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Research Article

Aqueous paramagnetic solutions for MRI phantoms at 3 T: A detailed study on relaxivities

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Abstract: Phantoms with known T_1 and T_2 values that are prepared using solutions of easily accessible paramagnetic agents are commonly used in MRI imaging centers, especially with the goal of validating the accuracy of quantitative imaging protocols. The relaxivity parameters of several agents were comprehensively examined at lower B_0 field strengths, but studies at 3 T remain limited. The main goal of this study is to measure r_1 and r_2 relaxivities of three common paramagnetic agents (CuSO₄, MnCl₂, and NiCl₂) at room temperature at 3 T. Separate phantoms were prepared at various concentrations of 0.05–0.5 mM for MnCl₂ and 1–6 mM for CuSO₄ and NiCl₂. For assessment of T_1 relaxation times, inversion recovery turbo spin echo images were acquired at 15 inversion times ranging between 24 and 2500 ms. For assessment of T_2 relaxation times, spin-echo images were acquired at 15 echo times ranging between 8.5 and 255 ms. Voxel-wise T_1 and T_2 relaxation times at each concentration were separately determined from the respective signal recovery curves (inversion recovery for T_1 and spin echo decay for T_2). Relaxivities r_1 and r_2 for all three agents that were derived from these relaxation time measurements are reported: $r_1 = 0.602 \text{ mM}^{-1} \text{ s}^{-1}$ and $r_2 = 0.730 \text{ mM}^{-1} \text{ s}^{-1}$ for CuSO₄, $r_1 = 6.397 \text{ mM}^{-1} \text{ s}^{-1}$ and $r_2 = 108.266 \text{ mM}^{-1} \text{ s}^{-1}$ for MnCl₂, $r_1 = 0.620 \text{ mM}^{-1} \text{ s}^{-1}$ and $r_2 = 0.848 \text{ mM}^{-1} \text{ s}^{-1}$ for NiCl₂. These results will serve as a practical reference to design phantoms of target T_1 and T_2 values at 3 T, in particular phantoms with relaxation times equivalent to specific human tissues.

Key words: T₁, T₂, relaxivity, 3 T MRI, paramagnetic phantoms, quantitative imaging

1. Introduction

Quantitative magnetic resonance relaxometry is a surging field of interest in MRI. By determining the relaxation time constants, one can generate quantitative tissue maps in vivo, to help distinguish healthy tissue from pathology [1]. Quantitative relaxometry also serves a critical role in characterization of MRI contrast agents that shorten the longitudinal relaxation time (T_1) and/or the transversal relaxation time (T_2) [2–5]. The efficacy of such contrast agents can be assessed via their relaxivities, i.e. the amount of shortening in T_1 or T_2 of nearby tissue per unit concentration of the agent. The reliability of these assessments depends on accurate quantification of relaxation time constants, typically necessitating prohibitively long scan times. Hence, improving the speed of T_1 and T_2 mapping sequences is an active area of research. With increased speed, however, one needs to ensure that the accuracy is not compromised. Therefore, it is desirable to validate the accuracy of the quantitative imaging protocols a priori on phantoms with known T_1 and T_2 values.

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Several paramagnetic contrast agents are frequently utilized by NMR/MRI researchers to prepare phantoms with desired T_1 and T_2 values, for example to mimic the T_1/T_2 of white matter tissue [6]. Among these agents are copper sulfate (CuSO₄), nickel chloride (NiCl₂), and manganese chloride (MnCl₂) [6–10], which are used due to their water-solubility, high stability, homogeneity, and the ease of preparing phantoms with relaxation times within the typical range for biological tissue [11,12]. The relaxivities of these agents were reported previously at low field strengths (0.5 T to 1.5 T), and these reports serve as a useful reference in preparation of phantoms with desired T_1/T_2 values. However, relaxivity parameters are known to show strong dependency on B₀ field strength [13,14]. With 3 T MRI scanners being extensively used in the clinic and in research settings [14–17], there is a pressing need for comprehensive relaxivity assessments at 3 T. A recent study looked at developing 3 T MRI phantoms that are similar to human tissues in terms of their relaxation times and conductivities [18]. The phantoms in that previous study were prepared using agarose, gadolinium chloride (GdCl₃), and sodium chloride (NaCl). Although Gd-based agents are gaining popularity [19–23], CuSO₄, MnCl₂, and NiCl₂ remain the most commonly used paramagnetic agents for making imaging phantoms [24–27]. However, except for a few studies on MnCl₂, the relaxivities for these agents have not yet been reported at 3 T.

Here, we measure and report the longitudinal (r_1) and transversal (r_2) relaxivities of three different paramagnetic solutions, CuSO₄, MnCl₂, and NiCl₂, at room temperature at 3 T. First, T₁- and T₂-weighted images are acquired for all three paramagnetic agents at various concentrations. The longitudinal (T_1) and transverse (T_2) relaxation times are obtained respectively from the exponential inversion-recovery and echodecay curves. T₁ fitting is performed using two different models: a conventional two-parameter model and a three-parameter model [28], which was recently shown to be more robust against B₁ inhomogeneities. The inverses of the relaxation times $(1/T_1 \text{ and } 1/T_2)$ of all three paramagnetic phantoms are found to be linear with concentration $(r^2 > 0.997)$. Furthermore, the three-parameter model significantly outperformed the twoparameter model based on a leave-five-out (L5O) cross-validation procedure (P < 0.001, paired Wilcoxon signed-rank test). Our results can serve as a practical reference for phantom design and for calibration of quantitative MRI imaging/analysis protocols at 3 T.

2. Theory

MRI image contrast can be greatly enhanced by exogenous contrast agents that significantly alter intrinsic T_1 and T_2 relaxation times of biological tissues. The relaxation times in the presence of such contrast agents can be approximated by:

$$\frac{1}{T_1} = \frac{1}{T_{1,dia}} + r_1 \cdot C \tag{1}$$

$$\frac{1}{T_2} = \frac{1}{T_{2,dia}} + r_2 \cdot C,$$
(2)

where the subscript 'dia' refers to diamagnetic host solution (water in our case), C [mM] is the concentration of the contrast agent, and $r_1 [mM^{-1} s^{-1}]$ and $r_2 [mM^{-1} s^{-1}]$ are the longitudinal and transverse relaxivities that reflect the efficiency of the agent. Here, the reciprocals of the relaxation times, $R_1 = 1/T_1 [s^{-1}]$ and $R_2 = 1/T_2 [s^{-1}]$, are called the concentration-dependent relaxation rates.

Various standard techniques are available for estimating relaxation time constants, such as inversion recovery (IR), look-locker (LL), saturation recovery (SR), or variable flip angle (VFA) method for T_1 mapping

[29], and Carr–Purcell–Meiboom–Gill (CPMG) multi-echo or single-echo spin echo (SE) sequences for T_2 mapping [30]. The optimal choice of relaxometry technique depends on the signal intensity, available scan time, and the required accuracy and precision for the estimation. Here we used the most common techniques for T_1 and T_2 mapping, IR and SE sequences, respectively.

In the IR sequence, the net magnetization is initially inverted by applying a 180° RF pulse. The magnetization is allowed to recover during a wait time called the inversion time (TI), which is followed by a 90° excitation RF pulse and data acquisition. Separate image acquisitions are performed at a range of distinct TI values. The conventional two-parameter signal model is then given by [25]

$$S = S_0 \left[1 - 2e^{-\frac{TI}{T_1}} + e^{-\frac{TR}{T_1}} \right],$$
(3)

where TR is the repetition time and S_0 is the signal amplitude after full magnetization recovery. Here the two parameters to be fitted to the acquired data are S_0 and T_1 . This idealized model assumes an exact 180° inversion pulse, which is rarely the case, as the effective flip angle depends on B_1 field uniformity, as well as T_1 and T_2 [31]. A recent study provided a more accurate model for the IR signal [32]:

$$S = S_1 + S_2 e^{-\frac{TT}{T_1}} \tag{4}$$

Here S_1 and S_2 can be seen as two separate components of the received signal: S_1 is the signal without any inversion pulses and S_2 stems from the inverted magnetization. Both S_1 and S_2 are complex valued, with S_2 having a 180° phase offset with respect to S_1 . When magnitude images (as opposed to complex MRI data) are utilized, S_1 and S_2 can be treated as real-valued parameters. In this case, S_1 will be positive valued and S_2 negative valued (see the Data analysis section for details on the extraction of signal positivity/negativity from magnitude images). Hence, the signal equation reduces to a three-parameter model, with the parameters S_1 , S_2 , and T_1 . Note that this model does not assume a perfect inversion pulse and it does not require TR >> T_1 . In this work, the T_1 mapping was performed via both the two- and the three-parameter models (Eqs. (3) and (4)), and the results were compared.

In the SE sequence that is used for T_2 mapping, a 90° excitation RF pulse flips the magnetization into the transverse plane. Any potential dephasing of the signal due to B_0 field inhomogeneity or chemical shift is corrected by applying a refocusing 180° RF pulse, followed by data acquisition at an echo time TE. Separate images are acquired at a range of distinct TE values. Assuming a monoexponential decay, the corresponding time constant T_2 is determined by

$$S = S_0 e^{-\frac{TE}{T_2}},$$
 (5)

where S_0 is signal amplitude without T_2 decay. Here the parameters to be fitted are S_0 and T_2 .

3. Materials and methods

3.1. Phantom preparation

Separate phantoms of manganese chloride (MnCl₂), copper sulfate (CuSO₄), and nickel chloride (NiCl₂) were prepared at 6 different concentrations, each with a total volume of 50 mL. MnCl₂ solutions varying between 0.05 and 0.5 mM concentration were prepared by dissolving anhydrous manganese chloride (99% purity, Sigma Aldrich) in double distilled water. Similarly, CuSO₄ and NiCl₂ solutions varying between 1 and 6 mM concentration were prepared by dissolving copper sulfate pentahydrate (CuSO₄.5H₂O, 98% purity,

Merck) and nickel chloride hexahydrate (NiCl₂.6H₂O, > 97% purity, Merck) in double distilled water. The concentrations were chosen to obtain similar ranges of T₁ values for all three paramagnetic agents (determined after preliminary MRI measurements, not shown) that were in the relevant range for biological tissue without [33] and with contrast injection [34] at 3 T. All solutions were prepared in sterilized polypropylene centrifuge tubes of 3-cm diameter and 12-cm length. Because T₁ and T₂ of pure water (approximately 5000 ms and 3200 ms, respectively [21]) is significantly higher than T₁/T₂ of these paramagnetic solutions, a pure water phantom was not included during MRI experiments (similar to previous studies such as [8,10,35–37]). Including pure water would require TR to be at least 3–4 times higher than the currently used value (section 3.2, MRI studies), which in turn would significantly prolong the imaging time.

3.2. MRI studies

Image acquisition was performed on a 3 T MRI scanner (Siemens Magnetom, maximum gradient strength of 45 mT/m and slew rate of 200 T/m/s) using a 32-channel receive-only head coil. For each paramagnetic agent, solutions prepared at 6 different concentrations were imaged concurrently (see Figures 1 and 2). The imaging parameters such as TR and the ranges of TI and TE were chosen to match the previous relaxivity study on MnCl₂ at 3 T [25], to enable a direct comparison of the results. The numbers of TIs and TEs were chosen based on preliminary experiments (results not shown) to give reliable r_1 and r_2 estimates. T_1 relaxation times were measured with an IR turbo spin echo sequence and TR = 3000 ms to allow for near-full recovery of magnetization. Images were acquired at 15 different TI = [24, 50, 100, 150, 200, 250, 500, 750, 1000, 1250, 1000, 1250, 1000, 1250, 1000, 11500, 1750, 2000, 2250, 2500 ms, with a total scan time of 2 min 41 s per image. A minimum TE = 12 ms and an acquisition matrix of 256×256 were prescribed. T₂ relaxation times were measured with a single-echo 135, 155, 175, 195, 215, 235, 255 ms, with a total scan time of 3 min 30 s per image. An acquisition matrix of 128×102 (i.e. 80% phase-FOV) was prescribed, and the final image was reconstructed with a matrix size of 256 \times 256. The remaining parameters were kept identical for both sequences: 4 mm slice thickness, 90 $^{\circ}$ flip angle, and 12 cm \times 12 cm field-of-view (FOV). Individual-coil images were sensitivity weighted and then linearly combined [38,39].

3.3. Data analysis

The MRI data were analyzed using an in-house script developed in MATLAB (MathWorks, Natick, MA, USA). For T_1 mapping, both two-parameter and three-parameter models were implemented. The reason for this choice was that, while it has been shown that the two-parameter model does not work well under B_1 field nonuniformity [28,32], it remains the most commonly used T_1 mapping method.

For each phantom, a circular region of interest (ROI) was chosen manually. Then pixel-wise T_1 values were determined in the selected ROIs (1214 \pm 124 pixels per ROI), first using the two-parameter model in Eq. (3). For magnitude MRI images, the sign of the signal S in Eq. (3) is not immediately available. To determine the sign, one first needs to determine the zero-crossing point of the inversion recovery curve (i.e. the TI for which the signal is zero). Hence, we first found the TI value, TI*, for which the voxel at hand had the minimum absolute signal level. Then, for all the TI values smaller than TI*, the sign of the signal was flipped. One cannot directly conclude whether the sign of the signal at TI* should also be flipped. Therefore, two different fittings were done: one where the signal at TI* remained positive valued and one where its sign was flipped.

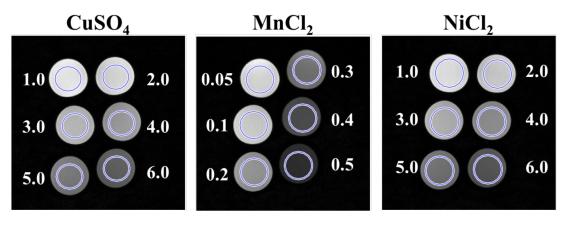


Figure 1. Example inversion recovery (IR) turbo spin echo images (acquired with TI = 100 ms) for all three paramagnetic solutions, showing the selected regions of interest (ROIs). The concentration for each phantom is denoted in units of mM. For each phantom, the T_1 values are determined in the selected circular ROI (1214 \pm 124 pixels per ROI). For this example image, TI = 100 ms corresponds to a time point before the zero crossing of the magnetization recovery curves for all samples. Hence, a lower signal level in the image denotes faster T_1 relaxation. Other imaging parameters were TR = 3000 ms, TE = 12 ms, 4 mm slice thickness, 90 ° flip angle, and 12 cm \times 12 cm field-of-view (FOV).

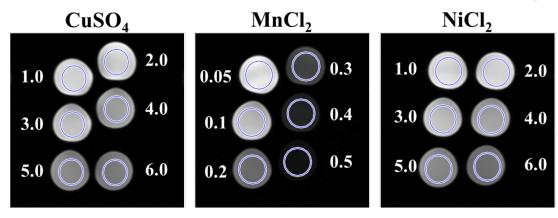


Figure 2. Example spin echo (SE) images (acquired with TE = 155 ms for CuSO₄ and NiCl₂, and with TE = 55 ms for MnCl₂) for all three paramagnetic solutions, showing the selected regions of interest (ROIs). The concentration for each phantom is denoted in units of mM. For each phantom, the T₂ values are determined in the selected circular ROI (1214 ± 124 pixels per ROI). In all three images, the signal level is lower for higher concentrations of paramagnetic ions, corresponding to faster T₂ relaxation. Other imaging parameters were TR = 2000 ms, 4 mm slice thickness, 90° flip angle, and 12 cm × 12 cm field-of-view (FOV).

Levenberg–Marquardt nonlinear least squares regression was performed on the resulting two data sets using Eq. (3). These two cases were then compared via the root mean square errors (RMSEs) of the fits and the one with the smaller RMSE was determined to be the correct case [32].

Next, the mean and standard deviation values for all pixels within the ROI were calculated. R_1 relaxation rate (i.e. $1/T_1$) was then plotted as a function of concentration of the paramagnetic phantoms (i.e. with 1214 \pm 124 points at each of six different concentration levels). Finally, the longitudinal relaxivity (r_1) was calculated via linear least squares regression on this plot. This entire procedure was repeated for the three-parameter model for T_1 mapping, using Eq. (4). The goodness-of-fit for the two-parameter vs. the three-parameter models was evaluated using the adjusted R^2 metric, and the results were compared via a paired Wilcoxon signed-rank test.

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One concern when using more parameters in a model is overfitting of the data [40]. To ensure that this was not the case with the three-parameter model, model performance was estimated via leave-five-out (L5O) cross-validation [41–43]. While the fact that the three-parameter model remains more robust under B₁ field inhomogeneities has been shown previously [28,32], a statistical confirmation that it does not overfit the data was not shown before. Accordingly, out of the 15 TI values, every third TI was removed from the data set (i.e. validation set had 5 TI values). The remaining 10 TI values acted as the training set for data fitting. The signal levels for the validation set were then estimated from the fitting results and compared with their actual values. This procedure was repeated three times by varying the validation set. The cross-validation results were evaluated using the adjusted \mathbb{R}^2 goodness-of-fit metric for both two-parameter and three-parameter models, and the results were compared via a paired Wilcoxon signed-rank test.

For T₂ measurements, a similar procedure was repeated using Eq. (5) (without the sign reversal step). The resulting mean and standard deviation values for R₂ relaxation rate (i.e. $1/T_2$) were plotted as a function of concentration and the transversal relaxivity (r₂) was calculated.

4. Results

In Figure 1, example inversion recovery images with TI = 100 ms are shown for all three samples. The particular TI value shown in Figure 1 corresponds to a point before the zero crossing of the magnetization recovery curves for all samples, so that a lower signal level in the image corresponds to a faster T_1 relaxation, which in turn corresponds to higher concentrations of the paramagnetic phantoms. Similarly, example spin-echo images for all three samples are given in Figure 2. As expected, the signal level is lower for higher concentrations of the paramagnetic phantoms, corresponding to faster T_2 relaxation.

The pixel-wise T_1 values were determined in the selected ROIs, and the measured signal intensities were fitted using Eqs. (3) and (4). The adjusted R^2 goodness-of-fit metric for the three-parameter model was found to be significantly higher (P < 0.001, paired Wilcoxon signed-rank test) than that of the two-parameter model. To visually show the difference between the two models, T_1 color map and adjusted R^2 color map for both models were computed for the MnCl₂ phantoms, as shown in Figure 3. As seen in this figure, the three-parameter model displays uniformly higher levels of adjusted R^2 , all very close to the ideal value of one. In fact, the adjusted R^2 values for the three-parameter model were higher for all pixels in the ROIs (1214 ± 124 pixels per phantom, and a total of 18 phantoms for all three paramagnetic solutions), indicating a better fit to the data points.

Next, we selected the pixel where the difference between the adjusted \mathbb{R}^2 values between the two models was the maximum. Figures 4A and 4B show the measured signal intensities as a function of TI for that pixel (in 0.5 mM MnCl₂ phantom), with fitted T₁ magnetization recovery curves overlaid. The two-parameter fit gave T₁ = 248.5 ms with adjusted $\mathbb{R}^2 = 0.9922$, whereas the three-parameter fit gave T₁ = 298.7 ms with adjusted $\mathbb{R}^2 = 0.9996$. Upon closer inspection of the fitted curves, one can see that the two-parameter fit deviates from the data points at low and high TI values. The three-parameter fit, on the other hand, provides a much better agreement with the data points. Similar behavior was observed at other concentrations of MnCl₂ (not shown). To overrule the possibility that the three-parameter model overfits the data points, L5O cross-validation was performed, where the three-parameter model outperformed the two-parameter model (P < 0.001, paired Wilcoxon signed-rank test). Hence, we conclude that the three-parameter model provides a more accurate representation of the inversion recovery curve, which could stem from its robustness against nonideal inversion RF pulse resulting from B₁ inhomogeneity.

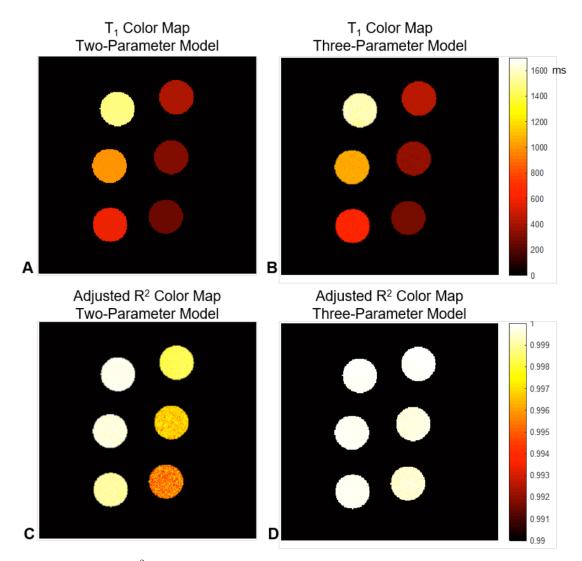


Figure 3. T_1 and adjusted R^2 color map comparisons for two- and three-parameter models for MnCl₂ phantom. For MnCl₂, (A) the two-parameter model slightly underestimates the T_1 values when compared to (B) the three-parameter model. The adjusted R^2 performance of (C) the two-parameter model gets worse for lower T_1 values. (D) The three-parameter model, on the other hand, has uniformly higher levels of adjusted R^2 values, all very close to the ideal value of one. Note that in this figure, T_1 and adjusted R^2 values are computed on a pixel-by-pixel basis, only for the pixels in the selected circular ROIs (see Figure 1). Also note that (A-B) share the same color scale and (C-D) share the same color scale to enable a direct visual comparison.

Figure 4C shows an example of the fitting for the T_2 signal decay curve. The measured signal intensities as a function of TE are plotted for a single pixel of the 0.05 mM MnCl₂ phantom. The resulting T_2 was 158.8 ms with adjusted $R^2 = 0.9999$, and the fitted curve agreed well with the measurements. The obtained mean values of T_1 and T_2 for all three paramagnetic ions for varying concentrations, along with their standard deviations, are tabulated in Table 1.

The relaxation rates $R_1 = 1/T_1$ and $R_2 = 1/T_2$ are plotted as a function of concentration for all three paramagnetic solutions in Figures 5 and 6, respectively. The error bars denote the mean and standard

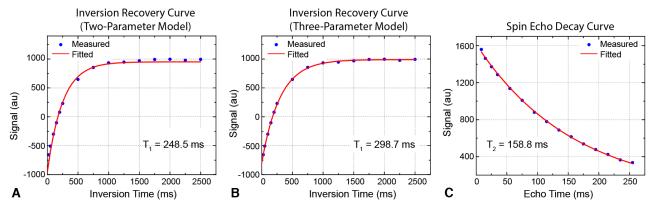


Figure 4. Inversion recovery curve for a single pixel of the 0.5 mM MnCl₂ phantom fitted using (A) the two-parameter model (Eq. (3)) and (B) the three-parameter model (Eq. (4)). The two-parameter model gives $T_1 = 248.5$ ms with adjusted $R^2 = 0.9922$, while the three-parameter model gives $T_1 = 298.7$ ms with adjusted $R^2 = 0.9996$. The two-parameter fit deviates from the data points at low and high TI values, while the three-parameter fit shows a much better agreement. (C) Spin echo signal decay curve for a single pixel of the 0.05 mM MnCl₂ phantom. The fitted curve has $T_2 = 158.8$ ms with adjusted $R^2 = 0.9999$.

| $CuSO_4$ | Concentration (mM) | | | | | |
|---|--------------------|--|------------------|------------------|------------------|------------------|
| | 1.0 | 2.0 | 3.0 | 4.0 | 5.0 | 6.0 |
| $T_1 (ms)$ | 1115 0 1 0 0 | <u> </u> | 4055 4 9 99 | | 000 0 1 1 55 | 0010 1 1 00 |
| 2-Parameter Model | 1115.7 ± 7.2 | 679.9 ± 3.3 | 485.5 ± 2.22 | 376.7 ± 1.66 | 308.3 ± 1.55 | 261.0 ± 1.22 |
| $T_1 (ms)$ | 11951 1 150 | color + colo | 494 5 1 4 9 | 979 5 1 0 7 | 205 2 1 2 0 | 0571 01 |
| 3-Parameter Model | 1135.1 ± 15.0 | 681.8 ± 6.2 | 484.5 ± 4.2 | 373.5 ± 2.7 | 305.3 ± 2.9 | 257.1 ± 2.1 |
| $T_2 (ms)$ | 894.9 ± 32.5 | 549.4 ± 11.6 | 399.2 ± 7.1 | 304.2 ± 3.4 | 246.7 ± 6.4 | 211.4 ± 1.8 |
| MnCl ₂ | Concentration (mM) | | | | | |
| | 0.05 | 0.1 | 0.2 | 0.3 | 0.4 | 0.5 |
| $T_1 (ms)$ | 1481.8 ± 12.9 | 990.2 ± 6.8 | 571.5 ± 3.5 | 407.3 ± 2.1 | 309.0 ± 2.6 | 040.0 + 0.4 |
| 2-Parameter Model | | | | | | 248.2 ± 2.4 |
| $T_1 (ms)$ | 1501 2 1 20 0 | 1046 5 1 16 4 | $c_{1c} 0 + 7c$ | 4407 1 2 0 | 9401 1 1 1 | 999.0 ± 4.9 |
| 3-Parameter Model | 1501.3 ± 28.0 | 1046.5 ± 16.4 | 010.9 ± 1.0 | 448.7 ± 3.8 | 348.1 ± 5.1 | 283.6 ± 4.8 |
| $T_2 (ms)$ | 163.5 ± 2.6 | 88.4 ± 0.3 | 44.5 ± 0.2 | 30.5 ± 0.1 | 22.8 ± 0.1 | 18.2 ± 0.1 |
| NiCl ₂ | Concentration (mM) | | | | | |
| | 1.0 | 2.0 | 3.0 | 4.0 | 5.0 | 6.0 |
| $T_1 (ms)$ | 10075 - 50 | C 4 9 C + 9 C | 469 7 1 1 0 | 2002 ± 0.0 | 0.070 + 1.0 | 050 7 1 1 0 |
| 2-Parameter Model $\left 1067.5 \pm 5 \right $ | | 643.6 ± 3.6 | 463.7 ± 1.8 | 360.3 ± 2.8 | 297.2 ± 1.3 | 250.7 ± 1.2 |
| $T_1 (ms)$ | 1000 0 1 12 0 | CAF 7 + 7 9 | 462 2 1 2 4 | 257.4 ± 4.9 | 0077 ± 0.0 | 0.47.7 + 1.0 |
| 3-Parameter Model | 1082.0 ± 13.2 | 040.7 ± 7.3 | 463.3 ± 3.4 | 357.4 ± 4.2 | 295.7 ± 2.2 | 247.7 ± 1.9 |
| $T_2 (ms)$ | 743.6 ± 17.6 | 474.1 ± 6.8 | 336.9 ± 6.5 | 262.0 ± 3.2 | 214.1 ± 2.8 | 179.3 ± 1.7 |

Table 1. The T_1 and T_2 relaxation times of CuSO₄, MnCl₂, and NiCl₂ measured at 3 T for various concentrations. The mean values and standard deviations are given for each selected ROI (1214 ± 124 pixels per ROI). The T_1 values are reported for both the two-parameter and the three-parameter model as shown in Eqs. (3) and (4), respectively.

deviation among all pixels in the ROI for a given concentration of a sample. The results of the linear least square regressions are shown with red solid lines in Figures 5 and 6, where the slopes correspond to r_1 and r_2

relaxivities ($r^2 > 0.997$ for all fitted lines). These relaxivity values are listed in Table 2 for all three paramagnetic ions, along with their 95% confidence intervals.

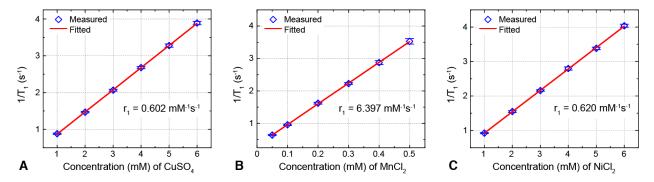


Figure 5. Longitudinal relaxation rates $(R_1 = 1/T_1)$ as a function of concentration for all three paramagnetic solutions, fitted using the three-parameter model. The slopes of the fitted lines correspond to the r_1 relaxivities: (A) $r_1 = 0.602$ mM⁻¹ s⁻¹ for CuSO₄, (B) $r_1 = 6.397$ mM⁻¹ s⁻¹ for MnCl₂, and (C) $r_1 = 0.620$ mM⁻¹ s⁻¹ for NiCl₂. The error bars show the mean and standard deviations of the relaxation rates for each selected ROI at the given concentration (1214 ± 124 pixels per ROI), and the solid red lines denote the linear least squares regressions with $r^2 > 0.999$.

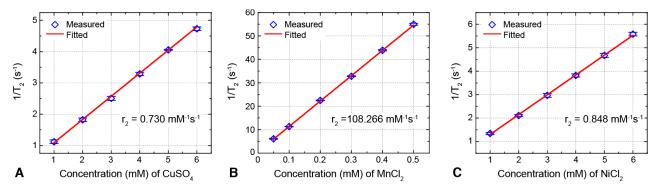


Figure 6. Transversal relaxation rates ($R_2 = 1/T_2$) as a function of concentration. The slopes of the fitted lines correspond to the r_2 relaxivities: (A) $r_2 = 0.730 \text{ mM}^{-1} \text{ s}^{-1}$ for CuSO₄, (B) $r_2 = 108.266 \text{ mM}^{-1} \text{ s}^{-1}$ for MnCl₂, and (C) $r_2 = 0.848 \text{ mM}^{-1} \text{ s}^{-1}$ for NiCl₂. The error bars show the mean and standard deviations of the relaxation rates for each selected ROI at the given concentration (1214 ± 124 pixels per ROI), and the solid red lines denote the linear least squares regressions with $r^2 > 0.997$.

Table 2. The longitudinal relaxivity (r_1) and transversal relaxivity (r_2) for CuSO₄, MnCl₂, and NiCl₂ measured at 3 T, along with their 95% confidence intervals. While we provide r_1 from both the two-parameter and the three-parameter models for the sake of completeness, the three-parameter model is more accurate. Hence, the last two columns are highlighted as the accurate r_1 and r_2 values for these parameters.

| | $\begin{array}{ c c c } r_1 \ (mM^{-1} \ s^{-1}) \\ \hline 2 - Parameter \ Model \end{array}$ | ${f r_1 \ (mM^{-1} \ s^{-1})} \ {f 3}	ext{-Parameter Model}$ | ${f r}_2 \ ({f m}{f M}^{-1} \ {f s}^{-1})$ |
|-------------------|---|--|--|
| $CuSO_4$ | $0.588 \ (0.5881 - 0.5885)$ | $0.602\ (0.6019 - 0.6028)$ | $0.730 (0.7285 {-} 0.7310)$ |
| MnCl ₂ | 7.444 (7.4375–7.4506) | $6.397 \ (6.3903 - 6.4031)$ | $108.266\ (108.2193{-}108.3126)$ |
| NiCl ₂ | 0.609 (0.6091 - 0.6096) | $0.620\ (0.6197 – 0.6206)$ | $0.848 (0.8465 {-} 0.8488)$ |

Finally, the concentrations of $CuSO_4$, $MnCl_2$, and $NiCl_2$ required to mimic the T_1/T_2 relaxation times of the basic tissue types such as gray matter, white matter, skeletal muscle, and blood at 3 T [33] have been determined and are shown in Table 3. Accordingly, $CuSO_4$ and $NiCl_2$ require significantly different concentrations for mimicking the T_1 vs. the T_2 of a given tissue. Hence, one can prepare either T_1 -mimicking phantoms or T_2 -mimicking phantoms with these agents, but not both. $MnCl_2$, on the other hand, can closely match both the T_1 and T_2 of the listed tissues at approximately the same concentrations, and hence is a better choice for tissue mimicking phantoms.

| Table 3. The concentrations of $CuSO_4$, $MnCl_2$, and $NiCl_2$ required for achieving T_1/T_2 of basic tissue types such |
|---|
| as gray matter, white matter, skeletal muscle, and blood along with the relaxation times of these tissues at 3 T [33] are |
| shown. For example, approximately 0.03 mM solution of $MnCl_2$ closely mimics both the T_1 and T_2 relaxation times |
| of blood. |

| | | $CuSO_4 (mM)$ | $MnCl_2 (mM)$ | NiCl ₂ (mM) |
|-----------------|-------------------------|---------------|---------------|------------------------|
| Gray matter | $T_1 = 1820 \text{ ms}$ | 0.468 | 0.036 | 0.394 |
| | $T_2 = 99 ms$ | 13.341 | 0.088 | 11.387 |
| White matter | $T_1 = 1084 \text{ ms}$ | 1.087 | 0.094 | 0.995 |
| | $T_2 = 69 ms$ | 19.359 | 0.128 | 16.568 |
| Skeletal muscle | $T_1 = 1412 \text{ ms}$ | 0.732 | 0.060 | 0.650 |
| Skeletal muscle | $T_2 = 50 \text{ ms}$ | 26.905 | 0.179 | 23.065 |
| Blood | $T_1 = 1932 \text{ ms}$ | 0.415 | 0.031 | 0.343 |
| Diood | $T_2 = 275 ms$ | 4.482 | 0.028 | 3.761 |

5. Discussion

As seen in Table 2, the r_1 relaxivities calculated using the two-parameter and the three-parameter models are similar, but do not match exactly. Specifically, the r_1 values calculated using the two-parameter model are 1%– 2% lower than those from the three-parameter model for CuSO₄ and NiCl₂. For MnCl₂, on the other hand, the two-parameter model gave a 16% higher r_1 value than the three-parameter model. We would like to note that the r_1 relaxivity of MnCl₂ at 3 T using the two-parameter model was previously reported as 7.4 mM⁻¹ s⁻¹ by Nofiele and Cheng [25]. Our experiments were conducted in the same range of MnCl₂ concentrations (up to 0.5 mM) as in Nofiele and Cheng's study, and our results agree perfectly when we also use the two-parameter model. Both the previous study and this work show that the R_1 relaxation rate at 0.5 mM displays increased standard deviation values or does not agree well with the fitted regression line. For the three-parameter model in Figure 5B, on the other hand, the r_1 relaxivity provides a much better fit to the $1/T_1$ vs. concentration data points. As explained by Nofiele and Cheng, this difference could stem from the fact that the three-parameter model does not assume a perfect 180° inversion pulse. Hence, in theory, whenever there is B₁ inhomogeneity or any variation in the flip angle, the three-parameter model will provide more accurate results. Therefore, we highlight the results of the three-parameter model in Table 2.

The previously reported $r_2 = 117 \text{ mM}^{-1} \text{ s}^{-1}$ for MnCl₂ [25] compares well with our result of $r_2 = 108 \text{ mM}^{-1} \text{ s}^{-1}$ (approximately 8% difference). This relatively small difference may be due to differences in signal-to-noise ratios in the MRI images between the two studies. In addition, although the ranges of MnCl₂ concentrations used in the two experiments match, the experiment in Nofiele and Cheng's study did not have any data points between 0.2 mM and 0.5 mM. In such a case, small errors in measurement and/or fitting at 0.5 mM may cause deviations in the fitted slope, potentially leading to the difference observed here.

The relaxivities of the paramagnetic agents used in this work were previously reported at 1.5 T. Accordingly, the relaxivity values (r_1, r_2) in units of mM⁻¹ s⁻¹ were given as follows: for Cu²⁺ (0.69 ± 0.04, 0.77 ± 0.04), Mn²⁺ (7.0 ± 0.4, 70 ± 4.0), and Ni²⁺ (0.7 ± 0.06, 0.7 ± 0.06) [44–45]. Comparing these values with the relaxivities at 3 T reported in this work, r_1 values are smaller and r_2 values are either comparable or larger at 3 T than at 1.5 T. This trend is consistent with previous works that list relaxivities at various field strengths for MnCl₂ [46], gadolinium [22], and iron oxides [47]. It should be noted that the actual trend of relaxivity vs. field strength is not necessarily monotonous if one looks at a wider range of field strengths [2–5,48]. Hence, these results cannot be generalized.

The relaxivities (r_1, r_2) in units of mM⁻¹ s⁻¹ for some of the clinically used gadolinium-based contrast agents were previously reported at 3 T: for Gadovist $(3.2 \pm 0.18, 3.9 \pm 0.16)$, Omniscan $(3.2 \pm 0.18, 3.3 \pm 0.16)$, and Gadomer $(13.0 \pm 0.1, 23.0 \pm 0.04)$ [21]. Comparing these values with the ones listed in Table 2, gadolinium-based agents have significantly higher relaxivites than CuSO₄ and NiCl₂. On the other hand, relaxivities of MnCl₂ are comparable or higher than these clinical contrast agents, which is one of the reasons for the popularity of manganese-based contrast agents in preclinical research at 3 T and at higher field strengths [23,26,27]. It should be emphasized that the dosage of the manganese utilized in preclinical/clinical settings should be carefully adjusted to minimize the toxic side effects [49]. Accordingly, increasing the biocompatibility and relaxivity of manganese-based agents with different chelates/ligands is an active area of research [23,50].

6. Conclusion

We report the longitudinal (r_1) and transversal (r_2) relaxivities of MnCl₂, CuSO₄, and NiCl₂ paramagnetic solutions at 3 T. The relaxivities of these agents were previously reported at lower B₀ field strengths, but a detailed study at 3 T was not available. These paramagnetic solutions are chemically and thermally stable, and their relaxation times are within the biological range. Hence, these paramagnetic agents are of practical importance when preparing MRI phantoms with desired T₁ and T₂ values for testing and/or calibrating various MRI sequences, especially for quantitative imaging methods.

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