb). Rare earth elements (REEs), which are normalized to chondrite (Boynnton, 1984), show significant enrichment in LREE (Figure 6d). REEs with generally sloping patterns show an enrichment in all REE, with

Table 4. Representative chemical composition and calculated mineral formulae of amphiboles from the JBG mafic dykes. Formulae calculated on the basis of 23 oxygens, $Mg\# = Mg / (Fe^{2+} + Mg)$.

Samples	jb16	5jb16	pjb16	3p5jb16	4p5jb16	S2jb35	S3jb35	S2jb35	S32jb35	S4b35	
Oxides (wt%)											
SiO ₂	43.12	42.61	43.61	43.52	43.22	40.88	41.33	41.36	41.50	40.96	
TiO ₂	1.87	1.95	2.06	2.19	2.01	3.62	3.44	2.50	3.27	2.83	
Al ₂ O ₃	13.09	12.99	11.15	11.78	12.53	13.11	13.10	12.60	12.23	12.01	
FeO	12.54	12.44	12.67	12.16	11.61	12.03	12.80	14.52	12.32	15.90	
MnO	0.24	0.22	0.17	0.14	0.18	0.09	0.17	0.24	0.26	0.22	
MgO	13.40	13.55	14.51	12.73	13.62	13.20	12.75	11.63	12.37	10.94	
CaO	11.40	11.63	12.31	11.82	11.84	11.67	11.46	11.64	11.72	12.15	
Na ₂ O	2.20	2.34	2.58	2.66	2.79	2.41	2.39	2.38	2.24	2.37	
K ₂ O	0.45	0.43	0.48	0.22	0.50	1.13	1.22	1.25	1.22	1.34	
Total	98.31	98.17	99.54	97.22	98.30	98.14	98.65	98.11	97.12	98.73	
(apfu) Cations											
Si	6.19	6.14	6.25	6.41	6.27	5.98	6.02	6.13	6.18	6.12	
Ti	0.20	0.21	0.22	0.24	0.22	0.40	0.38	0.28	0.37	0.32	
Al	2.21	2.21	1.88	2.04	2.14	2.26	2.25	2.20	2.15	2.11	
Al ^{IV}	1.81	1.86	1.75	1.59	1.73	2.02	1.98	1.87	1.82	1.88	
Al ^{VI}	0.40	0.35	0.14	0.45	0.41	0.24	0.27	0.32	0.32	0.23	
Fe	1.50	1.50	1.52	1.50	1.41	1.47	1.56	1.80	1.53	1.99	
Fe ³⁺	0.81	0.76	0.58	0.13	0.33	0.43	0.47	0.38	0.15	0.18	
Fe ²⁺	0.69	0.74	0.94	1.36	1.07	1.04	1.09	1.42	1.38	1.80	
Mn	0.03	0.03	0.02	0.02	0.02	0.01	0.02	0.03	0.03	0.03	
Mg	2.86	2.91	3.10	2.79	2.94	2.88	2.77	2.57	2.74	2.44	
Ca	1.75	1.80	1.89	1.86	1.84	1.83	1.79	1.85	1.87	1.94	
Na	0.61	0.65	0.72	0.76	0.78	0.68	0.68	0.68	0.65	0.69	
K	0.08	0.08	0.09	0.04	0.09	0.21	0.23	0.24	0.23	0.25	
B: Ca	1.75	1.80	1.82	1.74	1.76	1.83	1.76	1.85	1.81	1.89	
B: Na	0.25	0.20	0.18	0.26	0.24	0.17	0.24	0.15	0.19	0.11	
Mg#	0.66	0.66	0.67	0.65	0.68	0.66	0.64	0.59	0.64	0.55	
Oxides (wt%)											
SiO ₂	41.89	42	42.39	41.48	41.9	41.55	42.09	44.43	46.04	51.55	45.72
TiO ₂	3.72	3.05	2.45	3.67	3.78	3.67	3.68	2	1.75	0	1.65
Al ₂ O ₃	11.75	12.48	13.23	13.95	11.49	12.67	12.56	10.71	9.73	4.42	10.29
FeO	12.05	12.64	12.03	11.6	12.07	11.98	11.71	14.28	13.85	12.66	16.51
MnO	0.14	0.2	0.18	0.12	0.16	0.12	0.1	0.39	0.53	0.59	0.42
MgO	13.21	12.5	13.54	11.81	13.96	13.61	13.02	12.47	13.07	14.73	12.56
CaO	12.01	12.03	11.53	12.15	11.16	11.44	11.81	11.71	10.29	12.67	10.22
Na ₂ O	2.49	2.27	2.32	2.06	2.33	2.39	2.35	1.72	2.3	0.33	2.14
K ₂ O	1.26	1.23	1.09	1.19	1.2	1.27	1.29	1.06	0.26	0.14	0.37
Total	98.52	98.4	98.75	98.04	98.05	98.68	98.61	98.77	97.83	97.09	99.87

Table 4. Continued.

(apfu) Cations											
Si	6.16	6.18	6.11	6.12	6.1	6.03	6.15	6.46	6.63	7.46	6.46
Ti	0.41	0.34	0.27	0.41	0.41	0.4	0.4	0.22	0.19	0	0.18
Al	2.04	2.16	2.25	2.43	1.97	2.17	2.16	1.84	1.65	0.75	1.71
Al ^{IV}	1.84	1.82	1.89	1.88	1.9	1.97	1.85	0	1.37	0.54	1.54
Al ^{VI}	0.19	0.34	0.36	0.54	0.07	0.19	0.32	4.82	0.28	0.21	0.18
Fe	1.48	1.55	1.45	1.43	1.47	1.45	1.43	1.74	1.67	1.53	1.95
Fe ³⁺	0.09	0.13	0.6	0	0.64	0.52	0.12	0.48	0.86	0.28	1.26
Fe ²⁺	1.39	1.42	0.85	1.43	0.83	0.93	1.31	1.26	0.81	1.26	0.69
Mn	0.02	0.02	0.02	0.02	0.02	0.01	0.01	0.05	0.06	0.07	0.05
Mg	2.89	2.74	2.91	2.6	3.03	2.94	2.84	2.7	2.8	3.18	2.65
Ca	1.89	1.9	1.78	1.92	1.74	1.78	1.85	1.82	1.59	1.96	1.55
Na	0.71	0.65	0.65	0.59	0.66	0.67	0.67	0.48	0.64	0.09	0.59
K	0.24	0.23	0.2	0.22	0.22	0.23	0.24	0.2	0.05	0.03	0.07
B: Ca	1.81	1.84	1.77	1.86	1.85	1.75	1.8	1.79	1.53	1.92	1.55
B: Na	0.19	0.16	0.23	0.14	0.15	0.25	0.2	0.21	0.47	0.08	0.45
Mg#	0.66	0.64	0.67	0.64	0.67	0.67	0.66	0.61	0.63	0.67	0.58

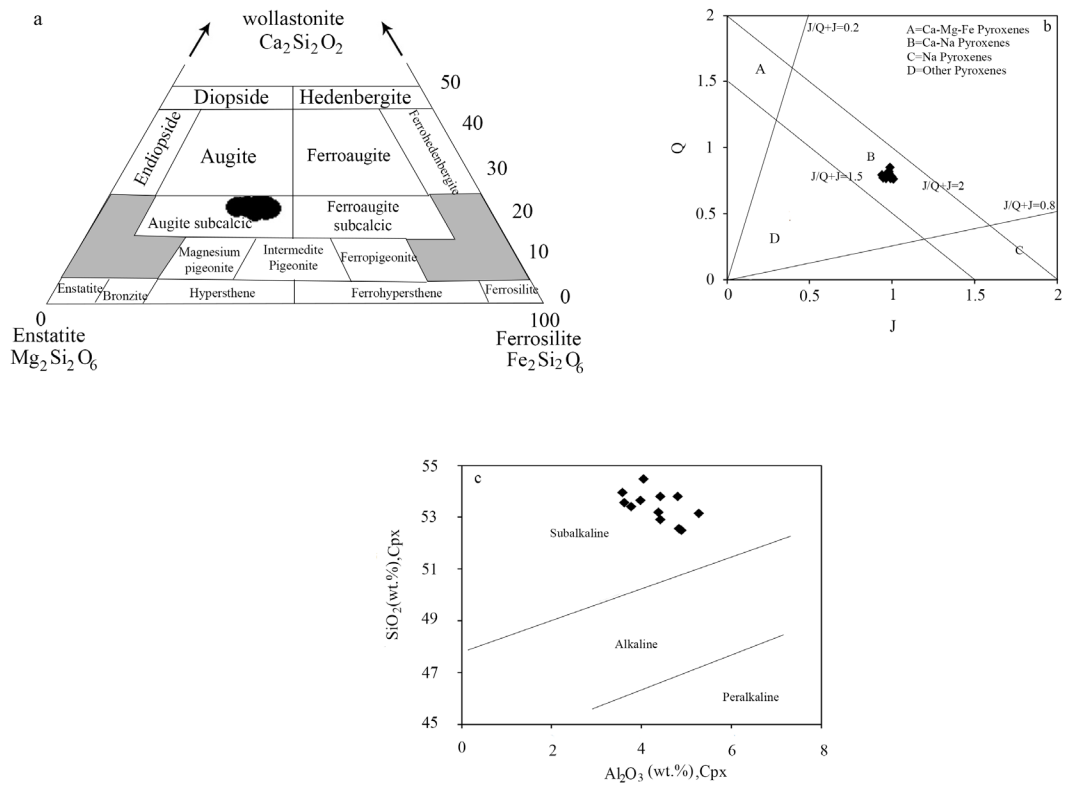


Figure 4. Classification and discriminant diagrams for clinopyroxene (after Morimoto, 1988) of the studied mafic dikes: (a) En–Wo–Fs (Mg₂Si₂O₆–Ca₂Si₂O₆–Fe₂Si₂O₆) plot, (b) Q–J (Q=Ca+Mg+Fe²⁺, J=2Na) plot and (c) SiO₂–Al₂O₃ plot.

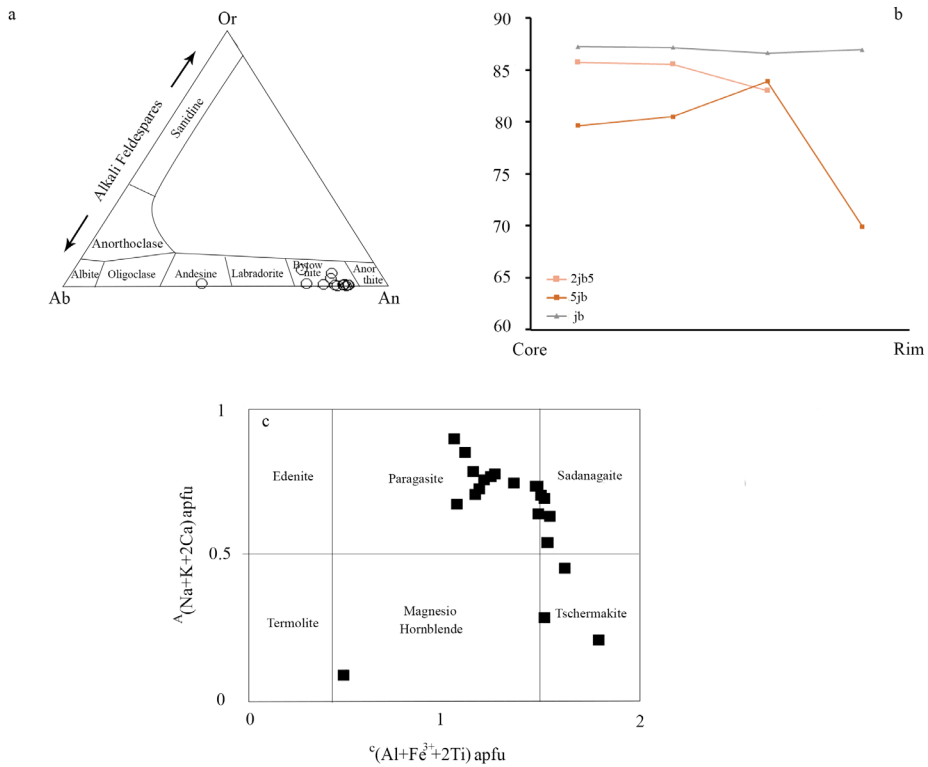


Figure 5. (a) Classification ternary diagram for feldspar (Ab–An–Or plot; Deer et al., 1992), (b) Variation diagram for plagioclase and (c) Classification diagram (Hawthorne et al., 2012) for amphibole of the studied mafic dikes.

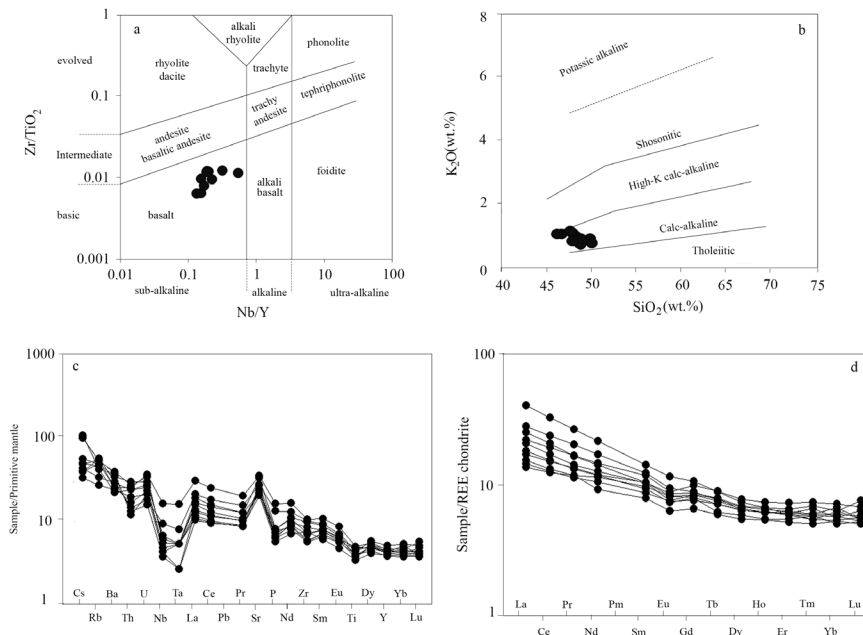


Figure 6. (a) Zr/TiO₂–Nb/Y classification diagram (Winchester and Floyd, 1977), (b) K₂O–SiO₂ classification diagram (Le Maitre et al., 1989), (c) Primitive mantle-normalized multielement patterns (McDonough and Sun, 1995), and (d) REE-normalized patterns (Boynton, 1984) for the JBG mafic dikes.

higher enrichment of LREE than HREE ($\text{La/Yb} = 3.2\text{--}10$). Eu shows weak negative anomaly ($\text{Eu/Eu}^* = 0.81\text{--}0.92$).

5. Discussion

5.1. Tectonic setting

The studied dikes and the host granitic rocks are a part of the UDMA. The UDMA has a complex geological history involving the formation of voluminous volcanic rocks from Eocene to Quaternary (Agard et al., 2011; Shahabpour, 2005), intrusion of igneous bodies into previous rocks from late Eocene to the Miocene (Dimitrijevic, 1973; Shafiei et al., 2009), and also collision and postcollision related events (Allen et al., 2013; Ghasemi and Talbot, 2006). Many authors think that this magmatic arc developed along an active continental margin (e.g., Berberian et al., 1982; Ghorbani et al., 2014; Rasouli et al., 2014) and formed due to subduction of Neotethyan lithosphere beneath the Central Iran zone (Aftabi and Atapour, 2000; Berberian and Berberian, 1981).

In order to identify the tectonic setting of the studied rocks, major and trace elements should be considered. Based on the Nb/Th-Y diagram (Swinden et al., 1989), all the samples plot within the arc field (Figure 7a). High Sr contents of the samples confirm their arc character (Figure 6c). On the Zr/Y vs. Zr diagram (Pearce, 1982), which distinguishes between continental arc and oceanic arc environments, all of the samples plot in the continental arc area (Figure 7b). On the Th vs. Hf vs. Nb discrimination diagram (Wood, 1980), all the samples plot in the supra-subduction zone (SSZ) field (Figure 7c). The Zr/Y rate affinities provide additional information about the tectonic environment. Accordingly, if $\text{Zr/Y} > 3$, the volcanic rocks are likely to form in a continental volcanic arc, and $\text{Zr/Y} < 3$ infers that the volcanic rocks formed in an oceanic volcanic arc (Pearce and Norry, 1979). With Zr/Y rate of the samples being higher than 3 (3.25–5.73), the samples are described as from a continental volcanic arc. According to $\text{Ba/Nb} > 32$, the studied samples exhibit characteristics of arc magma (Fitton et al., 1988). With $\text{Ba/La} > 15$ and $\text{La/Th} < 7$, the samples resemble active continental margin rocks (Gill, 1981).

The Na vs. Cr diagram (Kornprobst et al., 1981) shows the clinopyroxenes from the diabase dikes formed in a continental environment (Figure 7d). In Nisbet and Pearce's (1977) diagram, the studied clinopyroxenes plotted in the volcanic arc field (Figure 7e). Using the SiO_2 vs. Na_2O diagram (Coltorti et al., 2007) for amphiboles (Figure 7f), it can be deduced that a subduction zone is the origin of amphiboles in the diabase dikes. In order to separate different tectonic environments, a threshold of $\text{Al}^{\text{IV}} = 1.5$ was considered. Accordingly, the Al^{IV} value of high-pressure amphiboles (~10 kbar) formed in magmatic

arcs is higher than 1.5 (Jiang and An, 1984). Based on the Al^{IV} value, the studied amphiboles ($\text{Al}^{\text{IV}} > 1.5$) formed in a magmatic arc zone. Thus, mineral chemistry results are in agreement with those of whole-rock geochemistry studies inferring a volcanic arc setting for the studied rocks.

5.2. Fractional crystallization

Crystallization of mafic minerals from primitive melts at high pressure results in high concentration of Al_2O_3 in the basalts (Gust and Perfit, 1987). The Al_2O_3 is concentrated in the remaining melt and plagioclase crystallizes at lower pressure. The content of Al_2O_3 in the intermediate orogeny rocks was estimated as 16–18 wt.% (Gill, 1981). According to Al_2O_3 contents of the studied rocks, the samples are found to be metaluminous and orogenic rock type. The lack of negative correlation between P_2O_5 , TiO_2 and MgO can be explained by fractionation of accessory minerals such as apatite and Fe-Ti oxide (Kampunzu and Lubala, 2012) (Table 1). Low MgO (<6 wt.%), Mg# (27.41–44.26 wt.%), Cr (10–360 ppm) and Ni (5–123 ppm) indicate that the magma in the dikes is not primary (Wilson, 1989), i.e. the studied dikes originated from a primary magma subsequently affected by fractional crystallization. This understanding is further supported by the presence of pyroxene and plagioclase and positive correlations between MgO, Al_2O_3 , CaO, K_2O , Sr, and Cr. High Sr contents, along with the lack of Eu anomalies, suggest absence of plagioclase crystal fractionation (Figure 6d).

5.3. Geothermobarometry estimations

Crystallization temperature and pressure of pyroxene and amphibole can be estimated from their chemical composition, helping to understand the physical conditions of crystallization.

Aluminum distribution in tetrahedral and octahedral sites of clinopyroxene can be used as a standard for estimating environmental pressure of pyroxene formation in igneous rocks (Helz, 1973). Given that Al^{VI} is a function of formation pressure (Aoki and Shiba, 1973), the Al^{VI} vs. Al^{IV} diagram (Figure 8a) shows the pressure fields defined by Aoki and Kushiro (1968). Accordingly, the samples fall above the $\text{Al}^{\text{VI}}/\text{Al}^{\text{IV}} = 0.25$ line (Figure 8a). Since this line represents a pyroxene crystallization pressure of 5 kbar, the studied clinopyroxenes crystallized at some pressure above 5 kbar (Helz, 1973). Another method for calculating geothermobarometry of clinopyroxene was proposed by Soesoo (1997), based on crystallization pressure and temperature of the studied clinopyroxenes from 2 to 5 kbar (Figure 8b) and 1100 to 1150 °C (Figure 8c), respectively. Estimated values of clinopyroxene crystallization temperature and pressure using various methods are shown in Table 5. The clinopyroxene pressure (Putirka, 2008) vs. mg# diagram is graphically represented. According to

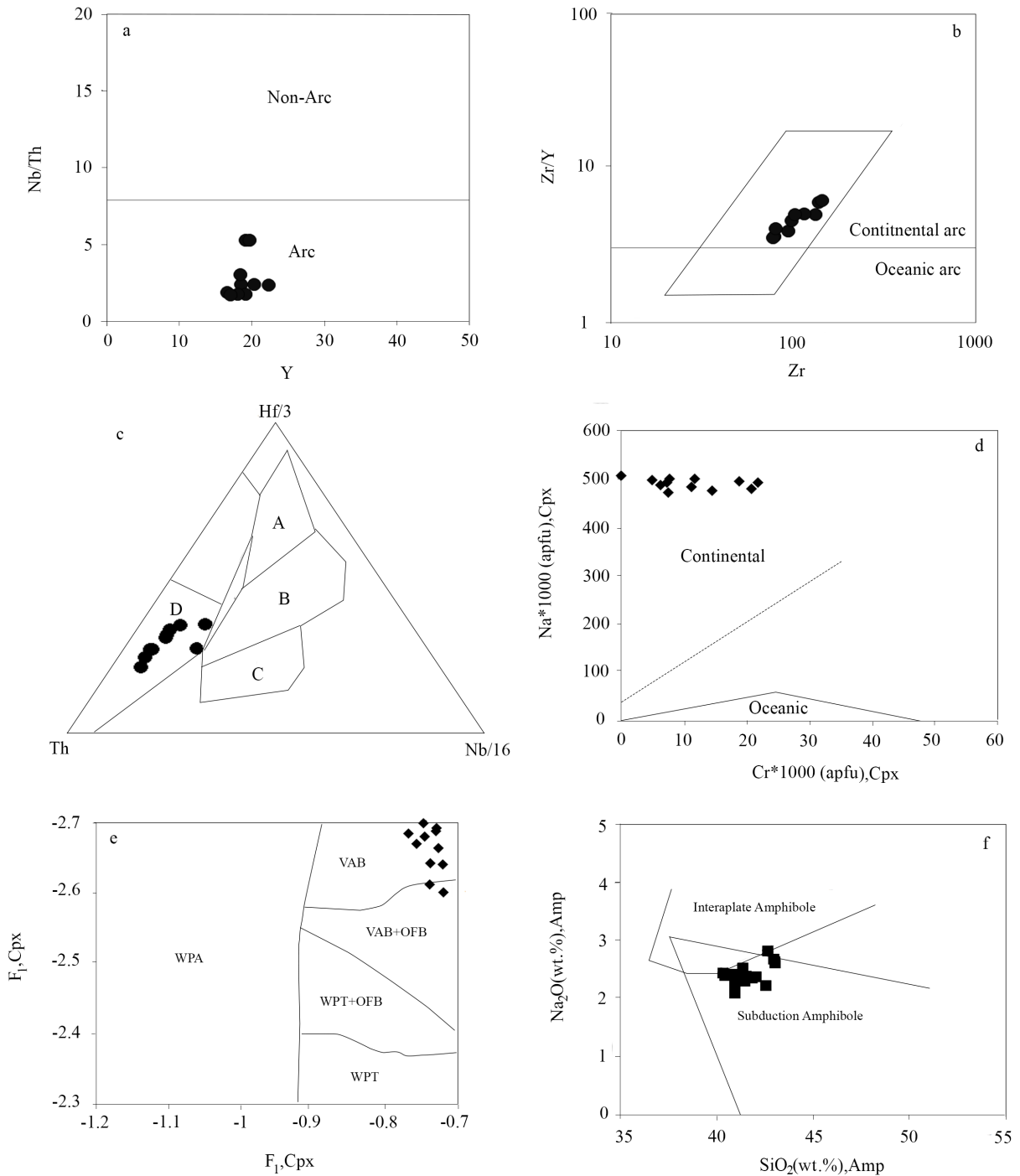


Figure 7. Tectonic setting discrimination diagrams based on whole-rock chemical compositions for the JBG mafic dikes: (a) Nb/Th–Y diagram (Swinden et al., 1989), (b) Zr– Zr/Y diagram (Pearce, 1982), (c) Th–Hf–Nb diagram (Wood, 1980). A - normal mid-ocean ridge basalts (N-MORB); B - enriched mid-ocean ridge basalts (E-MORB) and tholeiitic basalts within plate and differentiates; C - alkaline within plate basalts and within plate basalts and differentiates; D - destructive plate-margin basalts and differentiates (subduction zone). Tectonic setting determination diagrams based on mineral chemical data of the studied mafic dikes minerals: (d) Cr–Na diagram (Kornprobst et al., 1981), (e) $F_1 - F_2$ (Nisbet and Pearce, 1977) and, (f) $SiO_2 - Na_2O$ diagram (Coltorti et al., 2007) [$F_1 = -(0.012 * SiO_2) - (0.0807 * TiO_2) + (0.0026 * Al_2O_3) - (0.0012 * FeO^t) - (0.0026 * MnO) + (0.0087 * MgO) - (0.0128 * CaO) - (0.0419 * Na_2O)$, $F_2 = -(0.0469 * SiO_2) - (0.0818 * TiO_2) + (0.0212 * Al_2O_3) - (0.0041 * FeO^t) - (0.1435 * MnO) + (0.0029 * MgO) - (0.0085 * CaO) - (0.016 * Na_2O)$].

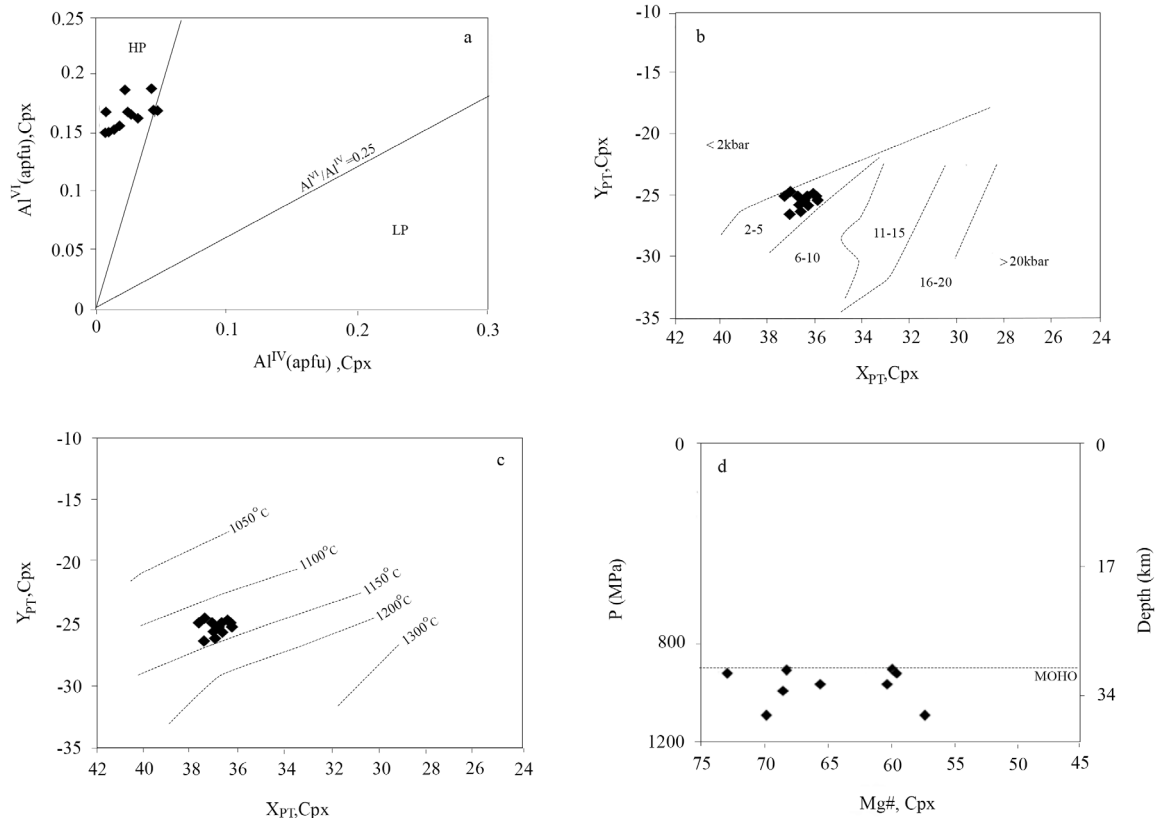


Figure 8. Clinopyroxene thermobarometry of the JBG mafic dikes: (a) Al^{VI} – Al^{IV} diagram (Aoki and Kushiro, 1968) [HP: High Pressure, LP= Low Pressure], (b and c) X_{PT} – Y_{PT} diagram (Soesoo, 1997) [$X_{PT} = (0.446 * SiO_2) + (0.187 * TiO_2) - (0.404 * Al_2O_3) + (0.346 * FeO) - (0.052 * MnO) + (0.309 * MgO) + (0.431 * CaO) - (0.446 * Na_2O)$], $Y_{PT} = (-0.369 * SiO_2) + (0.535 * TiO_2) - (0.317 * Al_2O_3) + (0.323 * FeO) + (0.235 * MnO) - (0.516 * MgO) - (0.167 * CaO) - (0.153 * Na_2O)$], (d) Pressure estimates according to equation 32b by Putirka (2008) vs. Mg# content, the vertical right axis shows the equivalence of pressure in depth.

continental crust density (3 g/cm), equivalence in depth of pressure values is shown in this diagram (Figure 8d).

This barometric data is useful for determining petrogenetic information about various stages of the magmatic history. The studied clinopyroxene composition is in agreement with a fractionation process in a magma chamber located at the mantle-crust transition or lower crust. The crystallization level at a depth of 30–40 km could indicate a probable magma chamber where these crystals had enough time to grow to large size (Figure 8d).

The Al^{VI} value of amphibole depends on temperature, which is affected in turn by water content of magma (Jiang et al., 1984). Using Al^{VI} and Al^I values of amphibole, the pressure is estimated based on amphibole composition (Blundy and Holland, 1990; Vynhal et al., 1991; Schmidt, 1992). According to Schmidt (1992), average amphibole crystallization pressure is 6.51 kbar. The amphibole crystallization temperature can be estimated using

amphibole-plagioclase couples. Considering Schmidt's (1992) average pressure and the latter method, average crystallization temperature of the studied amphibole is 1064°C. Table 6 shows estimated temperature and pressure of amphibole crystallization based on various methods.

5.4. Crustal contamination evidence

During dike emplacement into crustal levels, different degrees of crustal contamination occur in mantle-derived magmas (Mohr, 1987). The crustal contamination can be determined by checking the correlations between trace elements of the mantle and crust. For example, in continental crust, the values of Ce/Pb and Ba/Nb ratios are 3.9 and 57, respectively (Rudnick and Gao, 2003). Given that the Ba/Nb ratio of the studied samples are 57.57, the basaltic magma is found to be contaminated by crustal components. Furthermore, elevated abundances of Rb, U, and Pb and depletion of Nb, Ta, and Ti are indicative of crustal contamination of the samples (Figure 6c) (Rudnick

Table 5. Estimation of crystallization geothermobarometry of studied clinopyroxenes

Methods	P (kbar)		
	Min	Max	Average
	Putirka (1996)	6.9	10.5
Putirka et al. (2003)	7.1	10.8	9.72
Putirka, 2008 (equation 32b)	8.9	12.2	10.72
Nimis (1999)	5.3	9.2	7.33
	T (°C)		
Putirka (1996)	1242	1284	1271
Putirka et al. (2003)	1267	1312	1294
Putirka, 2008 (equation 32b)	1182	1210	1200
Nimis and Taylor (2000)	1197	1225	1208
Kretz (1994)	1205	1234	1218

Table 6. Estimation of crystallization geothermobarometry of studied amphiboles.

Methods	P (kbar)		
	Min	Max	Average
	By Al in amphibole		
Hammarstrom and Zen (1986)	6.32	8.28	7.07
Hollister (1987)	5.24	5.55	5.25
Johnson and Rutherford (1989)	5.84	6.70	6.23
Schmidt (1992)	6.72	8.53	6.51
Anderson and Smith (1995)	5.93	7.05	5.64
	T (°C)		
Otten (1984)	800	985	872
Colombi (1989)	797	993	867
By amphibole-plagioclase			
Holland and Blundy (1994)	984	1097	1064

and Gao, 2003). Moreover, the incompatible element ratios such as Nb/Th, Ce/Yb, and Zr/Y could be used to determine crustal contamination where Nb/Th and Ta/U ratios are known to capture the source characteristics and remain constant during basaltic magmatism (Pearce and Norry, 1979). In that respect, Nb/Th and Ta/U ratios decrease with increasing SiO₂ when mafic magmas are contaminated by the crust (Hu et al., 2016) (Figure 9a, 9b). Additionally, the Ce/Yb vs. Zr/Y diagram (Weaver and Tarney, 1984) shows the possibility of crustal contamination for the studied rocks (Figure 9c).

Fractional crystallization and crustal contamination play important roles in magma evolution and can change elemental compositions (DePaolo, 1981). Mafic dikes have relatively higher SiO₂ (46.8–50.2 wt.%) along with lower MgO (3.78–8.34 wt.%), Ni (5–45 ppm) and Mg# (0.27–0.44) compared to basaltic rocks, which is indicative of their highly evolved nature from mantle-derived parental melts through fractional crystallization of clinopyroxene and crustal contamination.

The Lu/Yb ratio of the mantle-derived magma is low (0.14–0.15) (Sun and McDonough, 1989), while continental crust exhibits relatively higher values of Lu/Yb ratio (0.16–0.18). The relatively high Lu/Yb ratios of the samples (0.14–0.18) are indicative of crustal contamination during magmatic evolution; in fact, the depletion in Nb and Ta confirms crustal contamination in magma (Figure 6c). Crustal components are rich in Zr, Hf, and Th and depleted in Nb and Ta (Rudnick and Gao, 2003), as the mantle-derived melt is depleted in Nb-Ta-Ti during subduction-zone magmatism and crustal contamination,

thus resulting in negative Nb-Ta and positive Zr-Hf anomalies in the spider diagram for mantle-derived melts (Figure 6c).

5.5. Origin of the studied dikes

Different mineralogical and geochemical characteristics along with related diagrams can be used to distinguish the nature of the source and genesis of the studied dikes. For example, as illustrated in Figure 10a (Jiang and An, 1984), amphiboles from the dikes fall in the mantle-derived amphibole field. Mafic magmas, which formed in the arc environment, are derived from mantle wedges and are later modified by slab-derived hydrous fluids (Kimura and Yoshida, 2006). LREE/HFSE and LILE/HFSE ratios of the arc magmas with slab melt-modified mantle origins are lower than those of magmas with fluid-modified mantle origins (Ayers, 1998; Class et al., 2000). The studied mafic dikes exhibit high Zr/Nb (17.41–25), low Th/U (2.25–4) and moderate Nb/Ta ratios (14.5–25), resembling the basalts/diabase from island arcs (Münker, 1998). The low La/Sm (2.24–3.1), Nb/Zr (0.04–0.09), and Th/Zr ratios (0.001–0.003) of the studied diabases reveal the mantle source was modified by slab fluids (Figure 10b).

The nature of the mantle source can be determined by the contents of HREEs in magma and garnet partition coefficient (Jenner et al., 1993). The high values of Sm/Yb (>5) and La/Yb ratios (>20) indicate that large quantities of HREEs remained in the garnet and amphibole of the source (Kay et al., 1994; Haschke and Gunther, 2003). According to Sm/Yb and La/Yb ratios (1.35–2.30, 3.23–6/17; respectively), the mantle source is garnet-free. The diagram of Sm/Yb vs. Ce/Sm (Çoban, 2007) shows partial

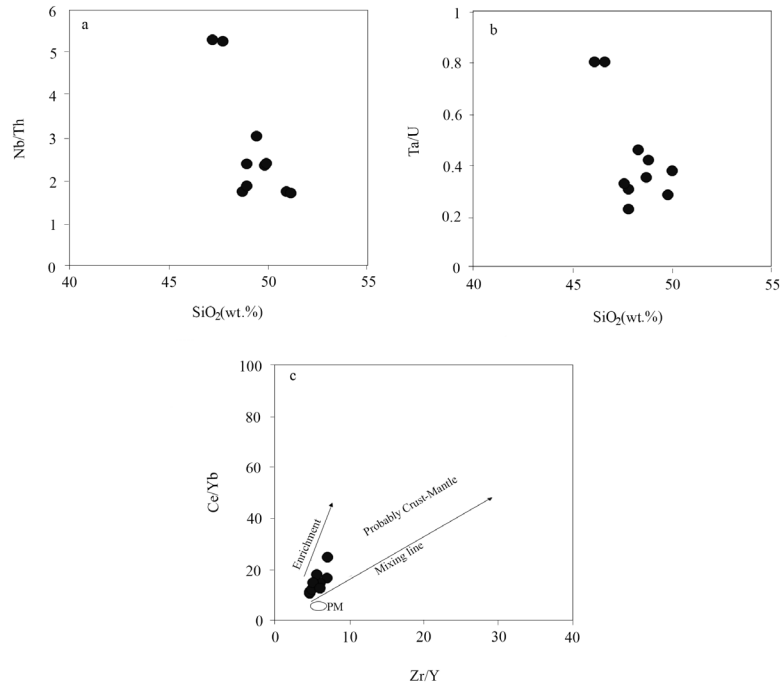


Figure 9. (a) Nb/Th–SiO₂ and (b) Ta/U–SiO₂ diagrams (Hu et al., 2016) of the JBG mafic dikes. The weak negative Nb/Th and Ta/U correlations with SiO₂ are consistent with crustal contamination during magma ascent. (c) Zr/Y–Ce/Yb diagram (Weaver and Tarney, 1984) for the studied mafic dikes.

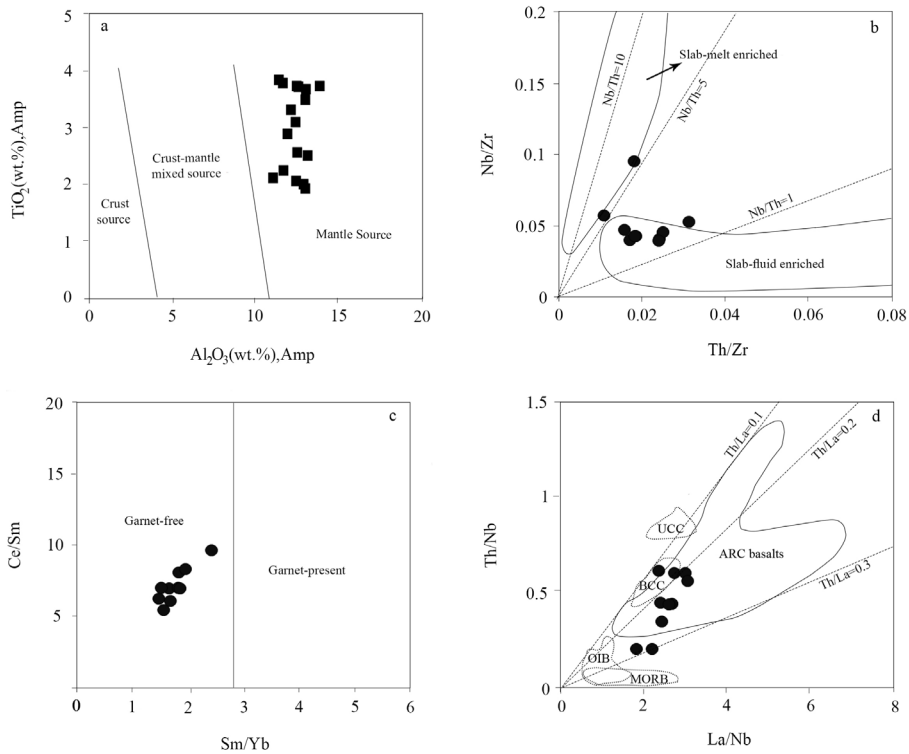


Figure 10. (a) Al₂O₃–TiO₂ diagram (Jiang and An, 1984), (b) Nb/Zr–Th/Zr diagram, (c) Ce/Sm–Sm/Yb diagram (Çoban, 2007), and (d) La/Nb–Th/Nb diagram (Plank, 2005) for the studied mafic dikes, UCC: average upper continental crust, BCC: average bulk continental crust.

melting process from a garnet-free mantle source (Figure 10c).

Given the presence of trace elements (Figure 6c), the primitive magmas from which these mafic dikes are sourced, were derived from mantle sources enriched in incompatible elements at higher rates than those of OIB sources (Figure 10d). Although melting of subcontinental lithospheric mantle metasomatized in a mantle wedge environment could produce primary and parental magma for the mafic dikes (Zhao et al., 2009). In Figure 9c, studied samples follow the enrichment trend, which is probably enriched by mantle metasomatism (Ahmad and Tarney, 1993). The Th/Nb-La/Nb ratios of all the studied samples are similar to those of arc basalt (Plank, 2005), indicating evolution by crustal components or melting of metasomatized mantle lithosphere with possible sediment input (Figure 10d).

6. Conclusions

1- The studied mafic dikes (diabases) trend N50–70W and N20–40W intruded into the Oligo-Miocene Jebal-E-Barez granitoids.

2- Small abundances of MgO, Ni and Mg# suggest that the parent magma underwent crystal fractionation of clinopyroxene and plagioclase.

3- The estimations of crystallization pressures and temperatures based on clinopyroxenes from the studied mafic dikes show ~8 kbar and ~1100 °C, respectively. According to this pressure, the crystallization levels were located at depths of 30–40 km and indicate it was probably a magma chamber placed at the mantle-crust transition or lower crust.

4- Mineral chemistry and whole-rock geochemistry data are consistent with generation of the primitive melts by partial melting of a garnet-free mantle source. The La_N/Sm_N and La_N/Lu_N ratios indicate the magma was generated from an enriched mantle source, but the parental magmas were further evolved by crustal components.

5- All evidence shows that the studied mafic dikes formed in an active continental arc environment.

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